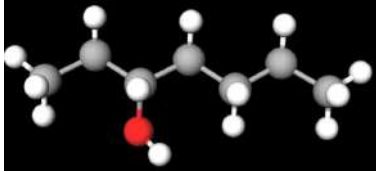
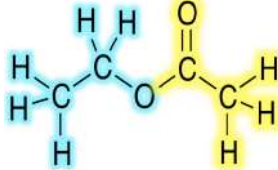
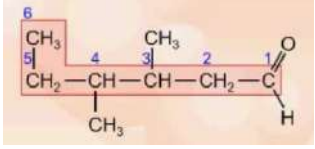
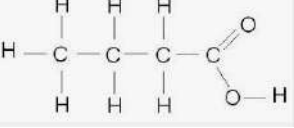
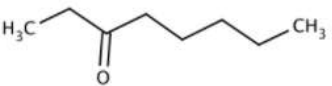
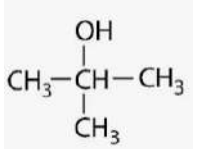
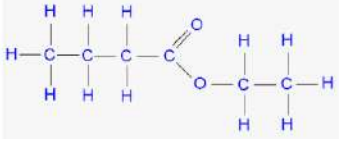
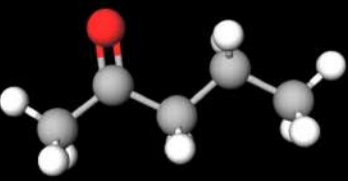
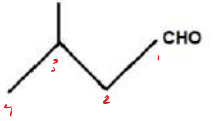
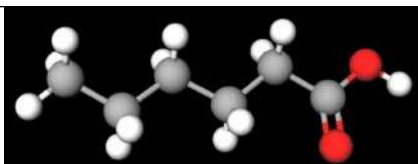


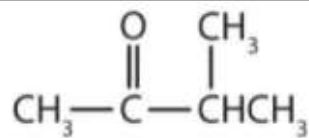
## Alcohols, Aldehydes, Carboxylic Acids, Ketones, and Esters Nomenclature Practice

Determine the compound type for each and name accordingly. There are **five each** of: **alcohols, aldehydes, ketones, carboxylic acids, and esters.**

 <p>3 - heptanol</p>	$\text{CH}_3\text{CH}_2\text{CH}_2-\overset{\text{O}}{\underset{\text{  }}{\text{C}}}-\text{CH}_2\text{CH}_2\text{CH}_3$ <p>4 - heptanone</p>	$\begin{array}{c} \text{H} \\   \\ \text{CH}_3\text{C}=\text{O} \end{array}$ <p>ethanal</p>
$\begin{array}{c} \text{CH}_3 \\   \\ \text{CH}_3-\text{C}=\text{O} \end{array}$ <p>propanone (or 2 - propanone)</p>	 <p>ethyl ethanoate</p>	$\text{CH}_3\text{CH}(\text{OH})\text{CH}_3$ <p>2 - propanol</p>
 <p>3,4 - dimethylhexanal</p>	 <p>butanoic acid</p>	 <p>3 - octanone</p>
$\text{CH}_3\text{CH}_2\text{CH}_2-\overset{\text{O}}{\underset{\text{  }}{\text{C}}}-\text{O}-\text{CH}_3$ <p>methyl butanoate</p>	 <p>2 - methyl - 2 - propanol</p>	 <p>ethyl butanoate</p>
 <p>2 - butanone</p>	 <p>3 - methylbutanal</p>	$\text{CH}_3-\overset{\text{O}}{\underset{\text{  }}{\text{C}}}-\text{OH}$ <p>ethanoic acid</p>

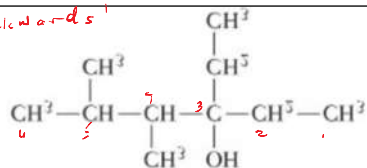


hexanoic acid

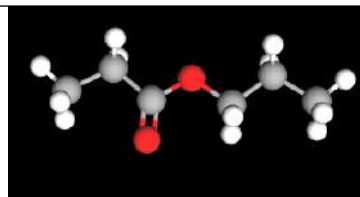


3-methyl-2-butanone

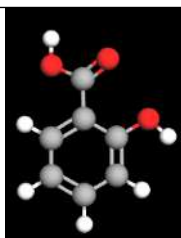
oops - backwards!



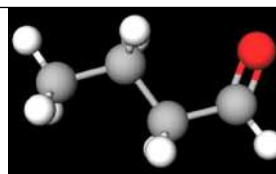
3-ethyl-4,5-dimethyl-3-hexanol



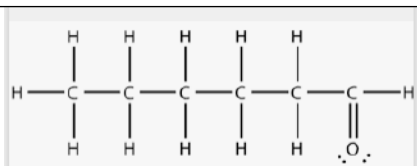
propyl propanoate



2-hydroxybenzoic acid



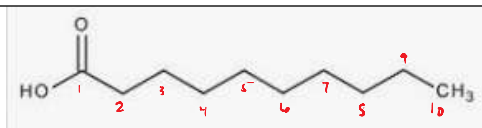
butanal



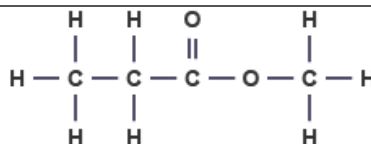
hexanal



2,3-pentanediol



decanoic acid



methyl propanoate