Chapter 8-2

Pressure

 $\begin{array}{l} Fressure = Force / Area \\ P = F / A \end{array}$

- Area = pi r 2
- SI unit is the pascal (Pa)
- $1 Pa = 1N/m^2$
- Atmosphere pressure at sea level=1.01x10⁵

Baby Bottle

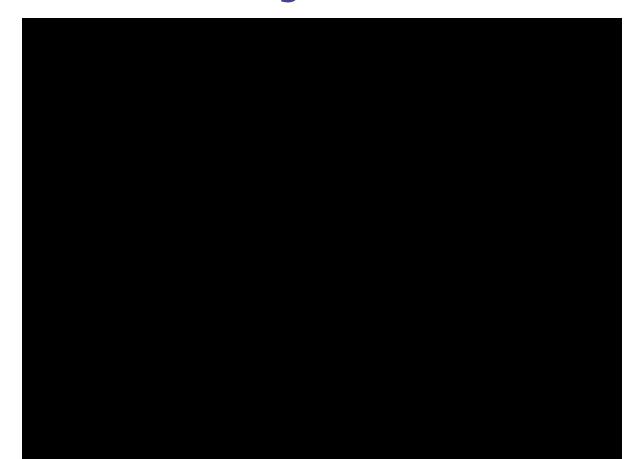


Bottomless Bottle Demo



Demos

Straw popping Bed of Nails Potato gun



Mentos and Coke



Mentos Rocket

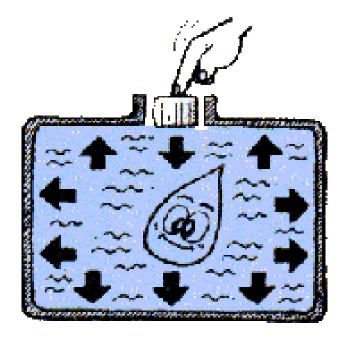


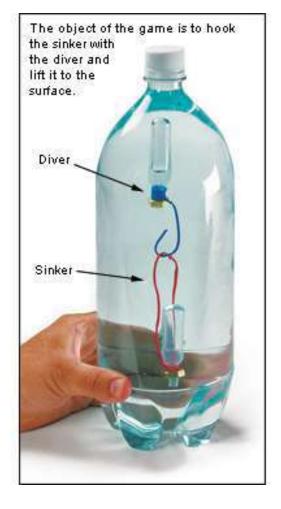
Pascal's Principle Pascal's Principle – Pressure applied to a fluid in a closed container is transmitted equally to every point of the fluid and to the walls of the container. $P = F_1/A_1 = F_2/A_2$

