

Does Temperature Affect Dissolved Gasses in Water FALL SEMESTER 2022

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timer

NY.	D 4
Name:	Date:
Introduction	
) just like terrestrial organisms. The oxygen nately 20%. The dissolved oxygen percentage of Many factors affect how much oxygen and other
Purpose:	
	fects the ability of water to hold a dissolved gas. he ability of water to hold a dissolved gas? If so,
HYPOTHESIS:	
Materials:	
 12 cc syringe (or larger) syringe cap 200 mL beaker thermometer 	 hot water tap water ice seltzer/diet soda

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Procedure:

- 1. Fill the beaker three-quarters full with water at a temperature between oo and 650 C.
- 2. Fill a 12 cc syringe with 7cc of seltzer.
- 3. Holding the syringe pointing upward, remove any excess air, leaving 5cc of soda/seltzer in the syringe. DO NOT SQUIRT THE SELTZER
- 4. Cap syringe.
- **5.** Place capped syringe in beaker, pointing capped end of syringe down.
- **6.** Time for three minutes.
- **7.** Measure how far the bottom of the plunger moved.
 - a. For example, starting at 5cc, if the bottom of the plunger moved to 8cc, you would record 3cc of gas.
- **8.** Empty soda/seltzer from syringe.
- **9.** Repeat for a total of three trials at each temperature, determining the average displacement for each temperature.
- 10. Graph the average volume of gas lost from solution.

Milliliters of CO ₂ Released from Seltzer in 3 Minutes						
Temperature (oC)	Trial 1	Trial 2	Trial 3	Average		
О						
5						
10						
15						
20						

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25		
30		
35		
40		
45		
50		
55		
60		
65		

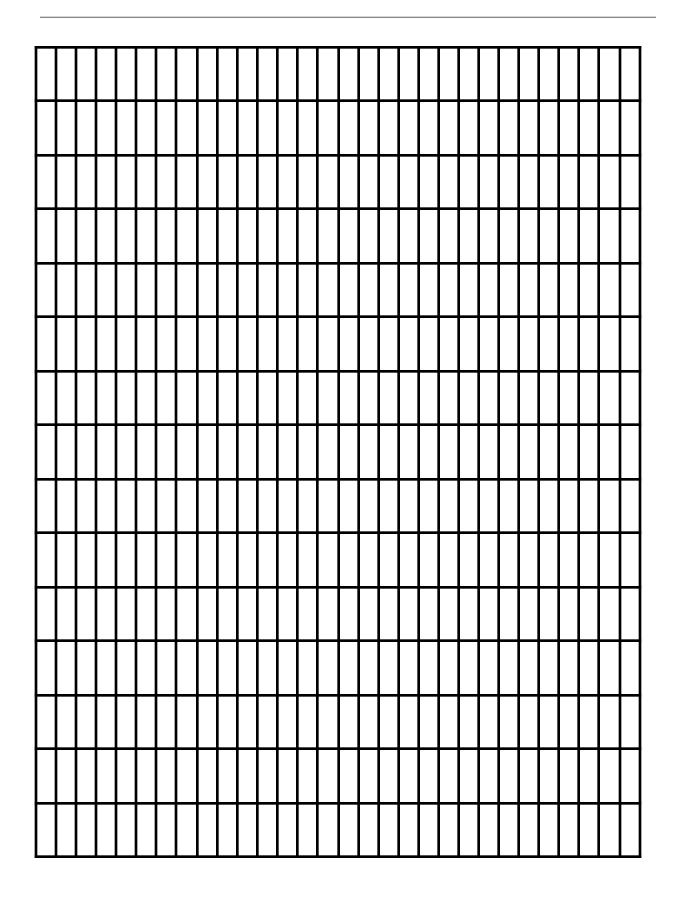
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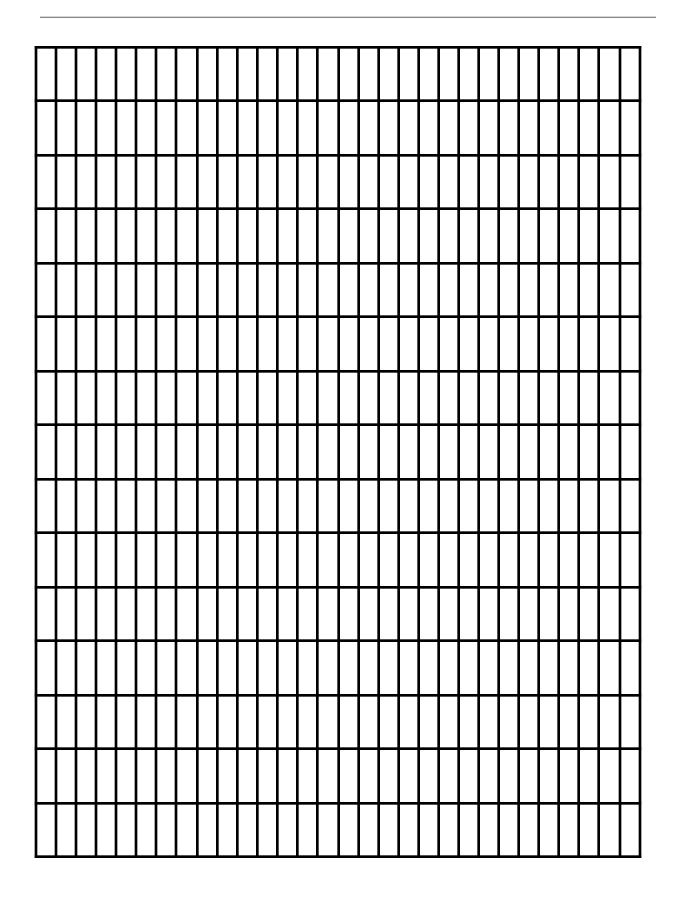
L - Labels

K - Key

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