## New Paltz Central School District Science Physical Science

Time	Essential Questions/Content	Standards/Skills	Assessments
September – October  November - December	<ul> <li>Unit 1: Energy and Energy Transfer</li> <li>What is energy? What are its different forms?</li> <li>How is energy transformed and conserved in everyday life?</li> <li>How can we build devices that convert energy from one form to another?</li> </ul>	<ul> <li>Safely and accurately use measurement tools</li> <li>Use appropriate units for measured and calculated values</li> <li>Recognize and analyze patterns and trends</li> <li>Sequence events</li> <li>Identify cause and effect relationships</li> <li>Use indicators and interpret results</li> <li>Follow safety procedures in the classroom and laboratory</li> </ul>	<ul> <li>Energy project</li> <li>Quizzes</li> <li>Tests</li> </ul> • Quizzes <ul> <li>Lab report</li> </ul>
	<ul> <li>and gasses?</li> <li>How does heat affect matter?</li> <li>How does the motion of particles determine the phases of matter?</li> </ul>	<ul> <li>Safely use laboratory burner and associated equipment</li> <li>Identify patterns and trends</li> <li>Identify cause and effect relationships</li> </ul>	• Unit test
January	<ul> <li>Unit 3: Force, Mass, and Weight</li> <li>What are force, mass, and weight?</li> <li>What is the relationship between weight and gravity?</li> <li>How are force, mass, and weight measured?</li> <li>How are mass and weight related?</li> </ul>	<ul> <li>Accurately use a spring scale and balance</li> <li>Use appropriate units for measured and calculated values</li> <li>Collect, plot, and graph data</li> <li>Use computer technology</li> </ul>	<ul><li>Lab report and graph analysis</li><li>Quizzes</li></ul>

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February -	<ul> <li>Unit 4: Heat and Heat Transfer</li> <li>How is heat measured?</li> <li>How is heat transferred?</li> </ul> Unit 5: Elements, Atoms, and the Poriodia Table	<ul> <li>Accurately use instruments for measuring heat</li> <li>Recognize patterns and trends</li> <li>Follow precisely a multistep procedure when carrying out</li> </ul>	<ul> <li>Heat transfer project</li> <li>Heat transfer lab report</li> <li>Quizzes</li> <li>Lab report</li> </ul>
March	<ul> <li>Periodic Table</li> <li>What is an element?</li> <li>How are elements the building blocks of the universe?</li> <li>What is an atom and what is its structure?</li> <li>What are subatomic particles?</li> <li>What is the periodic table?</li> <li>How does the position of an atom in the periodic table reflect the atom's structure?</li> <li>How do atoms interact?</li> <li>Concept of an element</li> <li>Properties of elements</li> <li>Types of subatomic particles and their properties</li> <li>Atoms and atomic structure</li> <li>Electron arrangement in an atom</li> </ul>	experiments, taking measurements, or performing technical tasks.  Determine the meaning of symbols, key terms, and other domain-specific words and phrases as they are used in a specific scientific or technical context.  Integrate quantitative or technical information expressed in words in a text with a version of that information expressed visually (e.g., in a flowchart, diagram, model, graph, or table).  Write arguments focused on discipline-specific content.  Write informative/explanatory texts, including scientific procedures/experiments, or technical processes.  Gather relevant information from multiple print and digital sources, using search terms effectively; assess the credibility and accuracy of each source, and quote or paraphrase the data and conclusions of others while avoiding plagiarism and following a standard format for citation.  Use appropriate units for measured and calculated values Recognize and analyze patterns and trends Determine the identity of an unknown element using physical and chemical properties Utilize the organization of the periodic table to predict the structure of an atom Utilize the organization of the periodic table to predict the properties of an atom Classify objects according to an established scheme	Tests/quizzes

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April	<ul> <li>Unit 6: Chemistry</li> <li>How do chemical changes produce new substances?</li> <li>How do chemical reactions provide evidence for the Law of Conservation of Mass?</li> <li>How are substances characterized by how they react?</li> </ul>	<ul> <li>Follow safety procedures in the classroom and laboratory</li> <li>Safely and accurately use measurement tools</li> <li>Use appropriate units for measured and calculated values</li> <li>Sequence events</li> <li>Identify cause and effect relationships</li> <li>Use indicators and interpret results</li> <li>Collect data</li> </ul>	• Quizzes
May - June	<ul> <li>Unit 7: Electricity and Magnetism</li> <li>What is a magnetic field?</li> <li>What is electricity?</li> <li>What is an electrical circuit?</li> <li>How are electricity and magnetism related?</li> <li>How is electrical energy transformed into other forms of energy?</li> </ul>	<ul> <li>Recognize magnetic field lines</li> <li>Recognize patterns and trends</li> <li>Diagram and build basic electrical circuits</li> <li>Collect data</li> <li>Apply learned concepts to new situations and tasks</li> </ul>	<ul> <li>Lab report</li> <li>Quizzes</li> <li>Electromagnetic invention project report</li> </ul>