Virtual Lab: Types of Chemical Reactions

Find the Link on Moodle for "Virtual Lab: Types of Chemical Reactions", or go to youtube and type in "5 Types of Chemical Reactions Lab" and click on the first link, or type in the following website: https://www.youtube.com/watch?v=nsEkKliOz7Q

Watch the video and follow along on your worksheet as you go.

- Observations before Reaction: What do you think is going to happen?
- Observations after reaction: What did you see happen?
- Word equation: write the reaction out in words
- Skeleton equation: write the reaction out using chemical formulas (not balance)
- Balanced equation: write the complete balanced equation with chemical formulas

1. Type of Reaction: ______

| Observations Before Reaction | Observations After Reaction | | |
|------------------------------|-----------------------------|--|--|
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| | | | |
| | | | |
| | | | |

Word Equation:

Skeleton Equation:

Balanced Equation:

2. Type of Reaction: ______

| Observations Before Reaction | Observations After Reaction | | |
|------------------------------|-----------------------------|--|--|
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| | | | |
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Word Equation:

Skeleton Equation:

Balanced Equation:

3. Type of Reaction: ______

| Observations Before Reaction | Observations After Reaction | | |
|------------------------------|-----------------------------|--|--|
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Word Equation:

Skeleton Equation:

Balanced Equation:

4. Type of Reaction: ______

| Observations Before Reaction | Observations After Reaction | Observations After Reaction | | |
|------------------------------|-----------------------------|-----------------------------|--|--|
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Word Equation:

Skeleton Equation:

Balanced Equation:

5. Type of Reaction: ______

| Observations Before Reaction | Observations After Reaction | | |
|------------------------------|-----------------------------|--|--|
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| | | | |
| | | | |

Word Equation:

Skeleton Equation:

Balanced Equation:

Questions: Balance the equation and state the type of reaction

| 1 | Na + | $\HOH \rightarrow \$ | NaOH + | _H₂ | Type of Reaction |
|---|------------------------------------|------------------------------------|-----------------------------------|------------------|---|
| 2 | Fe + | _O ₂ →Fe ₂ C | 93 | | Type of Reaction |
| 3 | Zn + | _CuSO₄→ | _Cu +ZnSO | 4 | Type of Reaction |
| 4 | Zn(C₂H₃C | 9 ₂)₂ +Na₃PC | $0_4 \rightarrow _\NaC_2H_3C_2$ | D ₂ + | Zn ₃ (PO ₄) ₂ Type of Reaction |
| 5 | $_H_2O$ → | H2 + | _O ₂ | | Type of Reaction |
| 6 | Al ₂ (SO ₄) | 3 +CaCl2 | \rightarrow AlCl ₃ + | CaSC | \mathcal{D}_4 |

Type of Reaction_____