



# ELECTRICITY

# Goals

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Explain the flow of electrons in terms of

- alternating and direct current.
- the relationship among voltage, resistance and current.
- simple series and parallel circuits.

Explain static electricity in terms of

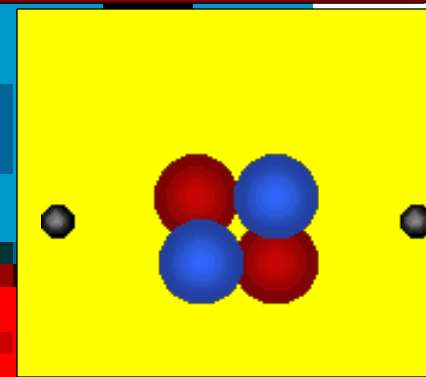
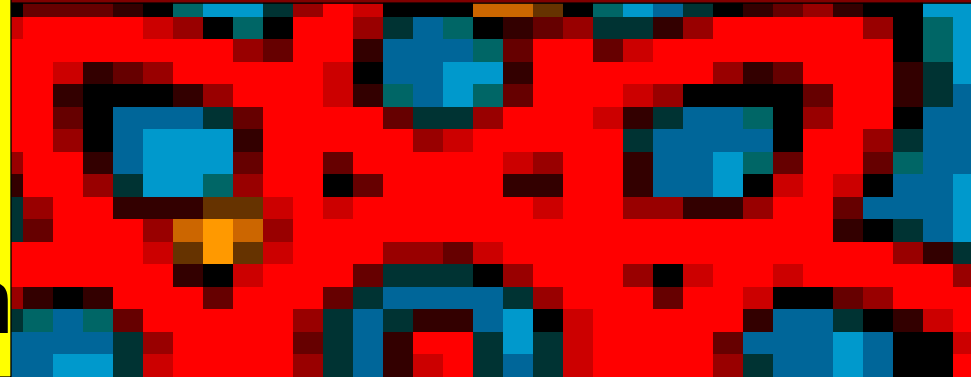
- friction
- induction
- conduction

Construct simple circuits

# Properties of Electric Charge

Atomic Structure: Composed of three main particles:

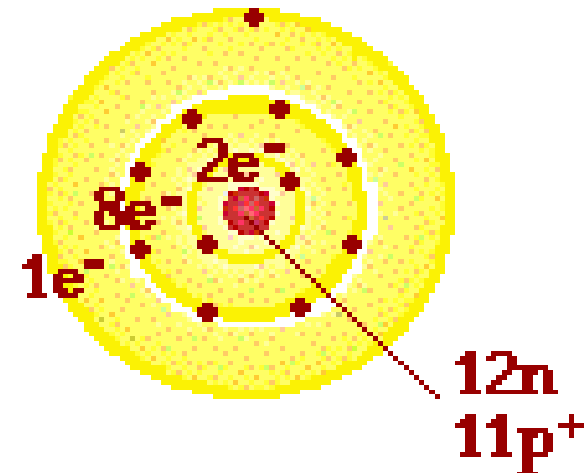
1. Proton
2. Neutron
3. Electron



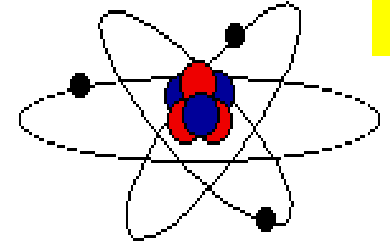
Oxygen Atom

Neon Atom

Sodium Atom



Atoms are composed of protons and neutrons located in the nucleus and electrons which are positioned in the surrounding regions of space known as electron shells.



