		Name	Period	Date		
		Mac	romolecules: Online Homework	Assignment		
]			em.htm This is a good summary			
			gical molecules,chemistry, is	s carbon chemistry. What makes		
	car	bon so important?				
I.	2.	Monosaccharides are	sugars. A disacchari	de is formed when		
	mo	onosaccharides are joined tog	gether by a reaction known as a	, or		
	col	ndensation, synthesis. The	splitting apart reaction is called	. Play the		
	an	imation for each. What is r	removed as a result of dehydration s	ynthesis?		
	3.	Polysaccharides (poly =	, =	sugar) are made of many sugar		
		molecules joined together b	ey synthesis r make a large molecule, the resulting	reactions. When many repeating		
			make a large molecule, the resulting	g molecule is called a		
		·				
	4.	Starch is the long term	storage form of glucose	e in It is found in		
		abundance in the	we eat, (_, wheat, barley and rye) and in		
		many vegetables, such as _	and corn.			
		Glycogen is the	energy storage form	of glucose. This polymer has the		
			n the same way as they are in the pla	nt starch. How is glycogen		
		different from starch?				
	5.	5. Cellulose differs from starch and glycogen in				
		is a	polymer o	Cellulose f glucose which is found in the		
		15 d	of plants. No	has an enzyme to break		
		down cellulose, so it canno	t be digested by them directly. We e	liminate ingested cellulose as		
		<u> </u>	, who eat wood, in their digestive tracts to bre	, depend upon		
			in their digestive tracts to bre	ak down the cellulose.		
	6. Chitin is made of glucose molecules linked in the same way they are linked in					
			, making it equally groups (NH ₂) attached to the gluco	It differs from cellulose		
		by having	groups (NH ₂) attached to the gluco	ose molecules. Chitin forms the		
			n of all arthropods (e.g	, spiders, lobsters and		
ΤΤ	7	crabs). Linids are defined by their	They do no	at dissolve in polar solvents like		
	7.	Lipids are defined by their	. They do no	dussolve in polar solvents like		
		and the	s are are all lipids. What is their	function?		
		One group, the	are the major structura	al elements of membranes.		
	8.	Triglycerides (fats) are com	are the major structura posed of a ba	ckbone and		
		acids. Wa	tch the animation. What process lin	nks the glycerol and fatty acids		
		together?	What breaks them apart? acids contain all of the			
	9.	fatty a	icids contain all of the	they can hold. There are		
		no carbon to carbon	bonds in saturated fatty fats and are believed to cause b	y acids. Saturated fatty acids are		
		typical of	iats and are believed to cause b	поскаде от		

which can lead to strokes and heart attacks.

	fatty acids do not contain all of the possible. One more to double bonds will be present in the carbon characteristic bonding to bonding to bonding to bonding to bonding to	
		ain.
	Note that each carbon involved in the double bonding has only one bond left for bonding to	
		Amino
		group
		\ m (
		\ H
		и— N—
		п
	Such molecules are not completely loaded with hydrogen so they are	Amin
11.	Proteins are polymers of This means that proteins a repeating units of molecules which have an (NH ₂) group at one end	are
	repeating units of molecules which have an (NH ₂) group at one end	
	which has a variable group attached to it. The letter stands for the var n	
	be substituted here. Label the diagram to show the key features of an amino acid	
12.	Proteins are formed by (condensation) synthesis which links ami	no
	acids together. Watch the animation. In this reaction, is removed and the from the amino end of one amino acid is joined to the is dwher many amino acids are joined together by bonds.	
	the from the amino end of one amino acid is joined to the	
	from the other end of another amino acid. A is \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	1
	many amino acids are joined together by bonds.	
	List the functions of proteins:	
	•	es. A
	•	es. A
	•	es. A nine the
. 14.	and, the two nucleic acids, are polymers of nucleotide nucleotide is made of a, a 5 carbon and a and the bases are ader (A), cytosine (C), guanine (G), and thymine (T). In RNA the sugar is and	es. A nine the
.14.	and, the two nucleic acids, are polymers of nucleotide nucleotide is made of a, a 5 carbon and a and the bases are ader (A), cytosine (C), guanine (G), and thymine (T). In RNA the sugar is and bases are A, C, G, and (U) instead of T.	
. 14. 15.	and, the two nucleic acids, are polymers of nucleotide nucleotide is made of a, a 5 carbon and a base. The sugar in DNA is and the bases are ader (A), cytosine (C), guanine (G), and thymine (T). In RNA the sugar is and bases are A, C, G, and (U) instead of T. DNA is composed of nucleotide chains connected to each other by	
. 14. 15.	and, the two nucleic acids, are polymers of nucleotide nucleotide is made of a, a 5 carbon and a and the bases are ader (A), cytosine (C), guanine (G), and thymine (T). In RNA the sugar is and bases are A, C, G, and (U) instead of T.	

5. Biomolecules: The Lipids Wisconsin Online

6. Biomolecules: Proteins Wisconsin Online