ATMOSPHERE

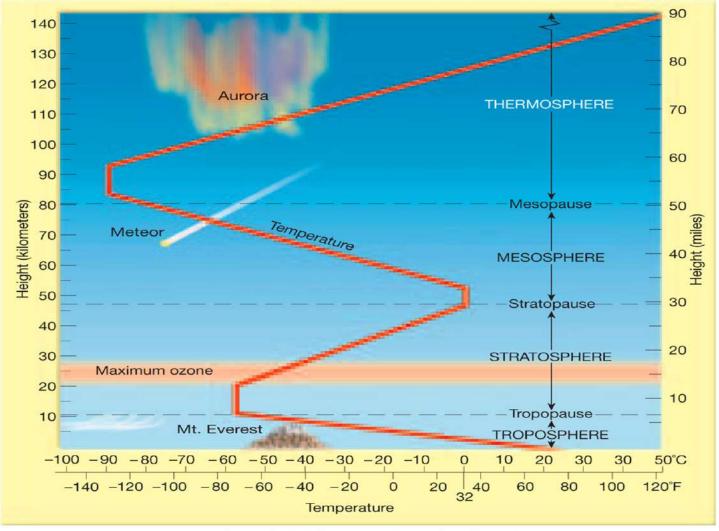


ESSENTIAL QUESTION:

• How are we tied to our atmosphere?

- How has the formation of an atmosphere affected life?
- How has life affected the atmosphere?
- How has the atmosphere changed?
- How has internal Earth processes affected our atmosphere, and life?
- How can small changes cause large feedback loops?
- In what ways are we altering our atmosphere?

LAYERS



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LAYERS OF ATMOSPHERE

Layers identified by changes in *temperature*

Troposphere-80% of total atmosphere

- ■~10 mi
- Weather occurs here
- Temp decreases3.5°F per 1000 feet (average)
- Boundary: <u>tropopause</u>

Stratosphere

- About 10 30 miles
- Temperature increases at top
- This is where **<u>ozone</u>** is found
- Outer boundary is named the stratopause

LAYERS, CONTINUED

Mesosphere

30-50 mi

Temp decreases (little ozone)

Mesopause is boundary

Thermosphere

50-90+ mi

Temp increases due to intense solar radiation

Ionosphere

90-500 mi

Air is ionized

ATMOSPHERIC GASES

Greenhouse gases

- CO2-traps heat and stores it near Earth's surface, preventing escape into space
- Increasing CO2 makes the Earth warmer!
- Without ANY greenhouse effect-Earth would be <u>FREEZING</u>!
- Ozone-O3
- Blocks UV radiation

| Principal gases in dry air | percentage |
|----------------------------|------------|
| Nitrogen | 78% |
| Oxygen | 21% |
| Argon | .9% |
| Carbon Dioxide | .036% |

ATMOSPHERIC GASES

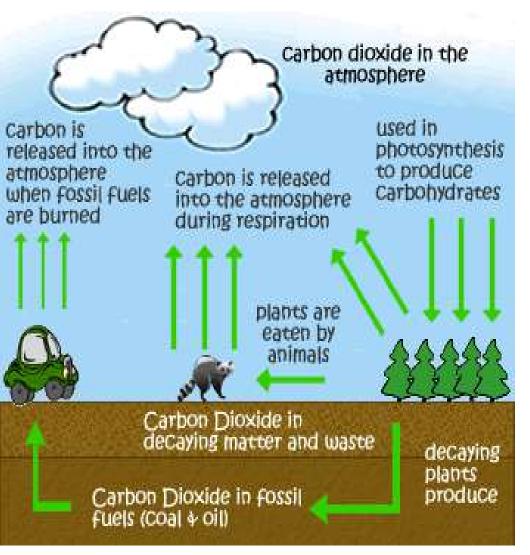
Gases remain constant due to recycling processes within Earth's "systems"

Balance is maintained due to equal parts leaving as entering USUALLY

Local events can disrupt balance

- Burning of fossil fuels and other organic materials increases atmospheric Carbon Dioxide CO2 (Traps heat)
- Volcanic eruptions increases atmospheric SO2 and other gases (Overall effect is reduced heat)