## Things to Remember:

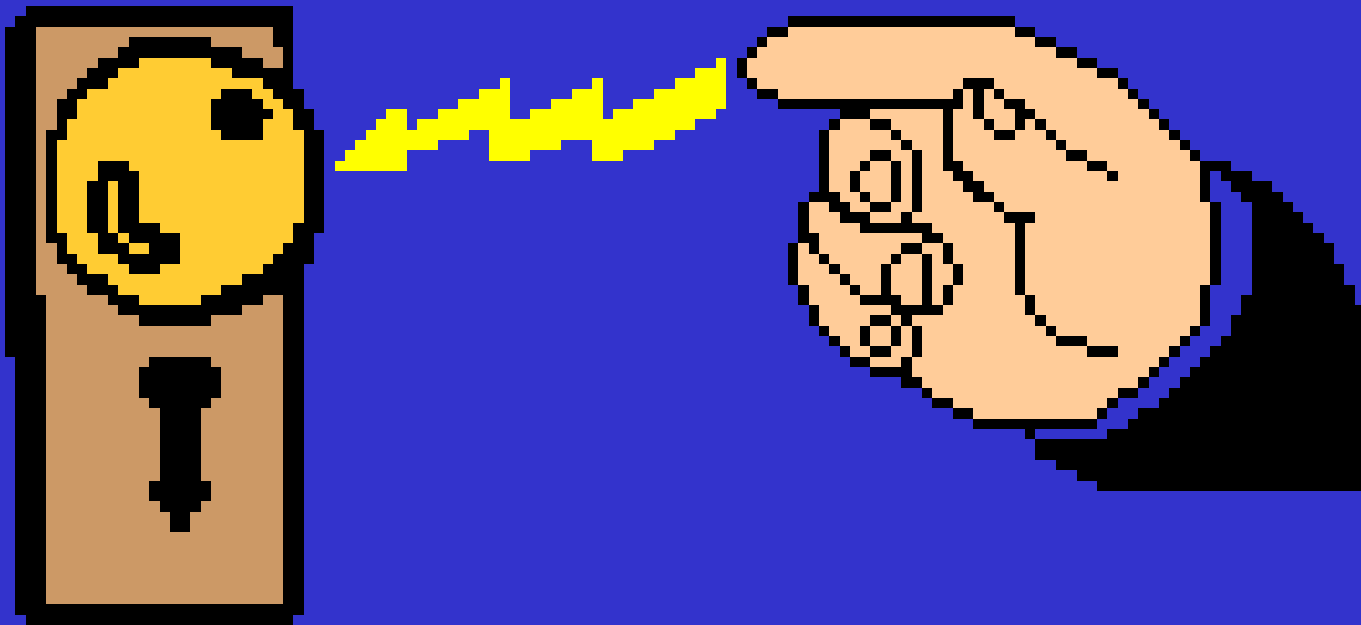
- **Everything is made of atoms.**
- **Electrons can move from one atom to another atom.**

# STATIC ELECTRICITY 5

- The temporary building up of charge on an object.
- Some atoms hold e- more tightly than others.
- Ex. Your shoes and Carpet



If you walk across a carpet, electrons move from the rug to you (because of friction). Now you have extra electrons. Touch a door knob and ZAP! The electrons move from you to the knob. You get a shock.



**STATIC ELECTRICITY IS  
ACTUALLY AN *imbalance*  
IN THE AMOUNTS OF  
POSITIVE AND NEGATIVE  
CHARGES IN THE  
SURFACE OF AN OBJECT**



American physicist  
Robert Jemison  
Van de Graaff  
invented the Van  
de Graaff  
generator in 1931



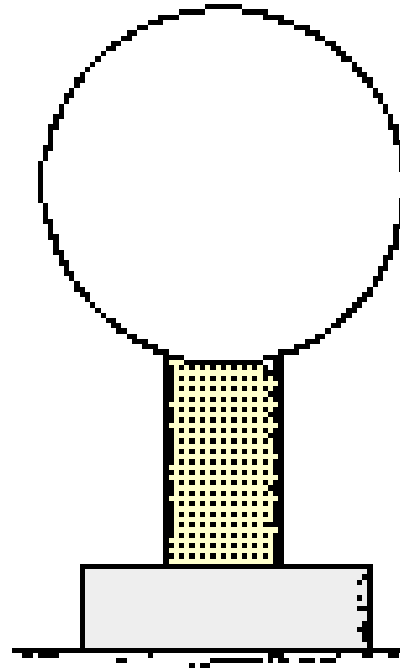
**Van de Graaff  
Generator Demo**



# Three ways to induce a charge in an object

- 1. Friction**
- 2. Induction**
- 3. Conduction (Contact)**

Involves the charging of one object by another without direct contact.



