WordWise	
Answer the questions by writing the corr Use the circled letter in each term to find write a definition for the hidden word.	rect vocabulary term in the blanks. the hidden vocabulary word. Then,
Clues	Vocabulary Terms
A mixture that results when substances dissolve to form a homogeneous mixture	<u> </u>
A substance that can be broken down into two or more simpler substances	
A change in which the composition of matter stays the same	
A solid that forms and separates from a liquid mixture	
A substance that cannot be broken down into simpler substances	<u>Q</u>
The ability of a material to allow heat to flow	
A classification for matter that always has the same composition	Q
The ability of a material to burn	Q
A homogeneous mixture containing particles that scatter light	_Q
The temperature at which a substance changes from a liquid to gas	
Hidden Term:	_
Definition:	

Class

Date _

write a acjustion for the matter word.	•
Clues	Vocabular
What is the process that changes a substance from a liquid to a gas below the substance's boiling point?	<u> </u>
Which gas law states that the volume of a gas is directly proportional to its temperature?	
What is the phase change in which a substance changes directly from a gas to a solid?	_Q
In what state does matter have both a definite shape and a definite volume?	Q
What is the phase change in which a substance changes from a gas to a liquid?	
What is the phase change in which a substance changes directly from a solid to a gas?	_Q
During what type of phase change does a substance release energy to its surroundings?	
During what type of phase change does a substance absorb energy from its surroundings?	Ω
Hidden Term: Definition:	

Chapter 3 States of Matter

WordWise

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Namo		
1/1/2/10/4		
LIGHER		
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Chapter 4 **Atomic Structure**

WordWise

Solve the clues to determine which vocabulary terms from Chapter 4 are hidden in the puzzle. Then find and circle the terms in the puzzle. The terms may occur vertically, horizontally, or diagonally.

e	m	а	s	s	n	u	m	b	e	r	u	'n
n	1	o	r	b	Ì	· t ·	- a -	ŀ	x	а	p	i
r	e	e	n	1	t	p	t	s	р	b	k	s
g	n	a	c	а	s	r	d	c	r	h	l	o
b	e	1	d	t	g	o	f	1	8	g	a	t
1	r	t	s	0	r	t	g	r	n	b	t	o
n	g	Ż	b	m	0	o	p	1	q	d	c	p
p	y	q	p	i	ù	n	n	m	a	s	s	e
s	1	n	m	c	n	n	u	c	l	e	u	s
t	e	u	e	n	d	r	i	o	l	k	m	r
r	v	С	1	u	s	v	а	b	t	0	р	k
Ż	e	1	x	m	t	w	e	s	r	n	u	е
p	1	e	m	ь	a	r	1	e	t	a	b	d
b	s	a	q	e	t	Z	o	Ç	m	r	n	\mathbf{k}
r	t	s	i	r	e	h	İ	n	s	f	1	t



Clues **Hidden Words** Dense, positively charged mass in the center of an atom Positively charged subatomic particle found in the nucleus Neutral subatomic particle found in the nucleus Number of protons in an atom of an element Sum of the protons and neutrons in the nucleus of an atom Atoms of the same element having different numbers of neutrons Possible energies that electrons in an atom can have Visual model of the most likely locations for electrons in an atom Region of space where an electron is likely to be found Term for an atom whose electrons have the lowest

possible energies