CARBON CYCLE-CO2 AND METHANE (CH_4)

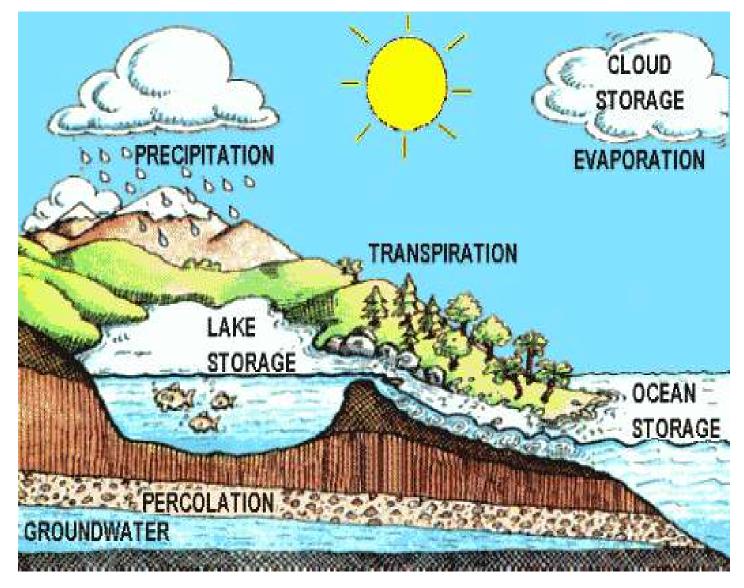


Carbon is crucial to every living thing! Chemical, not physical change Bio-compounds Minerals Plants remove CO₂ from atmosphere, release O₂ Animals (and us) inhale oxygen and exhale CO_2 .

CARBON CYCLES: ROLE OF HUMANS

- Currently-we have exceeded natural fluctuations
 Burning fossil fuels (geosphere→atmosphere)
 Higher than last 500,000, rising faster than ever
 Altered weather
- Oceanic chemistry-increased acidity, decreased productivity
 - +temp and +CO2=decomposition-CO2 returned to air quicker than natural
- Deforestation removes a crucial carbon sink
- Warmer temps=thermal expansion of ocean water
 - Water takes up more space as it warms and expands

WATER CYCLE



WATER CYCLE

Movement of water from different reservoirs and to different phases

Physical processes:

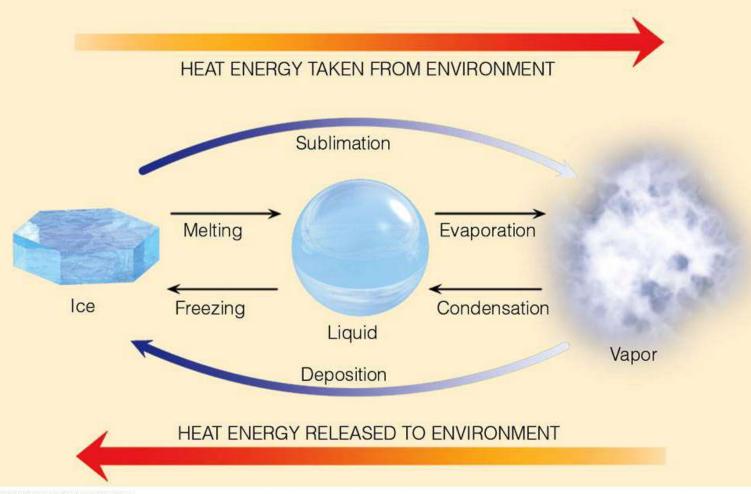
Evaporation, condensation, precipitation, Deposition (infiltration, runoff) and sublimation

86% of evap. Comes from oceans, keeping temps cooler

Phases:

• Liquid, solid (ice), gas (vapor)

PHASE CHANGES



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PHASE CHANGE DEFINITIONS:

- Opposites—one requires/absorbs energy, the other releases
- Sublimation-solid to gas (requires energy)
- Opposition: opposite (what to what?)
- Evaporation: liquid to gas (requires energy)
- Condensation: opposite
- Melting: solid to liquid (requires energy)
- Freezing: opposite

