

# **Physics 1 Honors: S.I. Units and Scientific Notation**

# Why use the SI System?

**In the U.S. we use the English or Standard System, most of the rest of the world uses the Metric or SI System.**

The logo for the International System of Units (SI) is displayed within a light gray rectangular border. It features the text "International System of Units" at the top in a black serif font. In the center, the letters "SI" are written in a large, bold, black serif font. At the bottom, the French text "le Système International d'unités" is written in a smaller black serif font, with the words "Système" and "International" highlighted in red.

**The SI (International System of Units) system is the form of measurement typically used by scientists.**

# Why use the SI System?

**Scientists use the SI System worldwide because:**

- **Measurements are easily understood by all scientists**
- **Measurements are easier to convert than the English system**

# MEASUREMENT SYSTEM COMPARISONS

MEASUREMENT	ENGLISH	SI SYSTEM
LENGTH	Yard / Inch	Meter / Centimeter
MASS	Ounce / Pound	Gram / Kilogram
VOLUME	Quart	Liter
TEMPERATURE	Fahrenheit	Celsius / Kelvin
TIME	Second	Second

**All Measurement systems have standards. Standards are exact quantities that everyone agrees to use as a basis of comparison.**

# BASIC TYPES OF MEASUREMENT

**In SI the basic units  
are:**

- ✓ Length is the meter (m)
- ✓ Mass is the gram (g)
- ✓ Volume is the liter (L)
- ✓ Temperature is Celsius ( C )

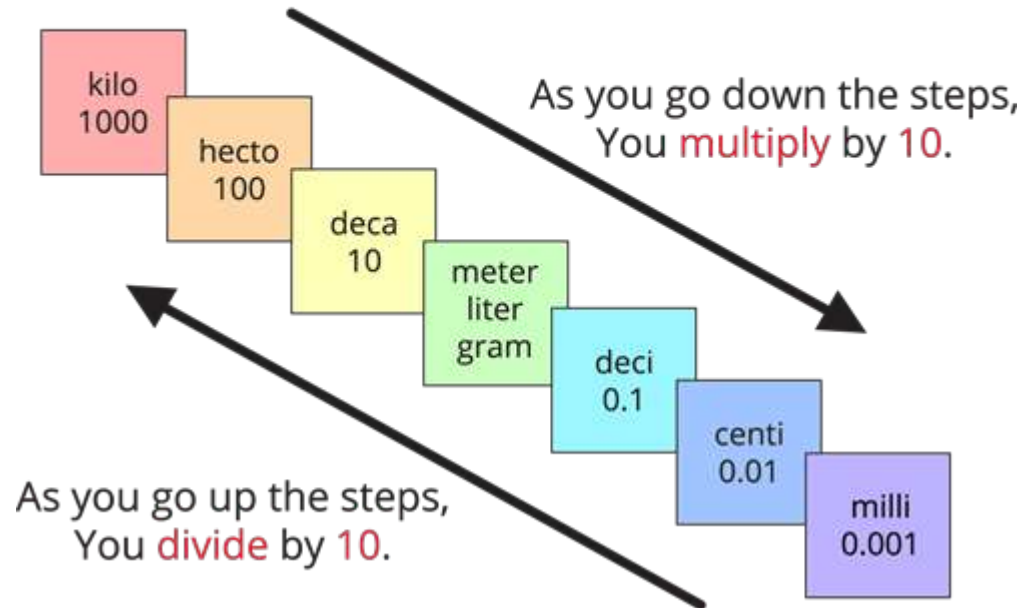
<b>Kilo</b>	<b>1000</b>
<b>Hecto</b>	<b>100</b>
<b>Deca</b>	<b>10</b>
<b>UNIT</b>	<b>1</b>
<b>Deci</b>	<b>1/10</b>
<b>Centi</b>	<b>1/100</b>
<b>Milli</b>	<b>1/1000</b>

**This system works with any SI measurement.**

**The UNIT becomes whichever type of measurement you are making. (mass, volume, or length)**

**It is the same system regardless if you are measuring length, mass, or volume.**

# Metric System Conversion Stairs



# More Conversions . . .

2,321.0 millimeters to meters =

521.0 grams to hectograms =

8.5 kiloliters to centiliters =

**NOTE:**      *The digits aren't changing, the position of the decimal is. In the English system the whole number changes.*

Kilo   Hecto   Deca   UNIT   Deci   Centi   Milli



# Dimensional Analysis

**When you are doing mathematical operations with values that have units, those mathematical operations also apply to units. Using the units all the way through the course of a problem is called “Dimensional Analysis”**

**For example:**

**10 seconds + 15 seconds = \_\_\_\_\_**

**2 meters \* 10 seconds = \_\_\_\_\_**

Practice on your own...

1.  $15 \text{ m} / 5 \text{ s} = ?$

1.  $2\text{m} * 2\text{m} = ?$

1.  $(1\text{m} * 10 \text{ s}) / 5 \text{ s} = ?$

# Scientific Notation Practice

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In physics we sometimes have to work with large numbers (the distance from the Earth to the Sun is about 149,000,000,000 meters) and very small numbers (the mass of an electron is 0.0000000000000000000016 kg). To properly work with these we use scientific notation.

Physics doesn't usually focus on significant figures as much as you did in chemistry.

# Scientific Notation Practice

149,000,000,000 meters from the Sun. Convert this into scientific notation.

1) The mass of an electron is 0.0000000000000000000000016 kg. Convert this into scientific notation.

1) The mass of the Earth is  $5.98 \times 10^{24}$  kg. Write this as a normal number.

## Multiplication in Scientific Notation

Add the exponents

$$(a \times 10^m)(b \times 10^n)$$

Multiply the coefficients

*Example:*

$$\begin{aligned}(3.1 \times 10^4)(2 \times 10^3) \\&= (3.1 \times 2) \times 10^{4+3} \\&= 6.2 \times 10^7\end{aligned}$$

## Division in Scientific Notation

Divide the  
coefficients

$$\frac{a \times 10^m}{b \times 10^n}$$

Subtract the  
exponents

*Example:*

$$\begin{aligned}\frac{9.06 \times 10^8}{3 \times 10^5} &= \frac{9.06}{3} \times 10^{8-5} \\&= 3.02 \times 10^3\end{aligned}$$