

Name \_\_\_\_\_

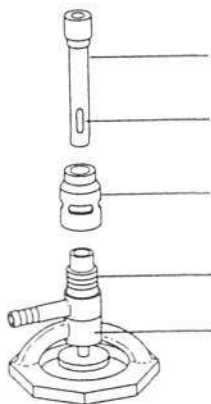
Section \_\_\_\_\_ Date \_\_\_\_\_

## Report for Experiment 1

# Chemistry of Fire

### Prelaboratory Questions

1. Label the Bunsen burner parts shown below.



2. What is the proper color for a burner flame?

### Data/Observations

#### Part 1 Use of the Burner

1. Draw a copper wire within the flame. Indicate the color of the copper wire at several places within the flame:



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## Part 2

### Efficiency of the Bunsen Burner

Record the information from each of the five trials in the table below:

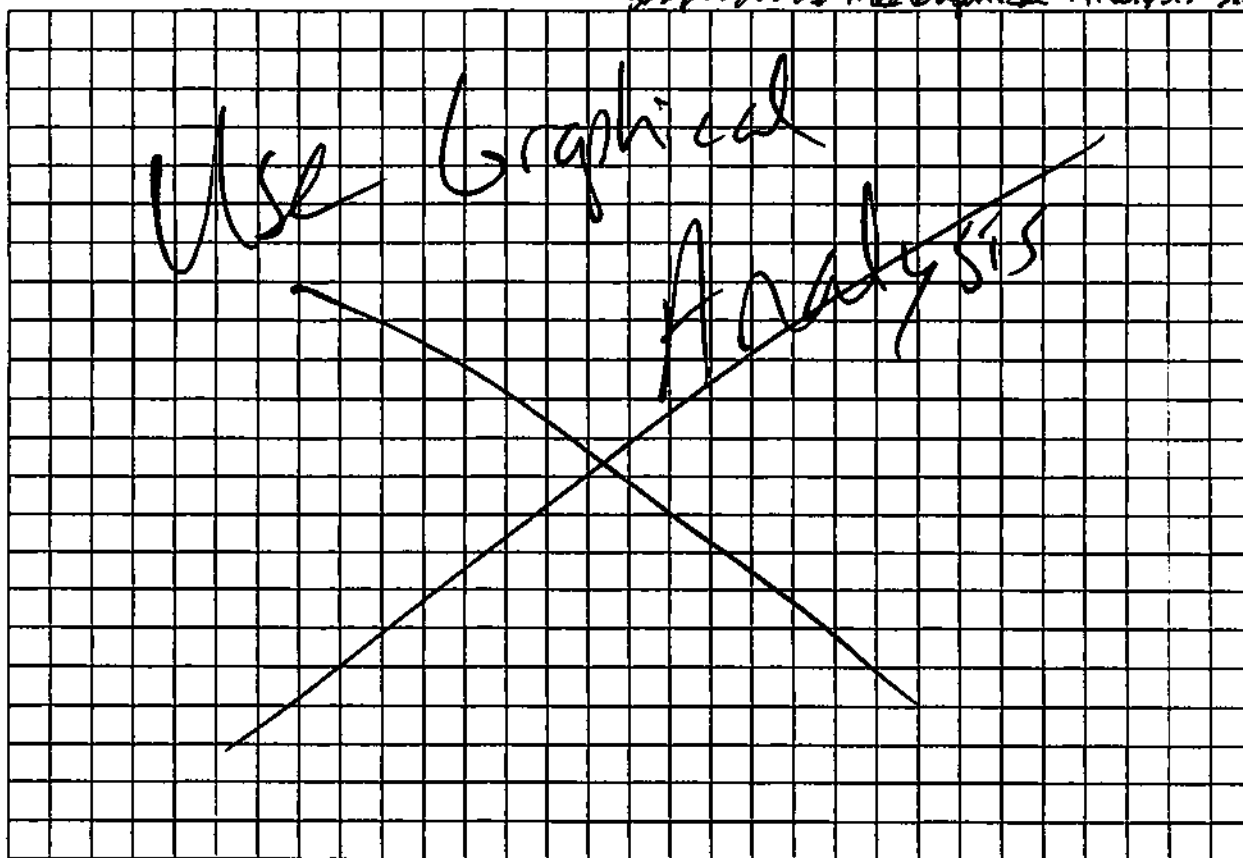
Trial	Volume of water	Height of beaker over burner	Temp Reading 1	Temp Reading 2	Temp Reading 3	Temp Reading 4	Temp Reading 5
1							
2							
3							
4							
5							

## Analysis and Conclusions

*Use Graphical Analysis for graph.*

1. Make a graph of temperature versus time. Plot time on the x-axis (horizontal) and temperature on the y-axis (vertical). Place the plots for all of the trials on the same graph. Label each line with the height of the beaker above the burner.

*2. Empty Extra Credit: Make the same graph using the Graphical Analysis Software.*



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2. Which line shows the most efficient heating of the water? Explain why you chose this line.

3. What are the advantages of using a blue flame instead of a yellow one for heating objects in the laboratory?

4. Where is the hottest part of the blue flame?

5. When heating a substance over a Bunsen burner where should the object be placed for most efficient heating? Why?

6. How does graphing the data help to determine the most efficient height for heating a liquid in a beaker?