**A negatively charged object is brought near to a neutral, conducting sphere. Electrons in the sphere are forced from the left side of the sphere to the right side.**

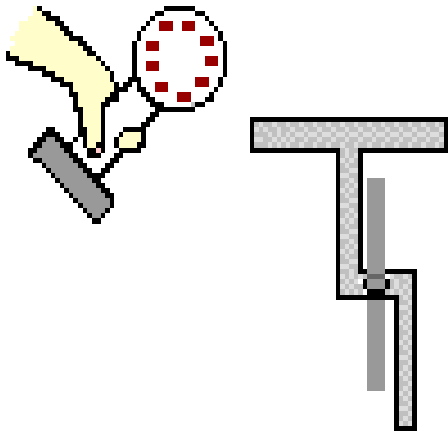
# Charging by Conduction

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involves the direct contact of a charged object to a neutral object.

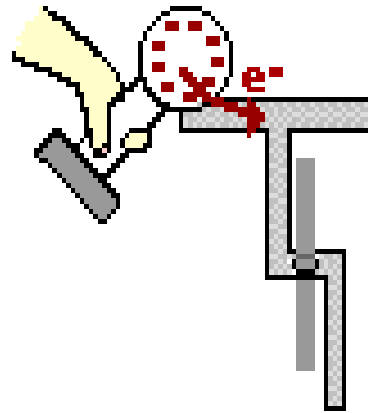
## Charging a Neutral Object by Conduction

Diagram i.



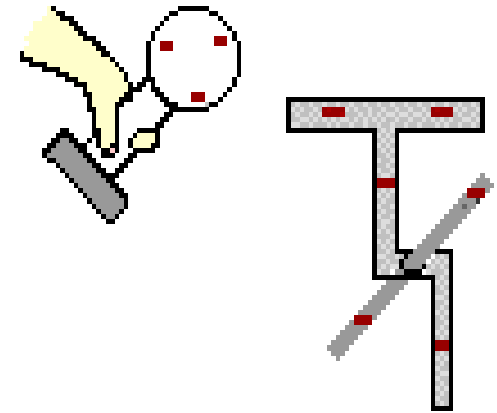
A metal sphere with an excess of - charge is brought near to a neutral electroscope.

Diagram ii.



Upon contact,  $e^-$  move from the sphere to the electroscope and spread about uniformly.

Diagram iii.



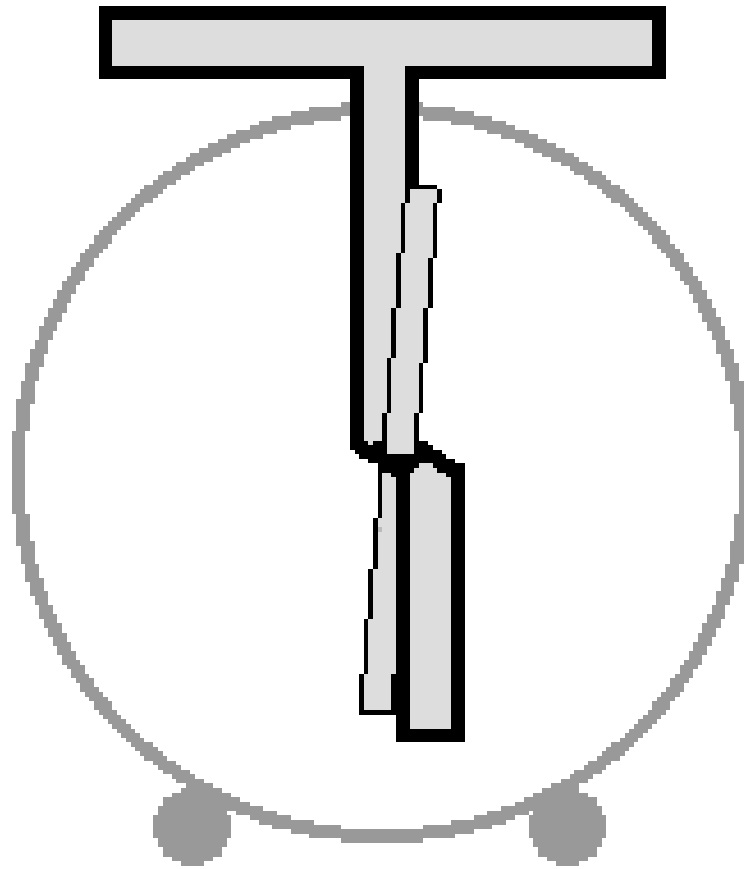
The metal sphere now has less excess - charge and the electroscope now has a - charge.

