

Coocoo Chemical and Freaky Physical

By: Niyana Santiago



Homogeneous vs. Heterogeneous



A homogeneous mixture is when you mix something and you can **NOT** see the pieces or ingredients.
(solution!) A heterogeneous mixture is when you **can** see the pieces or ingredients.

These are some examples of a Homogeneous mixture, (Kool-Aid, Milk, lotion, and Hand sanitizer) These are some examples of a Heterogeneous mixture,(Kool-Aid, oil and water together in a cup, trail mix, and lotion!)

Mixture vs. Compound

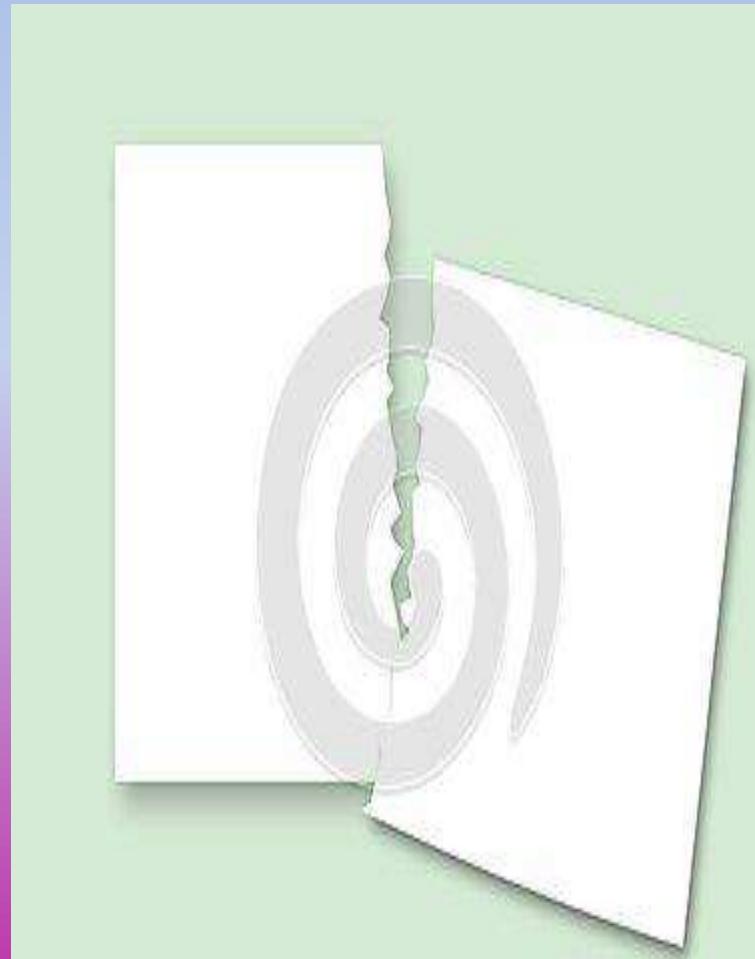


A mixture is a substance made by mixing other Substances.

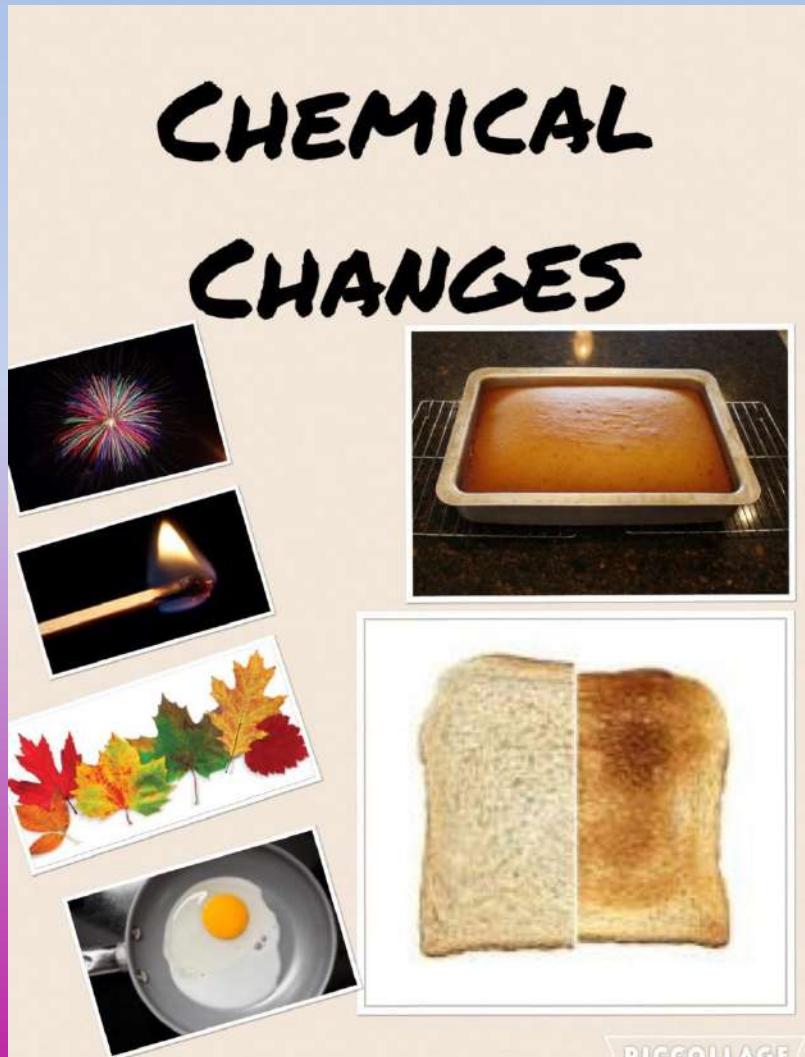
A Compound is composed by two or more seperate elements

