#### OUTLINING

#### Why and How to Create a Useful Outline

#### What is an outline?

- A graphic organizer for information
  - Shows the <u>order</u> of various pieces of information
  - Shows the relative importance of each
  - Shows the <u>relationship</u> between the various parts

#### Why create an outline?

- To make writing or studying easier
- To organize large amounts of information
- To study and learn information from a textbook

- One of the most important skills you could ever learn is outlining.
- In science you will outline chapter sections in the Physical Science text to build background knowledge.
- How will your outline be assessed?
  - Outlines will be collected and graded, or..
  - You will self-assess using a rubric, or...
  - You will be quizzed on the information and allowed to use your outline
  - Your outline must be your own work: no partners.

#### How do I create an outline?

- Analyze the information to determine what is important
  - Do not simply copy the information
  - Not all information is important: You be the judge!
  - Ask yourself: "Is this piece of information important for me to remember?"

# How do I create and outline? (cont'd)

- **Organize** the information
  - Determine the order of importance of the information (usually the way it is presented in the textbook)
  - Use phrases (use the least number of words, but keep the meaning)
  - Avoid using complete sentences

# How do I create and outline? (cont'd)

- Preferred format
  - Roman Numerals: I, II, III, IV, (most general)
  - Upper Case Letters: A, B, C,
  - Arabic Numerals: 1, 2, 3,
  - Lowercase Letters: a, b, c,
  - Lowercase Roman Numerals: i, ii, iii, iv, v (most specific)
- Go from most general to most specific

#### Format Example

I.First Main Topic – (Section Headings in Text)

(skip a line here)

- A. Use capital A, B, C for sub topics (Blue Headings in Text)
  - 1. Important info on this part of the topic
  - 2. More info
    - a. more detail
    - b. more detail
    - i. Example
    - ii. Example
    - iii. Example

(skip a line here)

II. Second Main Topic

#### Here's an example of an outline

- I. Montgomery Upper Middle School Students and Staff
  - A. 7th Grade
    - 1. Science
    - 2. Social Studies
    - 3. Math
    - 4. LA
    - 5. Cycle
    - 6. Language
  - B. 8th Grade
    - 1. Science
      - a. Blakemore
      - b. Chesbro
      - c. Kleinfield
      - d. Sowa
        - i. YOU 🕲
    - 2. Social Studies
    - 3. Math
    - 4. LA
    - 5. Cycle
    - 6. Language

### Outline Requirements (copy these onto right side of notebook now)

- <u>Chapter # and Title</u> Write and underline
- Section 1 Title Roman numeral I
- Blue Headings Upper Case letters
- **Details** under Blue Headings (including green titled sections) **Numbers** (1,2,3)
- Vocabulary terms Underline
- Examples use "Ex."
  - Ex. Rubber band
  - Ex. Balloon

#### Let's look at Ch. 1.3

Chapter 1: The World of Physical Science

#### **III. Using Models in Physical Science**

- A. What is a Model?
- 1. <u>Model</u> = a representaion of an object or system
- 2. Used to explain things in science that we can't see or touch
- 3. Never exactly like the real thing
- 4. Ex: model rocket, atom model, building model, cell model
- B. Models Help You Visualize Information
- 1. Objects as Models
- a. Must have similar characteristic to the real thing
- b. Ex. Slinky=sound wave
- 2. Ideas as Models
- a. Used when you have an idea but don't have an object

for a model

- b. Ex. Imagine a drop of ice tea with sugar dissolved into it.
- 3. Models in Weather Forecasting
- a. Satellite pictures
- b. Color-coded maps
- c. Radar images

### Now, you try.

- Read the next part of the section.
- Work with a partner to complete the next part of the outline.
- What will you write for "C"?
- What details will you include under C?
- How will they be labeled in your outline?

- C. Models Are Just the Right Size
- 1. Moon model used because moon is too big to work with
- 2. Model of salt can be used because salt is too small to work with
- D. Models Build Scientific Knowledge
- 1. Models can be tools for investigations
- 2. Testing Hypotheses with a model
- a. Use model to collect data
- 3. Illustrating Theories with a model
- a. Ex. Atomic theory to show atoms in a metal
- E. Models Can Save Time and Money
- 1. Ex. Car engineers use computer-simulated crashes to determine safety features without damaging a single car

#### Your Turn.....

- Read Chapter 1 Section 4
- Determine what information is important
- Follow format rules for outlining
- Do it!

Is it similar to the outline shown on the next slide?

#### Chapter 1: The World of Physical Science

- IV. Measurement and Safety in Physical Science
- A. The International System of Units (SI)
- 1. global measurement system developed in France in the late 1700s
- 2. used by most scientists in almost all countries
- 3. helps scientist share info
- 4. units based on "10" for easy conversion
- 5. Length
- a. Meters (m) = basic SI unit of length
- b. 1 km = 1000 m (used for large lengths)
- c. 1000 mm = 1 m (used for small lengths)

- 6. <u>Volume</u> = the amount of space that something occupies
- a. Volume is expressed in liters (L), which are based on the meter.
- b. 1 cubic meter = 1000 liters,  $1 \text{ cm}^3 = 1 \text{ mL}$
- c. Liquid volume is measured using a graduated cylinder
- d. Regular solid volume is calculated by l x w x h and is expressed in cm<sup>3</sup>
- e. Volume of an irregular shaped object is measured using water displacement, where  $1 \text{ mL} = 1 \text{ cm}^3$

- 7. Mass = the amount of matter that something is made of
- a. SI Unit for mass is the kilogram (kg)
- b. For small objects use the gram (g)
- c. 1000 g = 1 kg
- 8. <u>Temperature</u> = a measure of how hot or cold something is
- a. Scientists use degrees Celsius (°C)
- b. SI unit is the kelvin (K)
- c. Ex. Water boils at 100 °C, normal body temp is 37 °C, water freezes at 0 °C
- B. Derived Quantities = combinations of other measurements
- 1. <u>Area</u> = how much surface an object has
- a. Area = length x width
- b. expressed in square units, such as  $m^2$  or  $cm^2$
- 2. <u>Density</u> = mass per unit volume, the amount of matter in a given space
- a. D = m/V
- b. Density units are g/cm<sup>3</sup>