AP Chapter 7 Study Guide: The Cell Membrane and Transport Mechanisms (Rob Hamilton)

Teacher's Note: Chapter 7 is bread and butter biology and the concepts discussed in this chapter consistently occupy a large question set on the objective portion of the AP exam. Free response questions occurred most recently in'02, '98 and '92. You must develop your knowledge concerning the structure and function of the cell membrane because to remain alive, cells must constantly take in nutrients and rid themselves of waste. We will begin by examining cell membrane structure. Read pgs 124-127 and answer the following questions:

1. In the space provided, make a simple drawing of a phospholipid molecule. Label its polar/hydrophilic portion and its nonpolar/hydrophobic portion.



2. Now scale down the size of your phospholipids and draw a section of the cell membrane in the space below. Use brackets to identify the hydrophobic and hydrophilic portions of the membrane.

3. What laboratory evidence suggests this hypothetical arrangement of phospholipids may in fact be correct?

4. What is the name of the currently accepted model for the arrangement of molecules within the cell

membrane? ______

- 5. What is the primary function of this bilayer of molecules?
- 6. Cholesterol which is also a lipid may be found in the bilayer. Use a colored pen to add some cholesterol molecules to your bilayer drawing in the box for #2.
- 7. Describe the function of cholesterol within the membrane
- 8. The phospholipids bilayer serves as a medium for embedding protein which have many diverse functions. Read pgs 127-129 then look at the pictures and describe the function of the proteins.









9. What is the laboratory evidence that indicates that proteins are found in the bilayer?

Has it dawned on you that since DNA codes for protein, it is your genetic material that controls all of these membrane functions related to protein. <u>Wow</u>!!!!!! Many small, non-polar molecules like oxygen and carbon dioxide can simply pass between the shifting phospholipids. Large, polar molecules usually need to be transported into the cell by a protein. Read about the different types of passive transport on pages 130-134 and then answer the questions on the next page.

10.	 Suppose someone has added a teaspoon of sugar to a cup of hot water and watched the sugar dissolve. Now someone adds a cube of sugar to that same cup. (circle answer) 					
	a) Where is there more orderly arrangement of sugar molecules? a) Cup or b) Cube					
	b) Is the entropy of sugar higher in the a) Cup or b) Cube					
	c) As you watch, the entropy of the sugar cube will go a) up or b) down					
11.	Define diffusion:					
12.	2. Give a molecular level explanation as to why the diffusion of ethanol is faster in air than in water					
13.	What is the diffusion of water called?					
14.	Water generally osmoses from areas of water concentration to areas of water concentration					
Cir	cle the correct answer for questions 15 & 16					
15.	If a cell contains a lot of solute, its concentration of water would be high or low					
	This cell would be hypertonic or hypotonic to its environment?					
	This environment would be hypertonic or hypotonic to the cell?					
16.	If a cell contains very little solute, its concentration of water would be high or low					
	This cell would be hypertonic or hypotonic to its environment?					
	This environment would be hypertonic or hypotonic to the cell?					
17.	If a cell contains the same amount of solute as its environment, it is said to be					
18.	Plant, bacterial and fungal cells respond to hypotonic solutions differently than animal cells.					
	Explain why this is so.					
19.	In a hypertonic solution an animal cell will and a plant cell will					
	In an isotonic solution a plant cell will appear or limp. In a hypotonic solution, an					
	animal cell will and a plant cell will appear or firm.					
20.	How can an amoeba, lacking a cell wall, survive in fresh water?					

British botanists were quick to point out that in the case of plant, bacterial and fungal cells, water <u>does</u> <u>not</u> always move from areas of high water concentration to areas of low water concentration. They correctly indicated that plant cells, in distilled water, have equal amounts of water moving in and out, even though water is much more concentrated outside the cell. These botanists noted that in order to truly understand how water moves, you have to account for the pressure that can develop within cells with cell walls. Therefore, they constructed the concept of water potential (ψ) to quantitatively determine the movement of water into/out of cells. Now turn to pg 8-13 in your lab manual for the symbols and formulas needed to complete the required calculations below. If you have 7th grade math skills and a calculator, you <u>can</u> do this!





Symbols and Values for Beaker A

Write the symbol

Write the Value (include the unit, if any)

Pressure Potential			
Ionization Constant			
Molar Concentration			
Pressure Constant			
Temperature (in K)			
Calculations for Beake	er A		
Solute Potential	e the symbol)	(Write the form	nula)
Solute Potential	(Show the calc	rulation with the	units)
Solute Potential =	(Write the value)	he space below.	Utilize word followed by symbols.
	=		+
Now substitute the appr	opriate values for beaker A		
Ψ Beaker A =	++		
Ψ Beaker A =			

Now lets do the same for beaker B

Symbols and Values for Beaker B

	Write the symbol	Write the Value (include the unit, if any)			
Pressure Potential					
Ionization Constant					
Molar Concentration					
Pressure Constant					
Temperature (in K)					
Calculations for Beake	<u>r B</u>				
Solute Potential	=				
(Write	e the symbol) =	(Write the formula)			
Solute Potential					
	(Show the cale	culation with the units)			
Solute Potential =	(Write the value)				
Show the formula for ca	lculation of water potential in t	the space below. Utilize word followed by symbols.			
	=	+			
Now substitute the approx	opriate values for beaker B				
Ψ Beaker B =	+				
Ψ Beaker B =					
And now for the part th	at is really important!				
Which beaker has the gr	eatest water potential?				
If the two beakers were connected by a membrane permeable only to water, the net movement of water would be					
(a) Beaker A to Bea	aker B (b) Beaker B t	to Beaker A			
Now use the words " up	" or "down" to answer the foll	lowing:			
If temperature goes up, v	ψ goes	If pressure goes up, ψ goes			
If molar concentration g	oes up, ψ goes				

21.	Oxygen is a larger molecule than water. However, O ₂ diffuses more quickly across a phospholipid bilayer
	than H ₂ O. Explain how this is possible.
	Do O ₂ and H ₂ O molecules move with, or against the concentration gradient?
	Is this an active or passive transport mechanism?
22.	Molecules that are large or polar cannot diffuse though the non-polar tails of the phospholipids. Yet many large, polar molecules like tyrosine (amino acid) passively enter a cell on a regular basis. Explain how this is
	possible.
	Is this with, or against the concentration gradient?
23.	Describe what happens as sodium ions leave the cell by active transport
24.	Compare and contrast facilitated diffusion and active transport in two ways. (2 similarities and 2 differences)
25.	Describe the location and function of carbohydrates associated with the cell membrane.
26.	What is endocytosis?
	How do phagocytosis and pinocytosis differ?
27.	How is receptor mediated endocytosis more efficient than pinocytosis?
28.	What takes place during exocytosis?