

Coocoo Chemical and Freaky Physical

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Homogeneous vs. Heterogeneous



A homogeneous mixture is when you mix something and you can **NOT** see the pieces or ingredients.

(solution!) A heterogeneous mixture is when you **can** see the pieces or ingredients.

These are some examples of a Homogeneous mixture, (Kool-Aid, Milk, lotion, and Hand sanitizer) These are some examples of a Heterogeneous mixture, (Kool-Aid, oil and water together in a cup, trail mix, and lotion!)

Mixture vs. Compound



A mixture is a substance made by mixing other Substances.

A Compound is composed by two or more seperate elements

A mixture can be either Heterogeneous or Homogeneous, here are some examples of a mixture. (milk, juice, water and oil , lotion and trail mix.)

Chemical vs. Physical Change



Chemical Change Pic-Collage



Physical Change Pic-Collage

PHYSICAL CHANGES



Why does this matter?

It matters because we use the different properties every single day of our lives. Such as using electricity. Playing your favorite song on a CD. Or even Baking a cake!



References

Daniel, L., Hackett, J., Moyer, R., & Vasquez, J.
(2006) *Science*. New
York: Macmillan/McGraw-Hill

Mixtures Facts