# Detecting an Electric Charge



An  
uncharged  
Electroscope

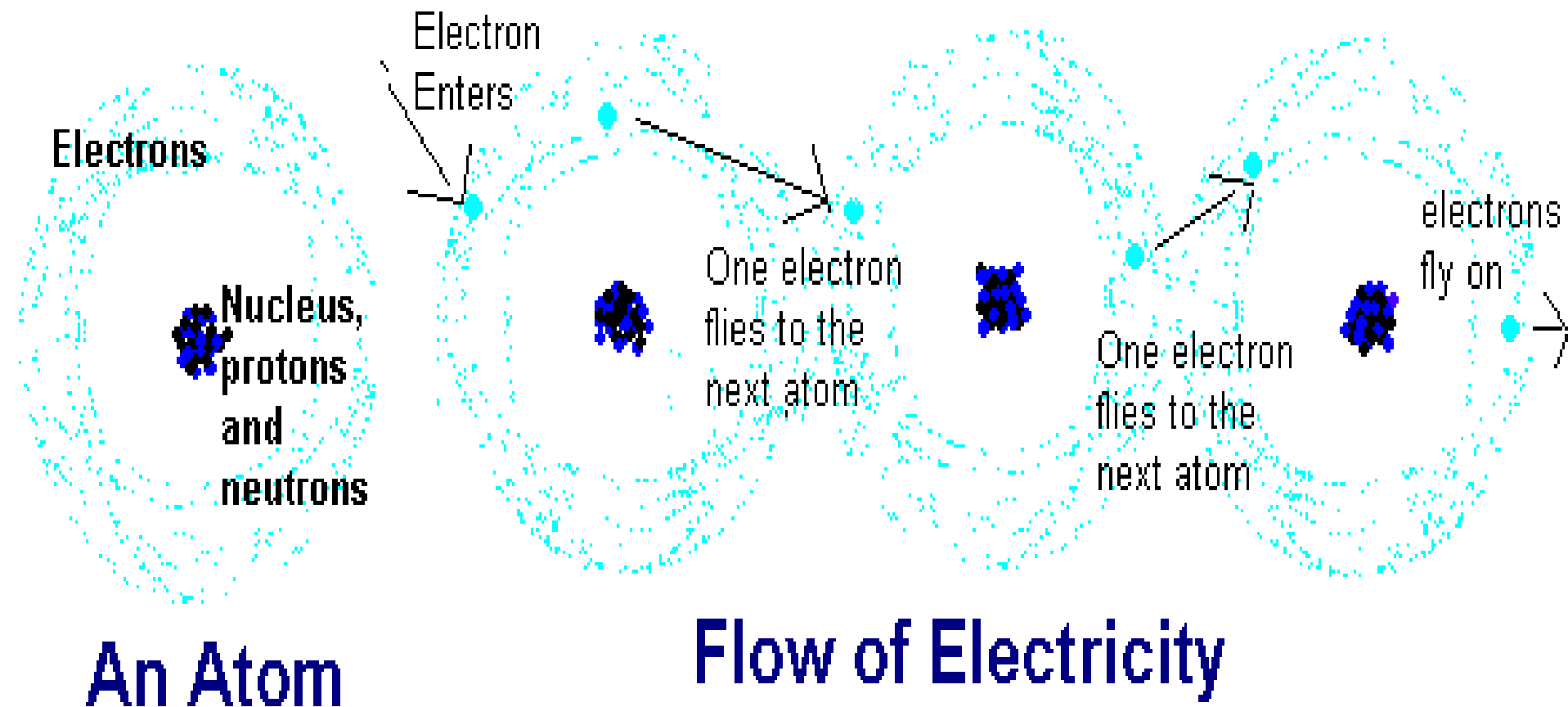
A charged electroscope.



