

The Periodic Table



Most of the elements in the periodic table are

- A. metals
- B. metalloids
- C. gases
- D. nonmetals



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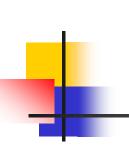




- A. hydrogen
- B. oxygen
- C. helium
- D. carbon

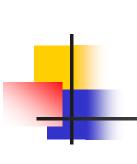
What is necessary for substances to burn?

- A. hydrogen
- B. oxygen
- C. helium
- D. carbon



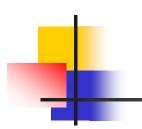
Which one of the following tells the physical state of an element at room temperature?

- A. The atomic number
- B. The color of the chemical symbol
- C. The atomic mass
- D. The element name



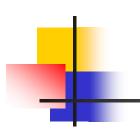
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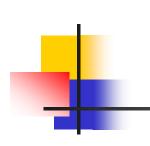
Mendeleev arranged the elements by

- A. density
- B. melting point
- C. appearance
- D. increasing atomic mass



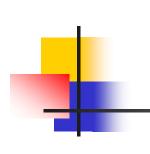
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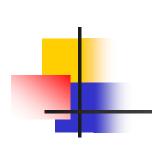
How do the physical and chemical properties of the elements change?

- A. Within a group
- B. Across each period
- C. Within a family
- D. Across each group



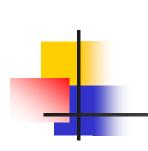
How do the physical and chemical properties of the elements change?

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- C. Within a family
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The horizontal row on the periodic table is called a(n)

- A. group
- B. family
- C. period
- D. atomic number



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- A. group
- B. family
- C. period
- D. atomic number



Transition metals are

- A. Good conductors of thermal energy
- B. More reactive than alkali metals
- C. Not good conductors of electrical current
- D. Used to make aluminum



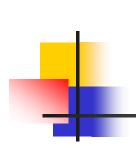
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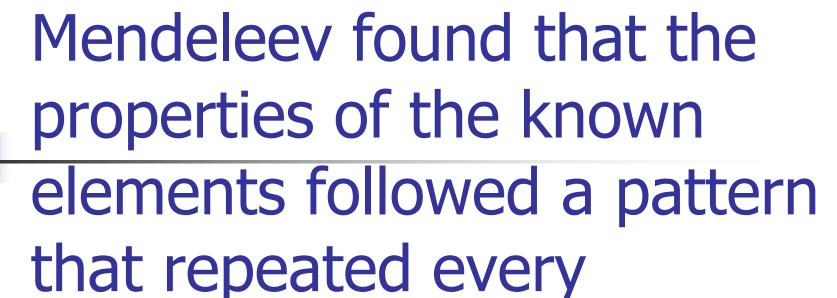
The vertical column of elements on the periodic table is called a(n)

- A. period
- B. semiconductor
- C. atomic mass
- D. group



The vertical column of elements on the periodic table is called a(n)

- A. period
- B. semiconductor
- C. atomic mass
- D. group



- A. 7 elements
- B. 5 elements
- C. 14 elements
- D. 10 elements

Mendeleev found that the properties of the known elements followed a pattern that repeated every

- A. 7 elements
- B. 5 elements
- C. 14 elements
- D. 10 elements



These metals react with water to form hydrogen

- A. alkali metals
- B. aluminum
- C. halogens
- D. argon



These metals react with water to form hydrogen

- A. alkali metals
- B. aluminum
- C. halogens
- D. argon



Light bulbs last longer when they are filled with

this gas

- A. carbon
- B. hydrogen
- C. oxygen
- D. argon



Light bulbs last longer when they are filled with this gas

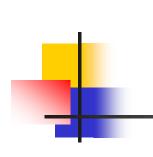
- A. carbon
- B. hydrogen
- C. oxygen
- D. argon



This metal, part of the Boron Group, is used for

aircraft parts.

- A. carbon
- B. lanthanides
- C. aluminum
- D. alkali metals



This metal, part of the Boron Group, is used for aircraft parts.

- A. carbon
- B. lanthanides
- C. aluminum
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This makes up about 80% of the air we breathe.