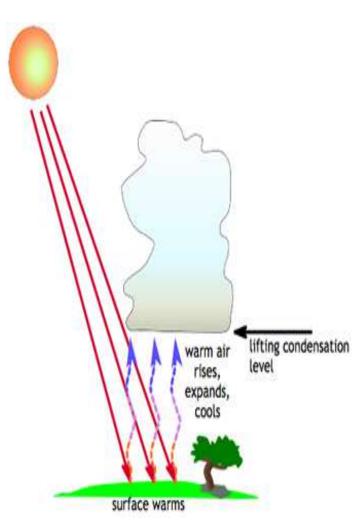
WAIT A SEC, WASN'T THIS UNIT ON ATMOSPHERE????

Yup.

It is all related-I promise! Promotes phase changes Clouds- (water vapor) affect incoming and outgoing radiation Prime regulator of Earth's Energy



ATMOSPHERIC HEATING AND ENERGY

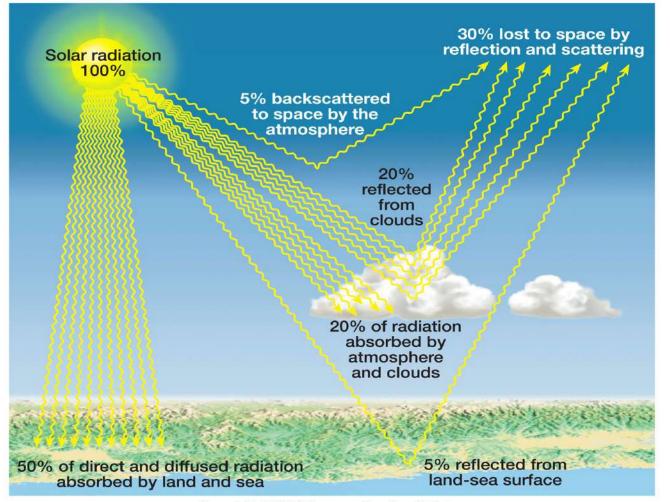


 All energy comes from the sun
 About 50% absorbed by land and sea-the rest radiated back to space

Sun heats ground, ground heats the air

 Warm air rises, expands and cools
 Clouds!!

SURFACE HEATING AND RADIATION



CLOUD FORMATION

- Cold air can hold less water vapor, so it condenses out to form water droplets
- Rising higher and higher→more expansion→more condensation
- Surface COOLING produces fog



HEAT AND THE ATMOSPHERE

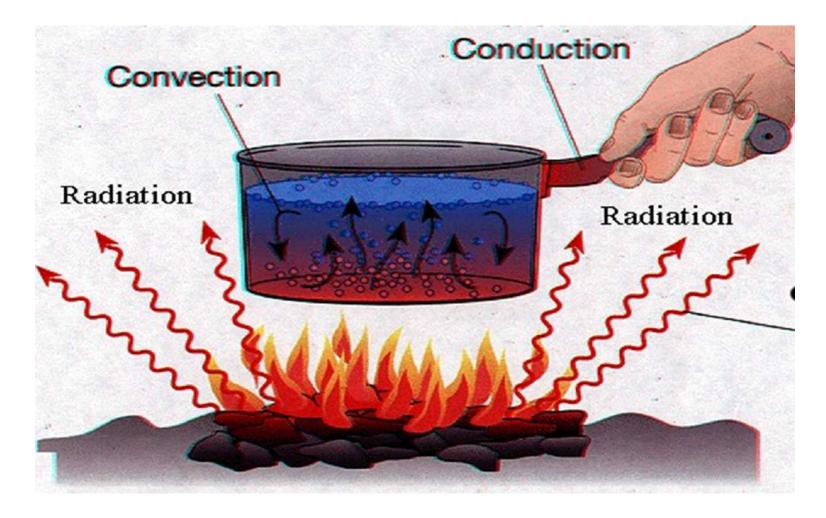
Heat: measure of how fast the atoms or molecules of a substance are moving.