**The electroscope is neutral as evidenced by the needle in a relaxed position.**

# *Electric Current*

**The constant flow of electrons.**



# Transfer of Electric Charge

**Has to do with the  
molecular structure of  
the material**



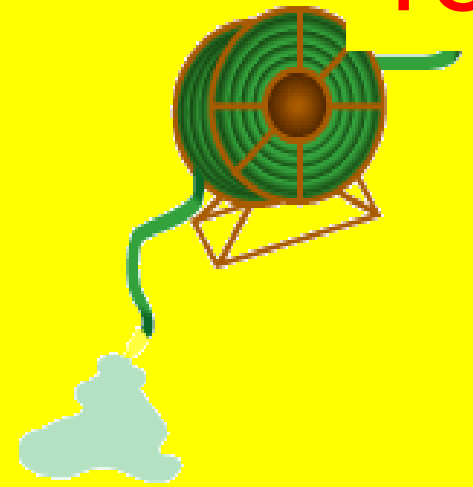
Conductors : Gold, Silver, Copper,  
Iron, Lead, Salt Water.

Insulators : Plastics, Glass, Dry Air,  
Wood.

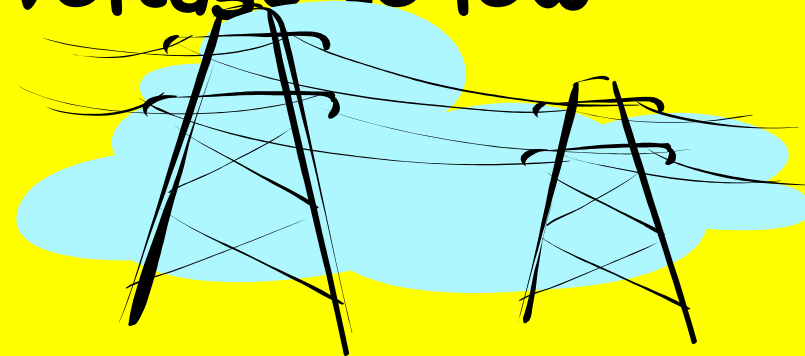
# Voltage

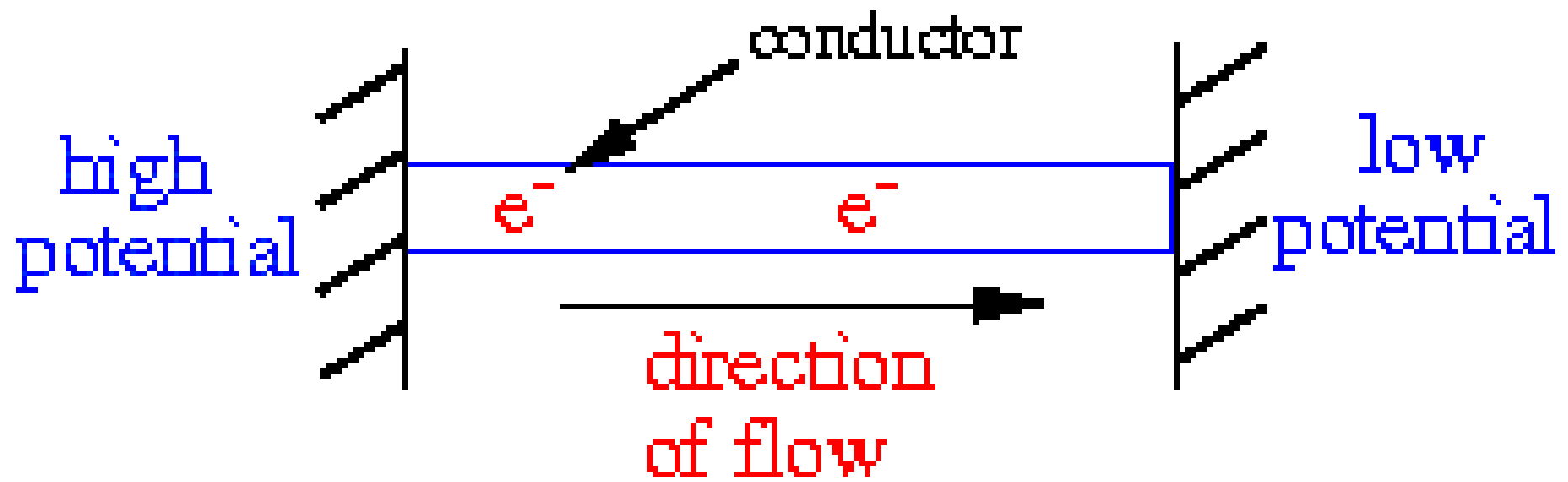
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For electrons to flow there must be a **potential difference** between to places.



