The Chemistry of Baking

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01 Recipe



Main Ingredients

-	Plain Flour	1 ³ / ₄ cups	Baking Soda	1½ tsp.	Large Eggs	2
	Unsweeten ed Cocoa Powder	³¼ cup	Salt	1 tsp.	Milk	1 cup
	Baking Powder	1½ tsp.	Granulate d White Sug <mark>ar</mark>	2 cups	Vanilla Extract	2 tsp.



Chemical Reactions

Baking Soda



Baking Soda Chemical Reaction



Undergoes a reaction when it is heated.

Why is it used?

CO₂ is a product.

Chemical Formula

NaHCO₃.

Chemical Equation

2NaHCO₃ her H₂O + CO₂ + Na₂CO₃

This is a decomposition reaction!

Baking Soda is Basic

- Batter is acidic
 - Neutralizes with baking soda
- Ingredients affect pH levels
- Affects the flavor





Two Main Purposes

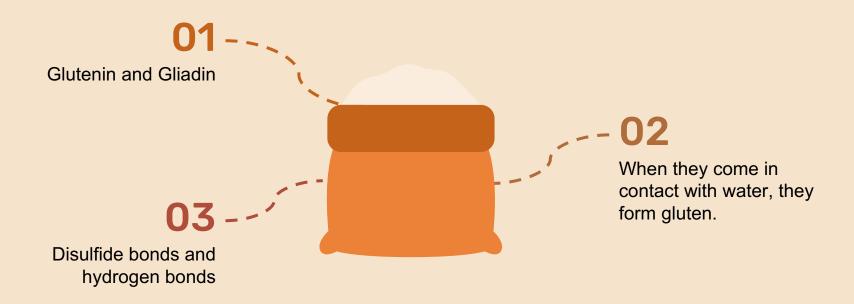
Gluten

Maillard Reaction





Gluten

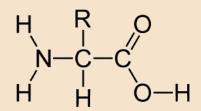




Disulfide Bonds

Two Sulfurs Bond

Amino Acids



Gluten Holds CO₂

Disulfide bonds make a stretchy web.



Maillard Reaction

The amino acids break apart.



02

Hundreds of new compounds

03

It also darkens the color of the surface.







The Purpose of Salt

Strengthens gluten and helps trap CO₂.



It Has Three Functions



Carameliza tion











Caramelization

01

Takes place at 356 degrees Fahrenheit.

04

The sugars react with each other.



02

$$C_{12}H_{22}O_{11}$$
 \longrightarrow
 $C_6H_{12}O_6 + C_6H_{12}O_6$

03

$$C_6H_{12}O_6 \longrightarrow H_2O + C_6H_{10}O_5$$

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Thanks!

Do you have any questions?

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