

Name: _____
Hour: _____ Date: _____

Chemistry: *Ionic Compounds: Polyatomic Ions*

Write the name of each of the following compounds.

- | | |
|---------------------------------------|-----------|
| 1. NH_4Cl | 1. _____ |
| 2. HClO_2 | 2. _____ |
| 3. $\text{Ca}(\text{BrO}_3)_2$ | 3. _____ |
| 4. BeSO_4 | 4. _____ |
| 5. $(\text{NH}_4)_3\text{N}$ | 5. _____ |
| 6. NH_4NO_3 | 6. _____ |
| 7. $\text{Sr}_3(\text{PO}_4)_2$ | 7. _____ |
| 8. $\text{Zn}(\text{ClO}_3)_2$ | 8. _____ |
| 9. AgIO_3 | 9. _____ |
| 10. $\text{K}_2\text{Cr}_2\text{O}_7$ | 10. _____ |

Write the chemical formula for each of the given names.

- | | |
|------------------------|-----------|
| 11. sodium chromate | 11. _____ |
| 12. barium nitrate | 12. _____ |
| 13. ammonium sulfate | 13. _____ |
| 14. aluminum hydroxide | 14. _____ |
| 15. calcium phosphate | 15. _____ |
| 16. cesium cyanide | 16. _____ |
| 17. sodium nitrite | 17. _____ |
| 18. calcium acetate | 18. _____ |
| 19. beryllium chlorite | 19. _____ |
| 20. rubidium sulfite | 20. _____ |

Name: _____
Hour: _____ Date: _____

Chemistry: Ionic Compounds: Polyatomic Ions with Multiple-Charge Cations

Write the name of each of the following compounds.

- | | |
|--|-----------|
| 1. $\text{V}(\text{ClO}_3)_5$ | 1. _____ |
| 2. $\text{Re}(\text{SO}_4)_3$ | 2. _____ |
| 3. $\text{Os}(\text{IO}_3)_3$ | 3. _____ |
| 4. $\text{Ir}_3(\text{PO}_4)_4$ | 4. _____ |
| 5. $\text{Pd}(\text{SO}_3)_2$ | 5. _____ |
| 6. AuNO_3 | 6. _____ |
| 7. FePO_3 | 7. _____ |
| 8. $\text{Ni}(\text{BrO}_3)_2$ | 8. _____ |
| 9. $\text{Pb}(\text{CN})_4$ | 9. _____ |
| 10. $\text{Mn}_2(\text{Cr}_2\text{O}_7)_7$ | 10. _____ |

Write the chemical formula for each of the given names.

- | | |
|--------------------------------|-----------|
| 11. copper (II) hydroxide | 11. _____ |
| 12. rhenium (VI) nitrate | 12. _____ |
| 13. niobium (III) sulfate | 13. _____ |
| 14. platinum (IV) iodate | 14. _____ |
| 15. molybdenum (III) phosphate | 15. _____ |
| 16. titanium (II) cyanide | 16. _____ |
| 17. gold (I) thiosulfate | 17. _____ |
| 18. iron (III) acetate | 18. _____ |
| 19. cobalt (II) chlorite | 19. _____ |
| 20. tin (IV) sulfite | 20. _____ |

Compound Names and Formulas

For the list on the left, name the compound. For the list on the right, give the chemical formula that corresponds to the name

Name		Formula	
1) NaF	_____	13) potassium fluoride	
2) K ₂ CO ₃	_____	14) ammonium sulfate	
3) MgCl ₂	_____	15) magnesium iodide	
4) Be(OH) ₂	_____	16) copper (II) sulfite	
5) SrS	_____	17) aluminum phosphate	
6) Cu ₂ S	_____	18) lead (II) nitrite	
7) ZnI ₂	_____	19) cobalt (II) selenide	
8) Ca ₃ (PO ₄) ₂	_____	20) silver cyanide	
9) NH ₄ I	_____	21) copper (II) bicarbonate	
10) Mn(NO ₃) ₃	_____	22) iron (II) oxide	
11) FePO ₄	_____	23) lithium cyanide	
12) CoCO ₃	_____	24) lead (IV) sulfite	

Review-- Naming Chemical Compounds

The following are a good mix of naming and formula writing problems to help you get some practice. I will expect that you know how to name both ionic and covalent compounds in your work.

Name the following chemical compounds:

- 1) NaBr _____
- 2) $\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$ _____
- 3) P_2O_5 _____
- 4) $\text{Ti}(\text{SO}_4)_2$ _____
- 5) FePO_4 _____
- 6) K_3N _____
- 7) SO_2 _____
- 8) CuOH _____
- 9) $\text{Zn}(\text{NO}_2)_2$ _____
- 10) V_2S_3 _____

Write the formulas for the following chemical compounds:

- 11) silicon dioxide _____
- 12) nickel (III) sulfide _____
- 13) manganese (II) phosphate _____
- 14) silver acetate _____
- 15) diboron tetrabromide _____
- 16) magnesium sulfate _____
- 17) potassium carbonate _____
- 18) ammonium oxide _____
- 19) tin (IV) selenide _____
- 20) carbon tetrachloride _____

Chemis-tree Assignment:

Due December 22 or 23rd (depending on which day you have class.)

1. Pick a chemist or invention or discovery (chemistry related)
2. Research. Yes, you will need to cite your works (MLA or APA)
3. Make an ornament symbolic of your topic
4. Write 2-4 paragraphs on your topic including
 - a. Why it was important (what you learned)
 - b. How your ornament expresses such
 - c. Why you chose that topic

Rubric: (out of 20)

Grade	Neatness	Information	Creativity	Effort
5	Pristine, Typed	Accurate, clear Thorough understanding of topic, its significance, and its relation to the course. Chem Related, works cited	Thought provoking. Original	4 well written ¶
4	Nice, but better with effort	Missing a couple details, but comprehensive and clear. Shows detailed understanding of topic. Chem related, works cited	Colorful, detailed	3 well written or 4 clear Paragraphs
3	Not too shabby.	Chem related, but missing info, shows a basic understanding of topic and significance. Works cited	Standard. Logical	3 understandable ¶
2	Thrown together	Science related, but not quite accurate or thorough, works mentioned. Shows basic understanding of topic and significance.	Not recognizable as it relates to topic	2 understandable ¶
1	messy	Science related, but not totally comprehensible. Does show at least some understanding of topic researched	Messy, thrown together. Not relevant	1 understandable ¶

* can check answers once finished.

Compound Names and Formulas

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1) NaF		13) potassium fluoride
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* 11) FePO ₄		23) lithium cyanide
* 12) CoCO ₃		24) lead (IV) sulfite

* = Transition Metal - use Roman Numerals