A mixture can be either Heterogeneous or Homogeneous, here are some examples of a mixture. (milk, juice, water and oil , lotion and trail mix.)

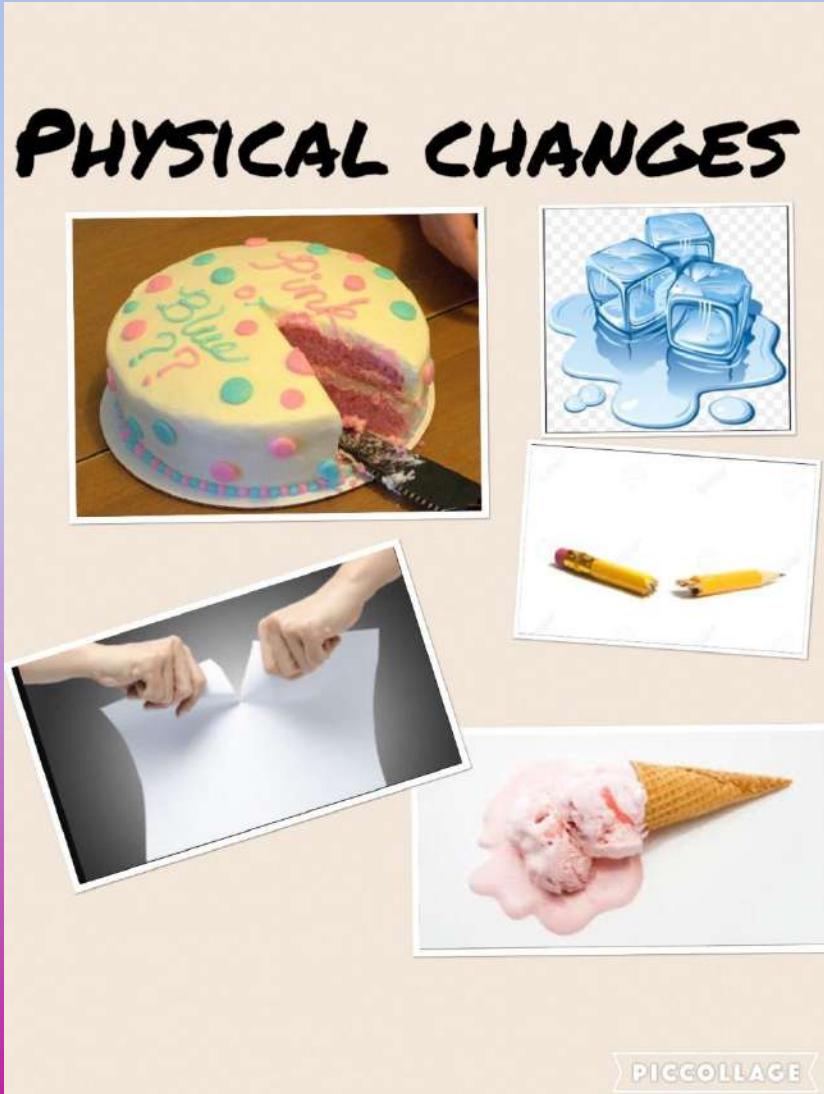
Chemical vs. Physical Change



Chemical Change Pic-Collage



Physical Change Pic-Collage



Why does this matter?

It matters because we use the different properties every single day of our lives. Such as using electricity. Playing your favorite song on a CD. Or even Baking a cake!



References

Daniel, L., Hackett, J., Moyer, R., & Vasquez, J.
(2006) *Science*. New
York: Macmillan/McGraw-Hill

Mixtures Facts