Pascal's Demo

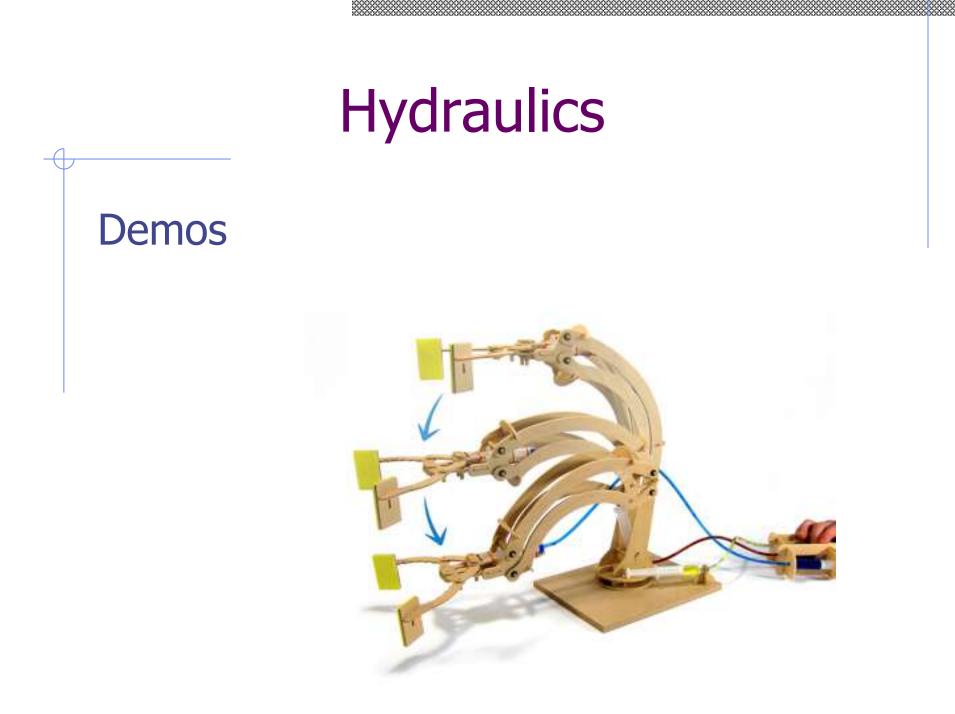


Cartesian Divers





In a car lift, compressed air exerts a force on a piston with a radius of 10 cm. This pressure is transmitted to a second piston with a radius of 15 cm. How large a force must the compressed air exert to lift a 330 N car?



Atmospheric Pressure is pressure from above. The weight of the air in the upper atmosphere exerts a pressure on the layers below.





