



Edmore Public School  
706 Main St, Edmore, ND 58330

**Physical Science Lesson Plans for  
November 28 – December 2, 2022  
1<sup>st</sup> Hour, 8:40 – 9:32 AM**

	Monday (Nov 28)	Tuesday (Nov 29)	Wednesday (Nov 30)	Thursday (Dec 1)	Friday (Dec 2)
<b>Performance Standards</b>	<b>HS-PS1-7</b> Use mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction.	<b>HS-PS1-7</b> Use mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction.	<b>HS-PS1-7</b> Use mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction.	<b>HS-PS1-7</b> Use mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction.	<b>HS-PS1-7</b> Use mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction.
<b>Topic</b>	Rate of Chemical Reaction	Mixtures	Dissolving	Solubility and Concentration	Molarity
<b>Objectives</b>	<ul style="list-style-type: none"> <li>• Discuss how to speed up the rate of reaction</li> <li>• Describe what catalyst do</li> </ul>	<ul style="list-style-type: none"> <li>• differentiate heterogeneous and homogeneous mixture</li> </ul>	<ul style="list-style-type: none"> <li>• explain why water is called the universal solvent</li> </ul>	<ul style="list-style-type: none"> <li>• define solubility</li> <li>• distinguish saturated, unsaturated and supersaturated solutions</li> </ul>	<ul style="list-style-type: none"> <li>• Compute for the molarity of the solutions</li> </ul>
<b>Bellringer</b>	(3 min) emulsion	(3 min) alloy	(3 min) miscible	(3 min) solubility	(3 min) vocab quiz
<b>Procedure/ Instructional Delivery</b>	<ul style="list-style-type: none"> <li>○ Review activity: law of conservation of mass</li> <li>○ Simulation lab: balancing chemical equation</li> </ul>	<ul style="list-style-type: none"> <li>○ Review of heterogeneous and homogeneous mixture</li> <li>○ Direct instructions on Types of Solutions and Mixtures</li> <li>○ Why it matters: gasoline production</li> <li>○ Close: section review</li> </ul>	<ul style="list-style-type: none"> <li>○ Lab on dissolving substances</li> </ul>	<ul style="list-style-type: none"> <li>○ Lesson introduction: Solubility</li> <li>○ Student activity: solubility and concentration</li> <li>○ Demonstration: saturated, unsaturated, and supersaturated solution</li> </ul>	<ul style="list-style-type: none"> <li>○ Direct instruction on molarity</li> <li>○ Independent practice: solving molecular weight and molarity of substances</li> </ul>
<b>Assessment</b>	worksheet	Section review	Lab paper	Exit ticket	Worksheet
Remarks					

Prepared by:

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