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## **Across**

- **1.** the study of conversions between thermal energy and other forms of energy
- **4.** the transfer of thermal energy with no overall transfer of matter
- **7.** a temperature of 0 kelvin; the temperature at which the motion of the particles theoretically stops
- **9.** the amount of heat needed to raise the temperature of one gram of a material by one degree Celsius
- **11.** a decrease in the volume of a material due to a temperature decrease
- **12.** the transfer of energy by waves moving through space
- **13.** a material that allows thermal energy or electricity to flow through easily
- **16.** an increase in the volume of a material due to a temperature increase
- **17.** a material that allows thermal energy or electricity to flow through poorly

## Down

- 2. occurs when a fluid circulates in a loop as it alternately heats up and cools down
- **3.** the transfer of thermal energy when particles of a fluid move from one place to another
- **5.** a measure of how hot or cold an object is compared to a reference point; a measure of the average kinetic energy of an object
- **6.** an instrument used to measure changes in thermal energy
- **8.** The \_\_\_\_\_ law of thermodynamics states that thermal energy can flow from colder to hotter objects only if work is done
- **10.** the transfer of thermal energy from one object to another because of a temperature difference
- **14.** The \_\_\_\_\_ law of thermodynamics states that absolute zero cannot be reached
- **15.** The \_\_\_\_\_ law of thermodynamics states that energy is conserved