- Temperature: is the average kinetic energy (Energy of movement)
- The faster the movement the higher the heat energy and the higher the temperature

HEATTRANSFER

- Radiation the movement of energy through empty space
 - Light travels from a sun across the solar system to the earth
 - Heat from a fire warms your hand without touching the fire.
- Conduction the movement of energy through a substance, on contact. Atoms or molecules collide with others to make them move
 - Heat moves through the handle of a hot pot to burn your hand
- Convection the rising and falling of a substance due to its change its temperature/density
 - Water in a pot boils, heat in a room rises, cold water sinks

HEAT TRANSFER MECHANISMS



ALBEDO

Measure of a surfaces' reflectivity

Very dark colors have an albedo close to zero (or close to 0%).

Very light colors have an albedo close to 100%



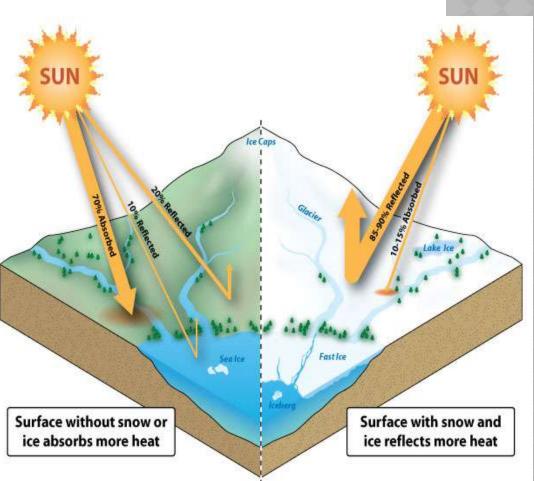
GROUND SURFACE HEATING

Albedo vs absorption Albedo

 Solar energy reflected from Earth back into space

Albedo-cool temps

Absorption-warm temps



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ALBEDO VS ABSORPTION CONTINUED

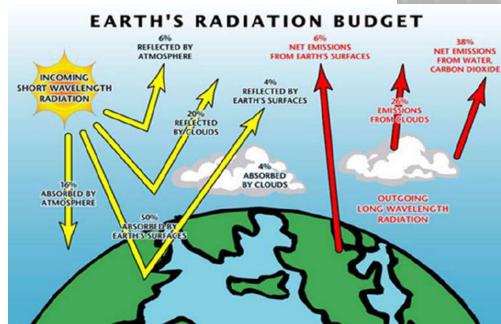
Two surfaces-Land and Water

- Dark materials (most land surfaces) absorb more heat
- Trees-low albedo, high absorption

Snow-high albedo, low absorption

temperature feedback

Clouds



BLACK CARBON

When fossil fuels are not burned completely,

- They produce black carbon or SOOT
- LARGE climate impact:
- Remains in atmosphere for days-weeks, absorbing sunlight and generating heat
 - When it settles, it darkens snow and ice, decreasing albedo, warming the area, melting the ice (feedback)





HUMAN IMPACTS ON THE ATMOSPHERE

Air pollution-any airborne gas or particle that occurs at a concentration capable of harming humans or the environment

Dust, pollen and mold can be suspendedSMOG

- Brownish haze formed mainly from auto exhaust
- When sunlight interacts with smog, can cause
 <u>ozone</u> to form
 - Although atmospheric ozone is helpful to us, groundbased ozone can cause respiratory problems
 - From the term Smoky Fog