- This is called voltage, which is the “Push” that causes electrons to flow.
- It’s electrical “Pressure”.
- Charges flow from high voltage to low voltage.
- Measured in Volts (V).

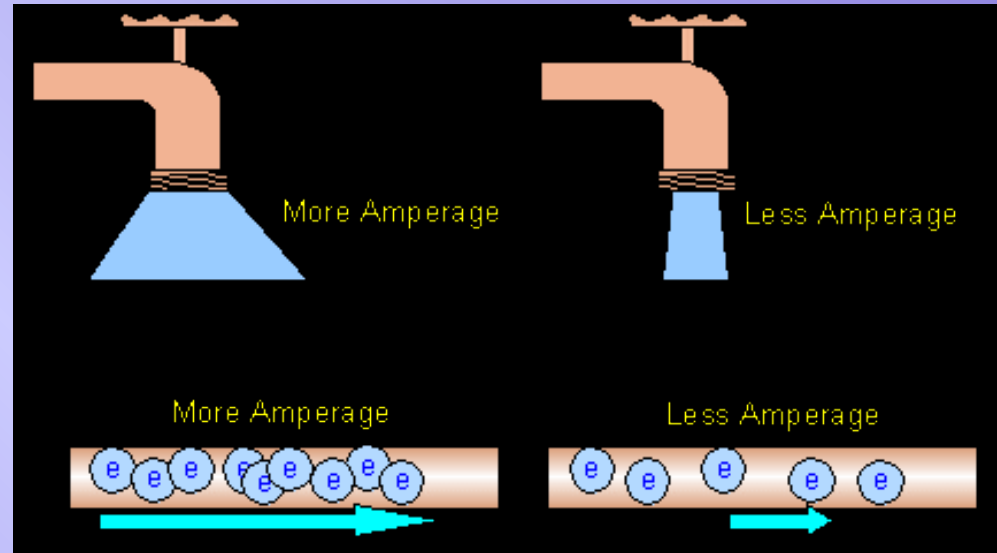




# CURRENT

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The measure of how many electrons per second are flowing through the wire is the amperage (A).



# *Electrical current*

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The # of  $e^-$  is called current (unit = Ampere or Amp. )

Electrical current is like the amount or volume of water flowing through the hose.

Water in a Hose	DC in a Wire	Electrical Units
pressure	potential (V)	Volts
volume	current (I)	Amps
friction	resistance (R)	Ohms



## Alternating Current (ac)

$e^-$                        $e^-$                        $e^-$                        $e^-$

## Direct Current (dc)

$e^-$                        $e^-$                        $e^-$                        $e^-$

# Resistance

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- The tendency for a material to **oppose** the flow of electrons.
- Different material have different amounts of resistance to the flow of electrons.
- The unit of resistance is **ohm**.

# Resistance

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**Ex: gold, silver, and copper have low resistance, which means that current can flow easily through these materials.**

**Glass, plastics, and wood have very high resistance, which means that current cannot pass through these materials easily.**



# Resistance in Wires

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- **Thick wire Vs. Thin Wire**

**Thin wires provide more resistance than do thick wires**



**Resistance also depends on temperature, usually increasing as the temperature increases resistance**

**Resistance in wires produces a loss of energy (usually in the form of heat), so materials with no resistance produce no energy loss when currents pass through them.**

# OHM'S LAW

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In a material, the current (I) is directly proportional to the voltage (V) and inversely proportional to the resistance.

