

Monday 01/08

Academic Standard # 3.4.10.A & B 3.4.12.A & B

Rigor/Relevance Quadrant **A** **B** **C** **D**

Lesson Objective:

- **Explain** the significance of a chemical formula.
- **Determine** the formula of an ionic compound formed between two given ions.
- **Name** an ionic compound given its formula.
- **Write** the formula of a binary molecular compound given its name.

Details:-return and review tests

-**Lab:** Ionic Puzzle Pieces review in class together

-Naming Binary Salts notes review and review Complete WS 1 in packet      **Homework:** complete WS #1 in nomenclature HW packet omit ion pairs

Tuesday 01/09

Academic Standard # 3.4.10.A & B 3.4.12.A & B

Rigor/Relevance Quadrant **A** **B** **C** **D**

Lesson Objective: see Monday's objectives

Details: -check and review HW

-Notes Stock System of Nomenclature (Notes on pp. 224 Stock naming system of nomenclature & look green sheet)

-complete WS in packet on Transition Metals & acids

-**Lab:** Billions of bottles      **Homework:** complete WS #2 & 3 in nomenclature HW packet omit ion pairs

Wednesday 01/10

Academic Standard # 3.4.10.A & B 3.4.12.A & B

Rigor/Relevance Quadrant **A** **B** **C** **D**

Lesson Objective: see Monday's objectives and naming molecular and hydrate compounds

Details: -check and review HW

-Notes on molecular compounds and naming hydrates and all ionic compounds

-complete molecular compounds WS      **Homework:** -complete Molecular compounds in Nomenclature HW Packet

Thursday 01/11

Academic Standard # 3.4.10.A & B 3.4.12.A & B

Rigor/Relevance Quadrant **A** **B** **C** **D**

Lesson Objective: see Monday's objectives and how to name acids

Details: -check and review HW

-Quiz on nomenclature

-Notes on naming acids      **Homework:** -complete naming acid in Nomenclature HW Packet

Friday 01/12

Academic Standard # 3.4.10.A & B 3.4.12.A & B

Rigor/Relevance Quadrant **A** **B** **C** **D**

Lesson Objective: see Monday's objectives and

- **List** the rules for assigning oxidation numbers.
- **Give** the oxidation number for each element in the formula of a chemical compound
- **Name** binary molecular compounds using oxidation numbers and the Stock system.

Details:-Check and review naming acids WS

-Quiz on nomenclature all compound types

-Notes on 7.2 Oxidation Numbers pp. 232-234

**Homework:** complete oxidation numbers worksheet (Matter)