

NOTES: Ch 8 – Metabolism and Enzymes

8.1 - METABOLISM

- Metabolism is the _____
- Metabolism arises from interactions between molecules within the cell

Organization of the Chemistry of Life into Metabolic Pathways

- A metabolic pathway begins with a specific molecule and _____
- Each step is catalyzed by a _____

Metabolism includes reactions that are:

- **CATABOLIC** pathways _____
_____ into simpler compounds

- **ANABOLIC** pathways _____
from simpler ones

Forms of Energy

- Energy is the _____
- Energy exists in various forms, some of which can perform work
- **Kinetic energy** is energy associated with _____
 - Heat (thermal energy) is kinetic energy associated with random movement of atoms or molecules
- **Potential energy** is energy that matter possesses because of its location or structure
 - Chemical energy is _____ () _____ available for release in a chemical reaction
- _____

Energy Transformations:

- A closed system is isolated from its surroundings
- In an open system, energy and matter can be transferred between the system and its surroundings
- _____ !

Living things have order!...this takes energy to achieve!

8.2 – Free Energy

- **FREE ENERGY:** the portion of a system's energy that is _____
- Systems tend to change spontaneously _____ ()

Exergonic and Endergonic Reactions in Metabolism

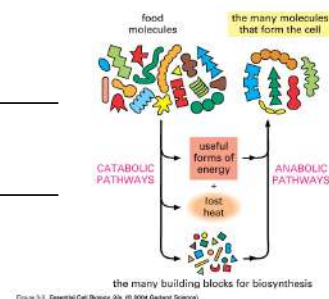
- An **exergonic reaction** proceeds with a net _____
and is spontaneous
- An **endergonic reaction** _____ from its surroundings
and is nonspontaneous

8.4 - ENZYMES:

Enzymes speed up metabolic reactions by lowering energy barriers

- A catalyst is a chemical agent that speeds up a reaction _____

- An enzyme is a _____



- Hydrolysis of _____ by the enzyme _____ is an example of an enzyme-catalyzed reaction

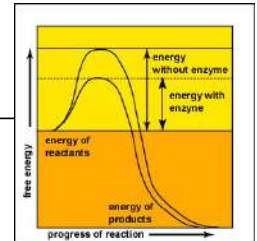
The Activation Energy Barrier

- Every chemical reaction between molecules involves _____ and _____
- The initial energy needed to start a chemical reaction** is called the free energy of activation, or _____

- Activation energy is often supplied in the form of heat from the surroundings

How Enzymes Lower the E_A Barrier:

- Enzymes catalyze reactions by _____
- Enzymes do not affect the change in free-energy (ΔG); instead, they hasten reactions that would occur eventually anyway



Substrate Specificity of Enzymes

- The reactant that an enzyme acts on is called the enzyme's _____
- The enzyme binds to its substrate, forming an _____
- The _____ is the region on the enzyme where the substrate binds
- ENZYMES** are very selective for which reaction they will catalyze
- ENZYMES** are not changed or "used up" by a reaction; _____
- ACTIVE SITE** = region of an enzyme which binds to the substrate

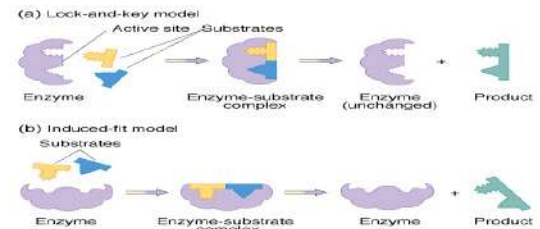
-is usually a _____

-determines enzyme's specificity

- " " between shape of enzyme's active site and shape of substrate

("Lock-and-Key" analogy)

- INDUCED FIT:** a change in the shape of an enzyme's active site, which is induced by the substrate
- Induced fit of a substrate brings chemical groups of the active site into positions that enhance their ability to catalyze the reaction



Catalysis in the Enzyme's Active Site:

- In an enzymatic reaction, the substrate binds to the active site
- The active site can lower an E_A barrier by

- _____

- _____

-Providing a favorable microenvironment

-Covalently bonding to the substrate

Enzyme Reaction Rate:

Substrate Concentration:

-the higher the substrate concentration, _____

-if the substrate concentration is high enough, the enzyme is saturated; in this case, the reaction rate can be increased by _____.