- A. carbon
- B. halogens
- C. nitrogen
- D. oxygen



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This is important to most living things

- A. calcium
- B. oxygen
- C. hydrogen
- D. carbon



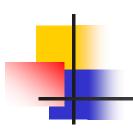
This is important to most living things

- A. calcium
- B. oxygen
- C. hydrogen
- D. carbon



Diamond and soot are forms of this

- A. lanthanides
- B. nitrogen
- C. aluminum
- D. carbon



Diamond and soot are forms of this

- A. lanthanides
- B. nitrogen
- C. aluminum
- D. carbon



Cement and chalk are compounds of this.

- A. calcium
- B. hydrogen
- C. argon
- D. carbon



Cement and chalk are compounds of this.

- A. calcium
- B. hydrogen
- C. argon
- D. carbon



This is a colorless, odorless gas

- A. nitrogen
- B. halogen
- C. hydrogen
- D. oxygen



This is a colorless, odorless gas

- A. nitrogen
- B. halogen
- C. hydrogen
- D. oxygen



Some of these reactive metals are used to make steel.

- A. aluminum
- B. lanthanides
- C. argon
- D. alkali metals



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- C. argon
- D. alkali metals

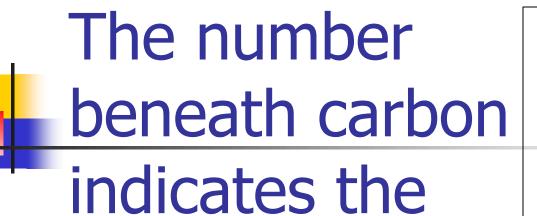
Chlorine and iodine are these.

- A. hydrogens
- B. lanthanides
- C. halogens
- D. alkali metals

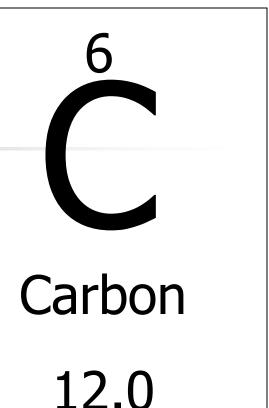
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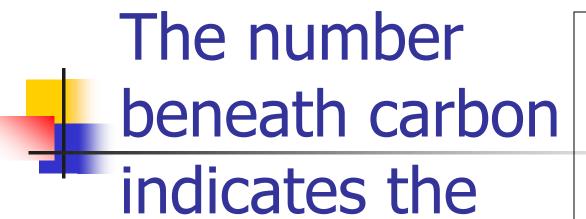


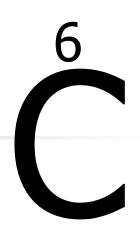
- A. hydrogens
- B. lanthanides
- C. halogens
- D. alkali metals



