Unit 1 Review Sheet Honors Biology

Name

Content Domain 1: Scientific Processes and Nature of Biology		
1.	List the steps of the scientific method, in order. Write a brief description of each step.	
	a	
	b	
	C	
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2.	In a fish respiration experiment (put goldfish in different temperatures of water and	
	count the respirations), what kind of data would be collected; quantitative or	
	qualitative? How do you know?	
3.	What is the independent variable? How do you	
	know?	
4.	What is the dependent variable? How do you know?	
_		
5.	What are some controlled variables (constants)?	
6.	what is the difference between a conclusion and an inference?	
7.	Describe when you would use the following types of graphs: a. Line graph	
	c. Pie graph-	
8.	List some common tools used by scientists to make observations.	
9.	Label the parts of the microscope.	
	9.	
	1.	
	2.	
	3.	
	4.	
	11.	
	3.	
	6.	
	7. 13.	
	8	
	What is the function of these?	
	a. Objectives	
	b. Coarse wheel adjustment-	
	c. Fine wheel adjustment-	

- d. Diaphragm-____

11.	What is the metric unit for the following?
	a. length b. mass c. volume
12.	Common Safety Rules: When should goggles be worn?
Wł	nen should you wash your hands?
13.	What is the subject of each of the following branches of biology?
	a. Botany
	b. Ecology
	c. Genetics study of
	d. Microbiology study of
	e. Taxonomy study of
	f. Zoology study of
	Content Domain 2: Cellular Basis of Life
1.	List the 3 parts of the cell theory: a. cell is the basic unit of; b. all
	organisms are composed of; c. all cells come from
2.	List the characteristics of living things:
	a. Made of
	b. Require food for to carry out life processes
	c. Respond to in their environment
	d offspring
	e and
	f. Maintain; a balance of internal conditions
	g. Have and pass on a code
3.	A organism has to perform all metabolic activities within
	one cell.
4.	A organism has specialized/ cells that
	perform certain jobs.
5.	In multicellular organisms, a group of related cells makes up , which
	makes up , which are grouped into , which
	work together to make up the organism.
6.	A cell does NOT contain a nucleus or membrane-bound
	organelles.
7.	A cell does have a nucleus and other organelles.
8	What is cell specialization?
0.	
	Does it occur in unicellular organisms only multicellular organisms only or both
	kinds?

Match the organelle with its function.

- a. Boundary that surrounds the nucleus
- b. Found in plants for support & protection
- c. Gel-like material where organelles are found
- d. Makes proteins

- 12. e. Gatekeeper; controls what comes in & out of cell
 - f. Control center of cell; where DNA is found
 - g. Powerhouse; where energy is made from food

- h. Stores substances
- i. Packages proteins
- j. Site of photosynthesis in plant cells
- k. Distributes proteins like a roadway
- 1. Structure inside nucleus that contains genetic code
- 14. ___plasma (cell) membrane
- 15. _____chloroplast
- 16. ____nucleus
- 17. ____nuclear membrane
- 18. _____cytoplasm
- 19. _____cell wall
- 20. ____Golgi body
- 21. ____endoplasmic reticulum
- 22. ____ribosome
- 23. _____chromatin/chromosome
- 24. ____vacuole
- 25. __mitochondria





Content Domain 3: Chemical Basis of Life

- 1. is anything that takes up space and has mass. The 3 states are
- 2. The ______ is the building block of all matter. There are 3 particles. Inside the nucleus are the _____, which have a _____ charge, and the _____, which have no charge. Surrounding the nucleus are the ______charged
- 3. Electrons found in the outermost shell are called _______ electrons. When this shell is full, the atom is ______ and does not bond with others. There can only be 2 electrons in the first shell, 8 in the second shell, and 8 or 18 in the third shell.
- 4. The atomic number tells you the number of _____, which equals the number of ______ in a neutral atom. The atomic mass minus atomic number tells you the number of ______.5. Draw the atomic structure for boron.
- 6. An is a substance made up of all the same kind of atoms. The most common ones found in living things are carbon, hydrogen, oxygen, nitrogen, sulfur, and phosphorus.
- 7. There are 3 types of chemical bonds: ionic, covalent, and hydrogen.
 - a. **ionic bond**: electrons are _____, resulting in charged atoms that are held together because opposites attract. Ex- NaCl (sodium chloride)
 - b. covalent bond: electrons are _____. Ex- CO₂
 - c. **hydrogen bond**: weak attractions between molecules. Exholds water molecules to other water molecules; holds nitrogenous bases together in a DNA molecule
- 8. A chemical ______ is when chemical bonds are broken and new bonds form.
- 9. An _____ compound always contains carbon and comes from living things.
- 10. An ______ compound does not contain carbon and comes from nonliving things.
- 11. The most important inorganic compound for living things is . It is polar (the hydrogen side is positive while the oxygen side is negative). It is a great (it can dissolve most things), necessary for most chemical reactions in living things.
- 12. The ______ scale, from 0 to 14, is a measure of how acidic or basic something is. Neutral is _____. Acids are ______ than 7. Acids release hydrogen ions (H⁺). Bases are _____ than 7. Bases release hydroxide ions (OH⁻). Stronger acids and bases are from 7, while weaker acids and bases are to 7.
- 13. The four basic types of organic compounds found in living things are _____, _____, _____, and _____

- 14. <u>Carbohydrates</u> contain these 3 elements: ____, ___, and ____. Carbs provide a major source of _____. Monosaccharide example: _____polysacchararide example:
- and

 15. Three types of lipids are

 blocks of lipids are 1

 and 3

 skin (& plant leaves); cover neurons to

- current; found in phospholipid ______, ___, ___, and sometimes ____. The building blocks are _____. They are used for growth and repair. A special type of protein that ______a chemical reaction is called an _____
- 17. Nucleic acids contain an organism's _____. The building blocks are _____. Ex: _____& _____