Tornado

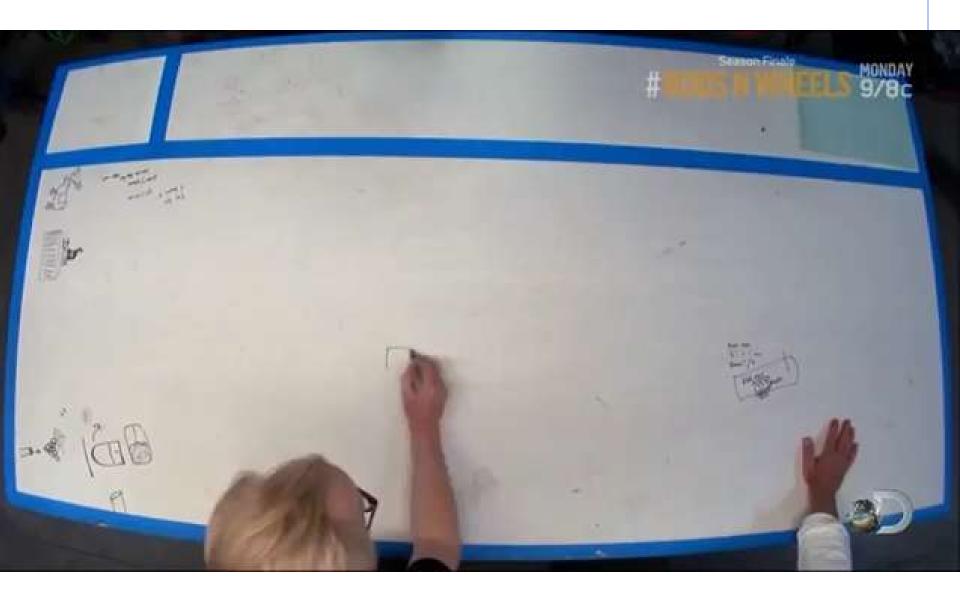


Vortex Cannons and Bubbles

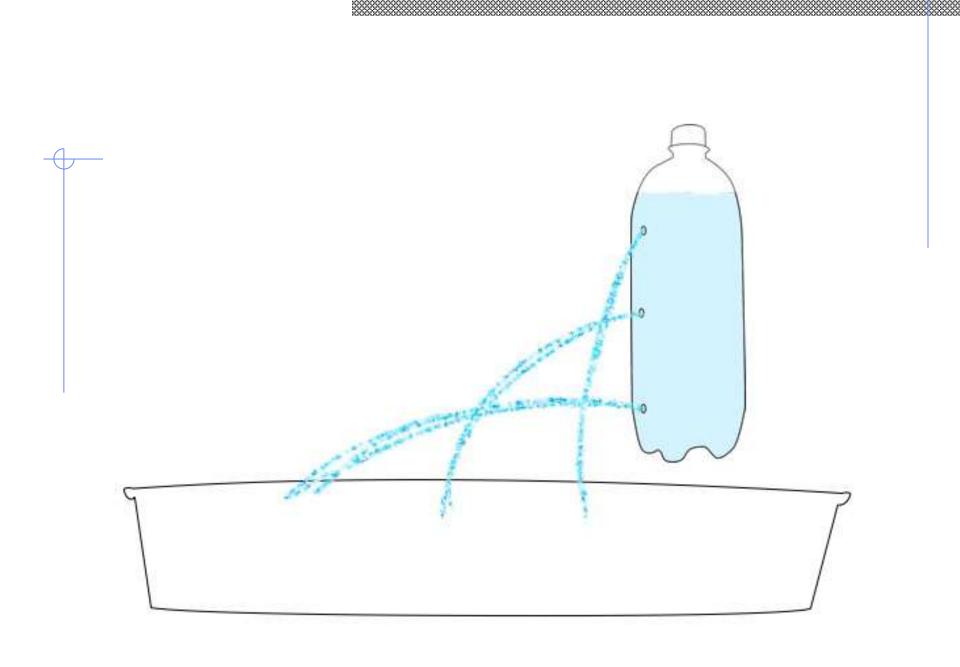


Vacuum Packing

right now, the atmospheric pressure outside the bag is BALANCED by the pressure inside the bag



Pressure and Depth Absolute pressure = atmospheric pressure + (density x gravity x depth) $P = P_{atm} + Dqh$



Kinetic Theory

Kinetic Theory of gases describes the origin of gas pressure. Gas particles are like a collection of billiard balls that constantly collide with one another.As they collide with the wall, they exert a force and this force per unit area is the gas pressure.

<u>Temperature</u>

- The higher the temperature, the faster the particles move.
- As the speed of the particles increase, the amount they hit the wall increases resulting in a higher force and therefore a higher pressure.
- Therefore, temperature and pressure are related.

Ivory Soap Demo



Chapter 8-3 and 8-4 Fluid Flow and Gas Laws

Laminar / Streamline

Laminar – if every particle that passes a particular point moves along the same smooth path. Also called **streamline**.

Different streamlines cannot cross each other.

At any point in a streamline flow, the direction and velocity of the fluid is the same throughout.

Turbulent

Turbulent – the flow of a fluid becomes **irregular** above a certain velocity or under conditions that can cause abrupt changes such as obstacles or sharp turns.

Irregular motions of the fluid, called **eddy currents**, are characteristic of turbulent flows.



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Bottle Demos

Viscosity

- The term viscosity refers to the amount of **internal friction** within a fluid.
- Internal friction occurs when layers of fluid slides past another layer.
- A fluid with a high viscosity flows more slowly through a pipe than does a fluid with a low viscosity.

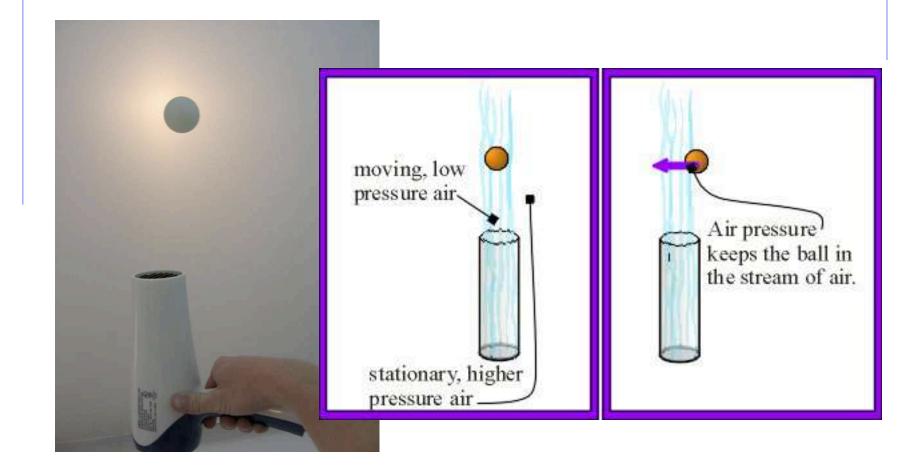
Ideal Fluid

- Ideal fluids are considered nonviscous, so they don't lose kinetic energy due to friction as they flow.
- Ideal fluids are considered steady flow. The velocity, density, and pressure at each point are constant.
- The flow is nonturbulent, which means no eddy currents.

