NamePeriod	
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Essential Question:

How do you find the percentage composition of sugar in bubble gum and water in popcorn?

Percentage Composition

In this experiment, you will determine the % of sugar in bubble gum and the percentage of water in popcorn.

BUBBLE GUM

- 1. You will need one person in your group to volunteer to chew gum (must chew for 15 minutes).
- 2. Measure the mass of the unchewed gum still in its wrapper (note flavor).
- 3. Unwrap gum (save wrapper) and chew gum for 5 minutes.
- 4. Place gum back on wrapper and reweigh.
- 5. Repeat for two more 5 minute chew cycles, reweighing after each cycle.
- 6. Calculate the % of mass lost by the gum. This is the % of sugar that was in the gum.

% sugar = (total mass lost ÷ original mass of gum) x 100

7. Draw a graph of mass of gum vs. time, and explain what the graph shows



- 8. Compare results involving various flavors of gum.
- 9. Complete a lab report. Write a conclusion paragraph.

POPCORN (Use Aluminum foil to make bowls similar to jiffy pop)

- Mass 3 different size samples of unpopped and popped corn. Your smallest sample should be 25 kernels.
- 2. For each example, calculate % of mass lost by popped corn. This represents the % of water that was in the corn.

% water = (difference in mass between unpopped and popped Corn ÷ mass of unpopped corn) x 100

- 3. Calculate the average % of water in your 3 samples.
- 4. Complete a lab report. Write a conclusion paragraph (include an explanation of how popcorn pops and how age may affect the popping process).

LAB REPORT FORMAT:

Title, Purpose, Data Table, Calculations, Graph (bubble gum only), conclusion

Percentage Composition

Part I: Bubble Gum					
Mass (unchewed gum in wrapper)					
Mass (after 5 minutes)					
Mass (after 10 minutes)					
Mass (after 15 minutes)					
% Composition of sugar: total mass lost x 100 = original mass of gum					
Draw a graph of mass of gum vs time and explain what the graph shows.					
Part II:Popcorn					
Sample #	Mass (unpopped)		Mass (popped)		
% Composition of water: difference in mass between popped and unpopped					
Sample 1: Sample 3:		Sample 2:			
Average % of water in your					