- A. Atomic number
- B. Atomic mass
- C. Chemical symbol
- D. Element name





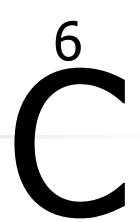


Carbon

12.0

- A. Atomic number
- B. Atomic mass
- C. Chemical symbol
- D. Element name



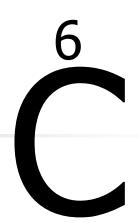


Carbon

12.0

- A. Atomic number
- B. Element name
- C. Atomic mass
- D. Chemical symbol



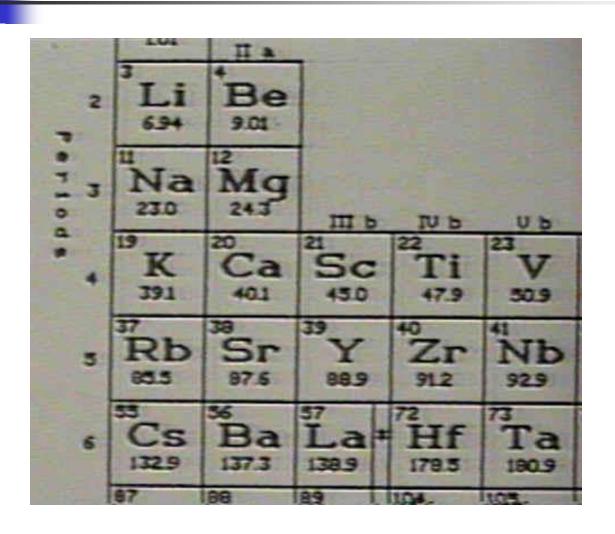


Carbon

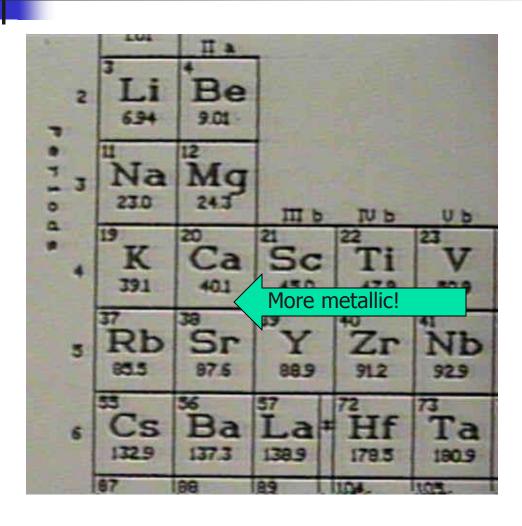
12.0

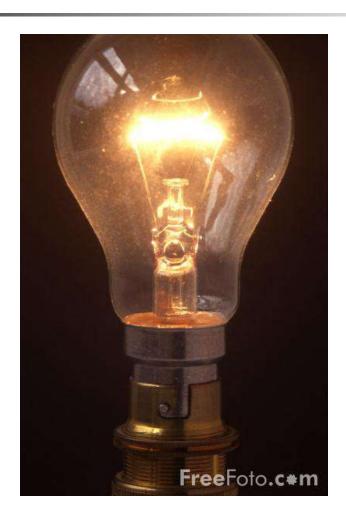
- A. Atomic number
- B. Element name
- C. Atomic mass
- D. Chemical symbol

Which element is more metallic, calcium (Ca) or vanadium (V)?

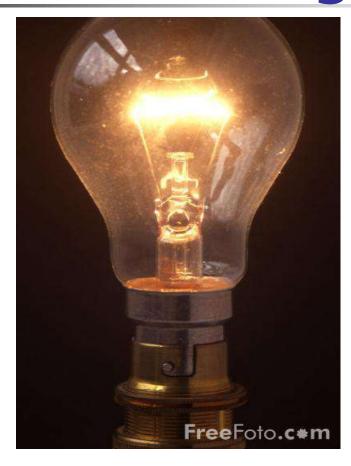


Calcium (Ca) is more metallic than vanadium (V) because it is farther to the left of the table.





Argon makes light bulbs last longer.





Boron group, used for aircraft parts = aluminum



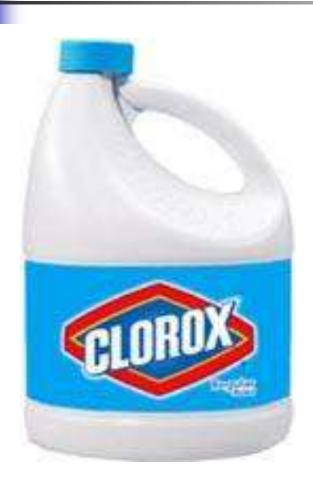




Cement and chalk are compounds of calcium.

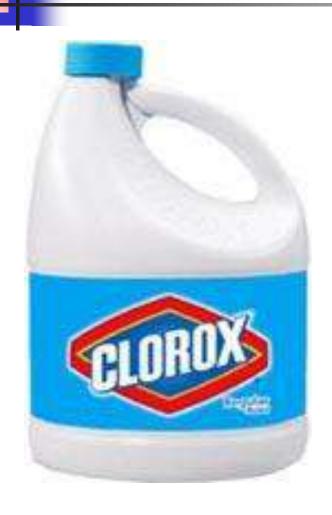








Chlorine and iodine are halogens.











Diamonds and soot are forms of carbon.











Oxygen is important to most living things.









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