Bernoulli's Principle

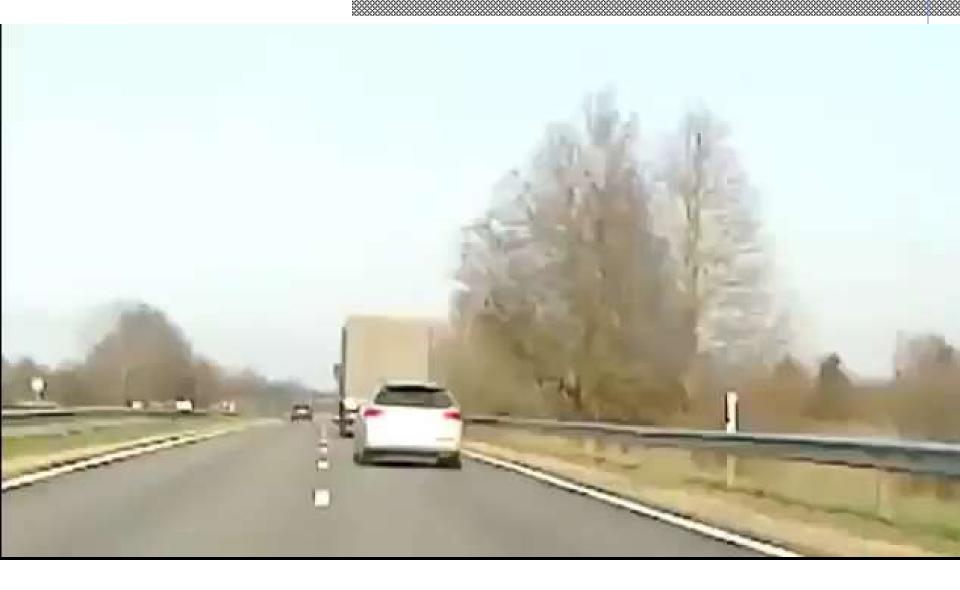
- The speed of fluid depends on cross sectional area.
- The pressure in a fluid is related to the speed of flow.
- Bernoulli's principle The pressure in a fluid decreases as the fluid's velocity increases.

Bernoulli's Demos



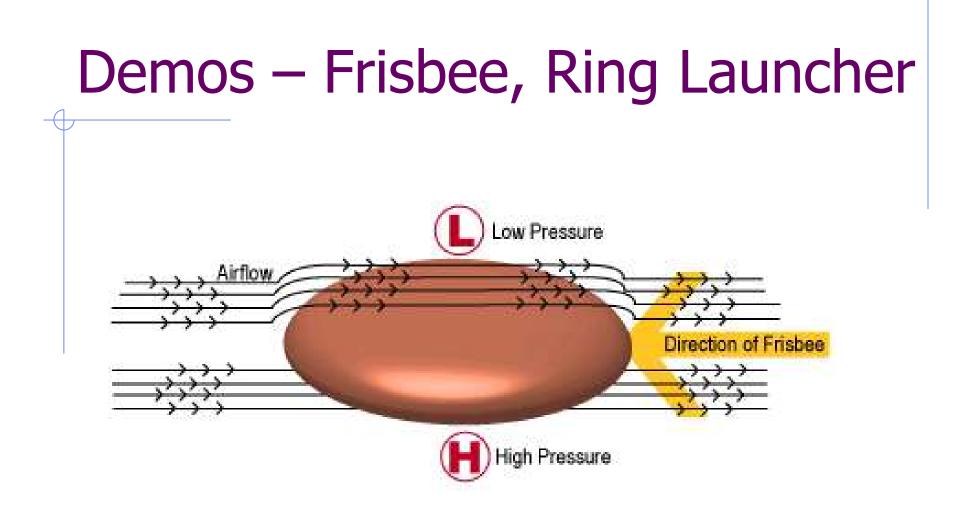
Lift on an Airplane

- The lift on an airplane wing can be explained with Bernoulli's Principle.
- Airplane wings are designed to direct the flow of air so that the air speed above the wing is greater than the air speed below the wing.
- This makes the air pressure above the wing less than the pressure below which creates and upward force called lift.



Bernoulli's Equation

- Pressure₁ + ¹/₂ density x
- velocity₁² + density x gravity x
- height₁ = Pressure₂ + $\frac{1}{2}$
- density x velocity₂² + density x
- gravity x height₂
- $P_1 + \frac{1}{2} Dv_1^2 + Dgh_1 = P_2 + \frac{1}{2} Dv_2^2 + \frac{1}{2} Dv_2^2$
 - Dgh₂



Loop wing, cuff, cups

