

Mass

[given] g
↑
from problem
of [known]

(conversion
g → mol)

1 mol
of [known]

Mole
(also mole ratio)

[#] mol of
↑
coefficient
from balanced
equation
[unknown]

[Molar mass] g
↑
from
PT
of
[known]

[#] mol of
↑
coefficient
from balanced
equation
[known]