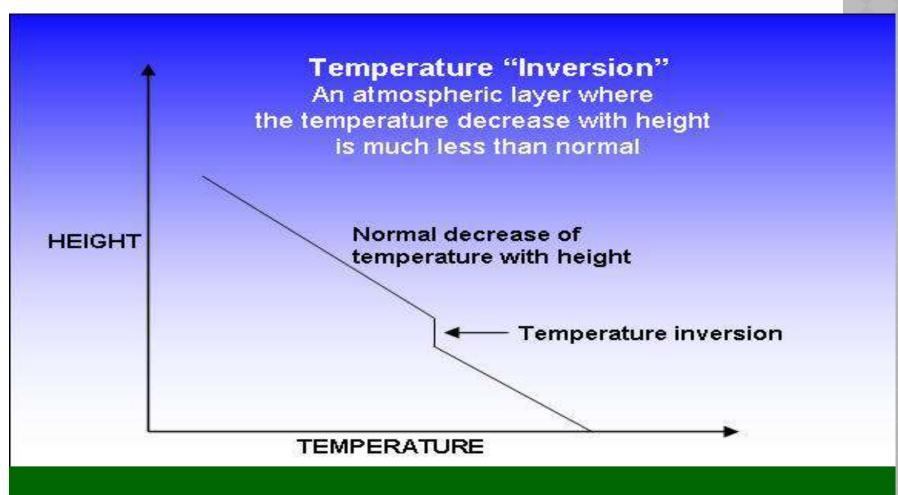
TEMPERATURE INVERSIONS

- When the air at Earth's surface is colder than the air above
- Convection does not occur
- Warm air above acts as a lid, trapping pollutants
- Smog rises to dangerous levels





TEMPERATURE INVERSION

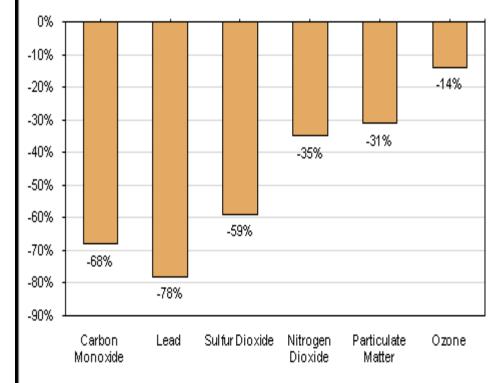


MAJOR POLLUTANTS

| | MAJOR Sources | HEALTH EFFECTS | ENVIRONMENTAL EFFECTS |
|-----------------|---------------------------------------|--------------------------------------------------------------------------------------|----------------------------------------------------------------------------------------------------|
| SO ₂ | Industry | Respiratory and cardiovascular illness | Precursor to acid rain, which damages lakes, rivers, and trees; damage to cultural relics |
| NO _x | Vehicles; industry | Respiratory and cardiovascular illness | Nitrogen deposition leading to over- fertilization and eutrophication |
| PM | Vehicles; industry | Particles penetrate deep into lungs and can enter bloodstream | Visibility |
| CO | Vehicles | Headaches and fatigue, especially in people with weak cardiovascular health | |
| Lead | Vehicles (burning leaded gasoline) | Accumulates in bloodstream over time; damages nervous system | Fish/animal kills |
| Ozone | Formed from reaction of NO, and VOCs | Respiratory illness | Reduced crop production and forest growth; smog precursor |
| VOCs | Vehicles; industrial processes | Eye and skin irritation; nausea; headaches; carcinogenic | Smog precursor |

Common pollutants

Percentage change in concentrations of six common air pollutants, 1990-2008



Source: "Air Trends", June 2009, En vironmental Protection Agency

See also: Australian National Pollutant Inventory - Substance Fact Sheets

ACID RAIN

Pollutants (sulfur dioxide and nitrogen oxides) react with water vapor in the air

- acid rain lowers pH
- Most life has limited range of pH

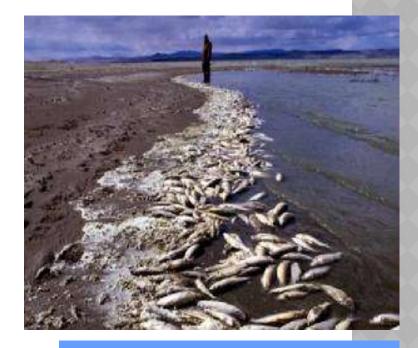
• Lakes and streams-large fish death

Regional Forests-strips away vital nutrients from soil, limiting tree growth and increases vulnerability to disease

■→tree death→less CO2 removal

Damages structures-limestone and marble buildings and gravestones, Parthenon in Greece



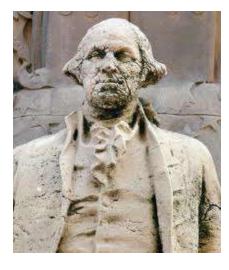


Acid Rain Effects on Sculptures





1969





OZONE DEPLETION

Not a layer...

