## Instructions for Froot Loop & Pretzel Compound Building Activity

Directions for *lonic Compounds* -

- 1. Place appropriate element cards on the table.
- 2. Place Froot Loops around the elements to represent the electrons. (Use different colors for different elements)
- 3. Transfer the appropriate "Froot Loop" electrons around until you come up with the correct formula for the ionic compound.
- 4. Once you have finished, raise your hand so your teacher can check your model.
- 5. Draw the electron dot diagrams for the final compound on your answer sheet.

Directions for Molecular Compounds -

- 1. Place the appropriate element cards on the table.
- 2. Place Froot Loops around the elements to represent the electrons. (Use different colors for different elements)
- 3. Once you have built the electron dot structure, build the molecular structure by using pretzel sticks to represent the bonds.
- 6. Once you have finished, raise your hand so your teacher can check your model.
- 7. Draw the structural formula for the final compound on your answer sheet.

## Froot Loop and Pretzel Compound Building Activity Answer Sheet

## IONIC COMPOUNDS

Build a model of the electron dot structure using Froot Loops, then write the correct empirical formula in the boxes below:

Calcium and Chlorine	Aluminum and Chlorine
Magnesium and Nitrogen	Cesium and Chlorine
Calcium and Sulfur	Barium and Bromine
Potassium and Nitrogen	Potassium and Oxygen

## MOLECULAR COMPOUNDS

Build a model of the molecular structure using Froot Loops and Pretzel sticks. Then draw the structural formula in the boxes below:

O <sub>2</sub>	HCN
HF	H <sub>2</sub> O <sub>2</sub>
CH₄	H <sub>2</sub> O
F <sub>2</sub>	CO2

Ca	Cl	Cl	Cl	Al
K	K	Mg	Mg	Mg
S	Ba	Br	Br	N
0	0			
		С	K	Cs
	N			