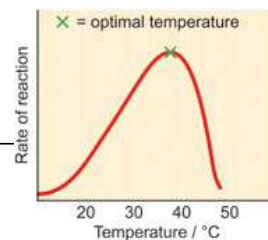
Factors Affecting Enzyme Activity

1) Temperature:

-as temp increases, _____

-if temp gets too high, enzyme denatures and _____

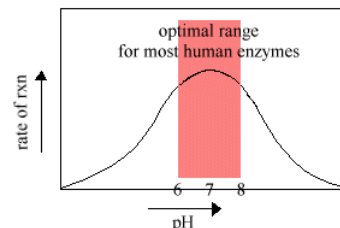
-optimal range for human enzymes: _____



2) pH:

-optimal range for most enzymes: **pH 6-8**

-some enzymes operate best at extremes of pH (e.g. digestive enzyme pepsin, found in the acidic environment of the stomach, works best at pH 2)



3) Cofactors: small non-protein molecules _____

-may bind to active site or substrate

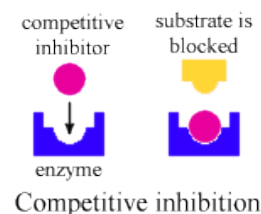
-some are inorganic (e.g. Zn, Fe, Cu)

-some are organic and are called **coenzymes** (e.g. vitamins)

4) Enzyme inhibitors:

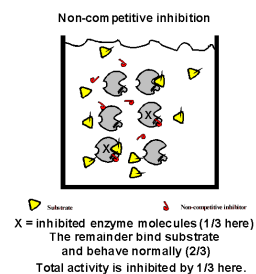
-COMPETITIVE INHIBITORS: chemicals that resemble an enzyme's normal substrate and compete with it for the active site

→ _____
(example: _____)



-NONCOMPETITIVE INHIBITORS: enzyme inhibitors that do not enter the active site, _____ molecule

- cause enzyme to change its shape so active site cannot bind substrate (less effective!)
- may act as metabolic poisons (e.g. DDT, some antibiotics)



8.5 - Regulation of enzyme activity helps control metabolism

- Chemical chaos would result if a cell's metabolic pathways were not tightly regulated
- To regulate metabolic pathways, the cell switches on or off the genes that encode specific enzymes

Allosteric Regulation of Enzymes

- Allosteric regulation is the term used to describe cases where a protein's function at one site is affected by binding of a regulatory molecule at another site
- Allosteric regulation may either inhibit or stimulate an enzyme's activity

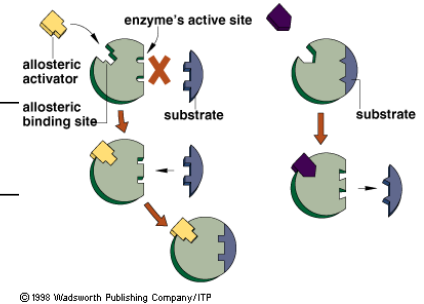
Allosteric Activation and Inhibition

- Most allosterically regulated enzymes are made from polypeptide subunits

- Each enzyme has _____ and _____ forms

- The binding of an **activator** _____ of the enzyme

- The binding of an **inhibitor** _____ of the enzyme



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- **Cooperativity** is a form of allosteric regulation that can amplify enzyme activity

- In cooperativity, binding by a substrate to one active site stabilizes favorable conformational changes at all other subunits

Feedback Inhibition

- In feedback inhibition, the end product of a metabolic pathway _____

- Feedback inhibition prevents a cell from wasting chemical resources by synthesizing more product than is needed

Specific Localization of Enzymes Within the Cell:

- Structures within the cell help bring order to metabolic pathways

- Some enzymes reside in specific organelles, such as: _____