Chloroflourocarbons (CFC's)

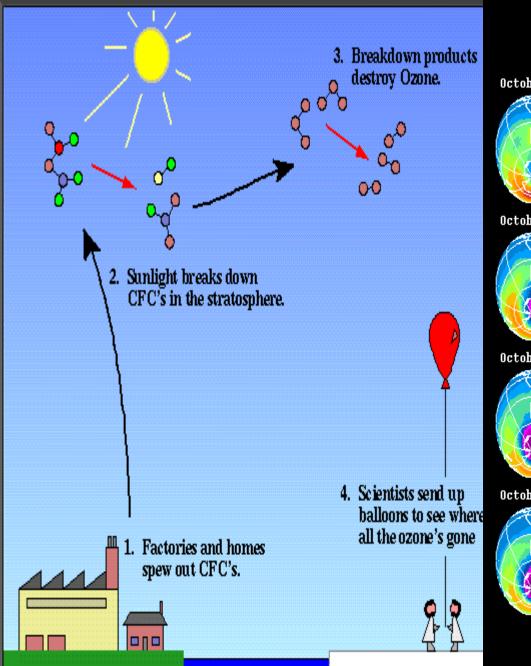


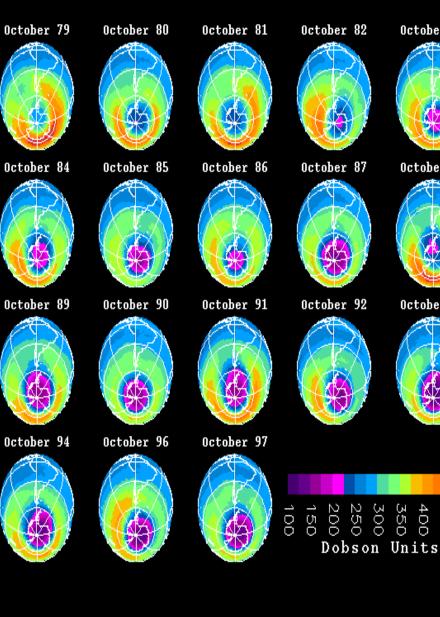
from aerosol sprays-breaks apart ozone into O2 and a free radical

• "PFC's" have been proposed to re-unite the ozone

●Lack of UV protection→skin cancer, harm food crops, destroy phytoplankton (largest CO2 absorbers)

Antarctic Spring "hole"





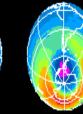
TOMS Total Ozone Monthly Averages



October 8

October 8

October 87



October 92

250

003 50



October 9

450 400

FEEDBACK LOOPS

WATER: Atmosphere → water → atmosphere

- Condensing to form rain, running off into oceans, evaporating or collecting in deposits
- Can also leave oceans through tectonic/volcanic activity
- Carbon Dioxide—LIFE has affected the atmosphere!
- ●CO2 \rightarrow temperature!
- As forests bloomed and died, microbes decomposed material and released CO2
 - As layer upon layer of dead material piled up with no contact to Oxygen, level of CO2 dropped \rightarrow cooled the planet
 - +MY of heat/pressure, buried material is transformed into coal, petroleum and natural gas (hence FOSSIL fuel)

FEEDBACK-CONTINUED

- Oceanic carbon sinks-absorb the largest amount of CO2 (plankton)
 - Sequestering by rocks is slower than CO2 produced
 - ■Increases oceanic salinity→decrease in oceanic biosphere (everything dies)
 - Decrease in biosphere=less CO2 uptake
- Carbon dioxide+ water=carbonic acid
 - Absorbed by rocks through weathering

SUMMARY:

- Formation of the Atmosphere---life
- Layers of atmosphere
- Gases in the atmosphere
- Phase changes
- Water, the water cycle, and water storage
- Ozone, smog and temperature inversions
- Acid rain, black carbon and pollution
- Heat transfer, albedo vs. absorption
- FEEDBACK LOOPS—examples of