Naming and Formula Writing

b. BaI₂

Second Semester Study Guide (Part 1)	Name
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Reviewing concepts for the final- Naming compounds writing formulas Naming and formula writing- (originally from a Sem 1 final review) 27. Which is the correct name for the compound N_2O_3 ? a. dinitro trioxide trinitrogen dioxide dinitrogen trioxide b. nitrogen (II) oxide (III) 28. Which is the correct name for the compound PCl₅? a. monophosphorus tetrachloride phosphorus pentachloride b. phosphorus chloride pentaphosphorus chloride 29. Which is the correct name for the compound N₂O₄? a. tetranitrogen dioxide dinitrogen quadoxide b. dinitrogen tetroxide nitrogen oxide 30. Which is the correct formula for the compound dinitrogen monoxide? a. NO_2 N_2O_1 b. $(NO)_2$ d. N₂O 31. Which is the formula for the compound sulfur hexachloride? a. S₆C1 c. SCl₅ d. S₂Cl₃ 32. Which is the correct name for the compound HClO₃? a. hydrogen chlorate hydrogen chloride trioxide b. chloric acid d. hydrochloric acid 33. Which is the correct name for the compound HNO₂? a. hyponitrous acid nitrous acid b. hydronitrogen dioxide hydronitric acid d. 34. Which is the correct formula for the compound phosphoric acid? a. H₃P c. H₃PO₄ b. HPO₃ d. H₃PO₃ 35. The name of an oxyacid has the suffix -ic acid. What is the suffix of the oxyanion? c. ate a. ic d. ous 36. Which of the following is NOT a cation? a. iron(III) ion Ca²⁺ d. mercurous ion b. sulfate 37. An -ate or -ite at the end of a compound name usually indicates that the compound contains ... a. fewer electrons than protons only two elements b. neutral molecules d. a polyatomic anion 38. Which of the following compounds contains the Mn³⁺ ion? a. MnS c. Mn₂O₃ b. MnBr₂ d. MnO 39. How are chemical formulas of binary ionic compounds generally written? a. cation on left, anion on right b. anion on left, cation on right c. Roman numeral first, then anion, then cation d. subscripts first, then ions Which of the following formulas represents an ionic compound? c. N_2O_4 a. CS₂

d. PCl2

	41.	Which set of chemical name and chemical for a. iron(II) oxide, Fe ₂ O ₃	ormula c.	
		b. aluminum fluorate, AlF ₃		potassium chloride, K ₂ Cl ₂
	42.	Which set of chemical name and chemical for a. ammonium sulfite, (NH ₄) ₂ S		for the same compound is correct?
		b. iron(III) phosphate, FePO ₄	d.	
	43.	What is the ending for the names of all binar		
·	τЭ.	aide	c.	
		bite	d.	
	44.	Which of the following shows both the corre a. HClO ₂ , chloric acid	ect form	
		b. HNO ₂ , hydronitrous acid	d.	HI, iodic acid
	45.	What is the formula for sulfurous acid?		
		a. H ₂ SO ₄	c.	H_2SO_2
		b. H ₂ SO ₃	d.	H_2S
	46.	What is the formula for phosphoric acid?		
		a. H ₂ PO ₃	c.	HPO_2
		b. H ₃ PO ₄	d.	HPO₄
	47.	What is the formula for hydrosulfuric acid?		
		a. H ₂ S ₂	c.	HSO ₂
		b. H ₂ SO ₂	d.	H_2S
	48.	What is the correct name for the compound (CoCl ₂	?
		a. cobalt(I) chlorate	c.	cobalt(II) chlorate
		b. cobalt(I) chloride	d.	cobalt(II) chloride
	49.	What is the correct formula for barium chlor	ate?	
		a. Ba(ClO) ₂	c.	\ J/A
		b. Ba(ClO ₂) ₂	d.	$BaCl_2$
	50.	What does an -ite or -ate ending in a polyato		
		a. Oxygen is in the formula.		Nitrogen is in the formula.
	<i>-</i> 1	b. Sulfur is in the formula.	d.	
	51.	Which of the following is the correct name f	_	_
		a. nitrous oxide	c. d.	nitrogen dioxide nitrate oxide
	52	b. dinitrogen pentoxide What is the correct name for Sn. (BO.) 2	a.	mtrate oxide
	52.	What is the correct name for $Sn_3(PO_4)_2$?		tin(III) ah saabata
		a. tritin diphosphateb. tin(II) phosphate	c. d.	tin(III) phosphate tin(IV) phosphate
		(1-) Proobrance	۵.	(- ·) Prosprane
Showx	our	work for the formulas then write them and id	lantify	if they are ionic or covalent

Show your work for the formulas, then write them and identify if they are ionic or covalent.

- 99. Write the formula for the compound barium oxide.
- 100. Write the formula for the compound rubidium phosphide.
- 101. Write the formula for the compound boron chloride.

103.	Why is it necessary to use prefixes in naming covalent compounds?