$$I = \frac{V}{R}$$

OR

$$V = IR$$

# *Electric Circuits*

**A PATHWAY FOR  
ELECTRONS TO  
FLOW.**

# Four Parts of the Circuit

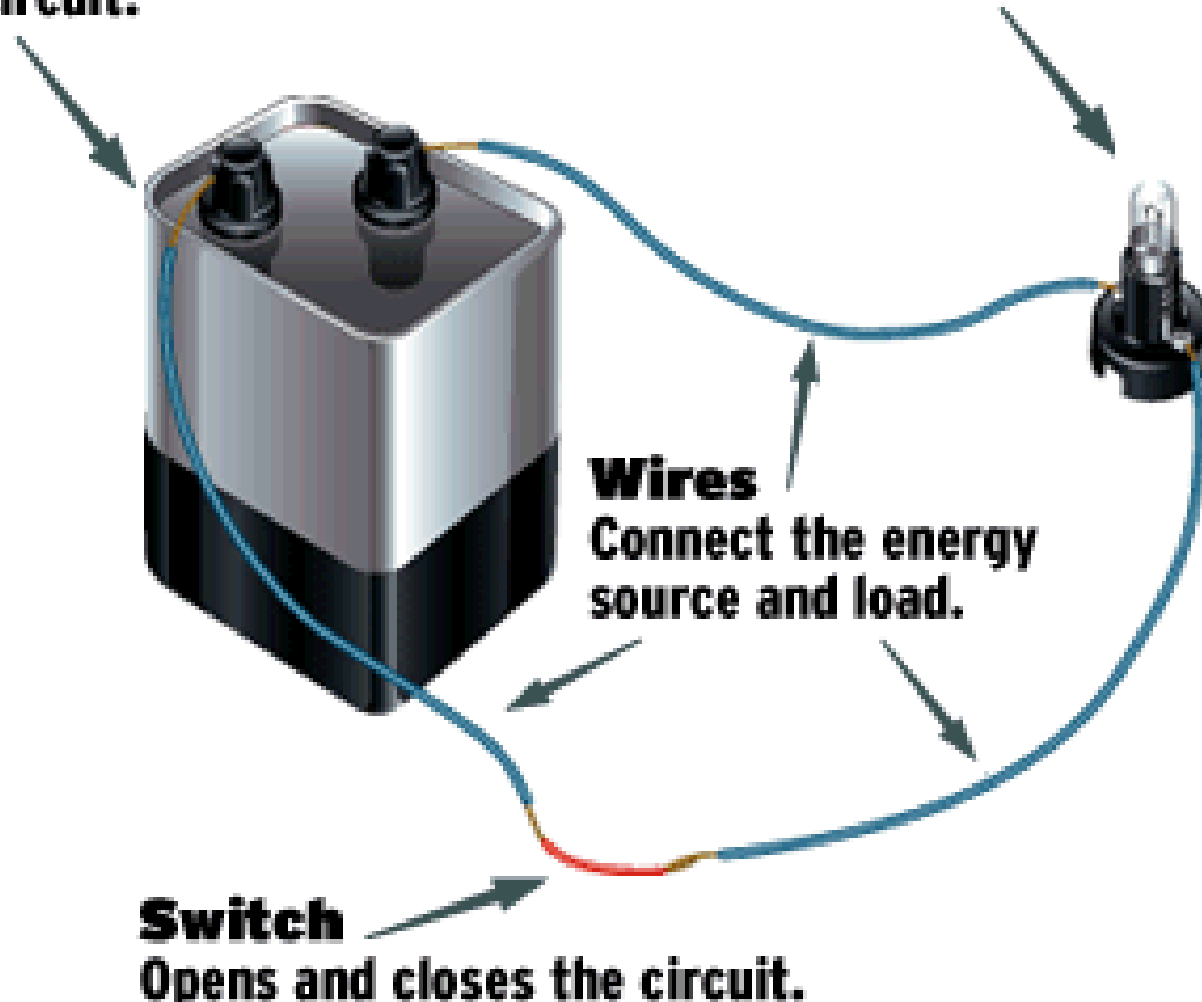
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## Energy Source

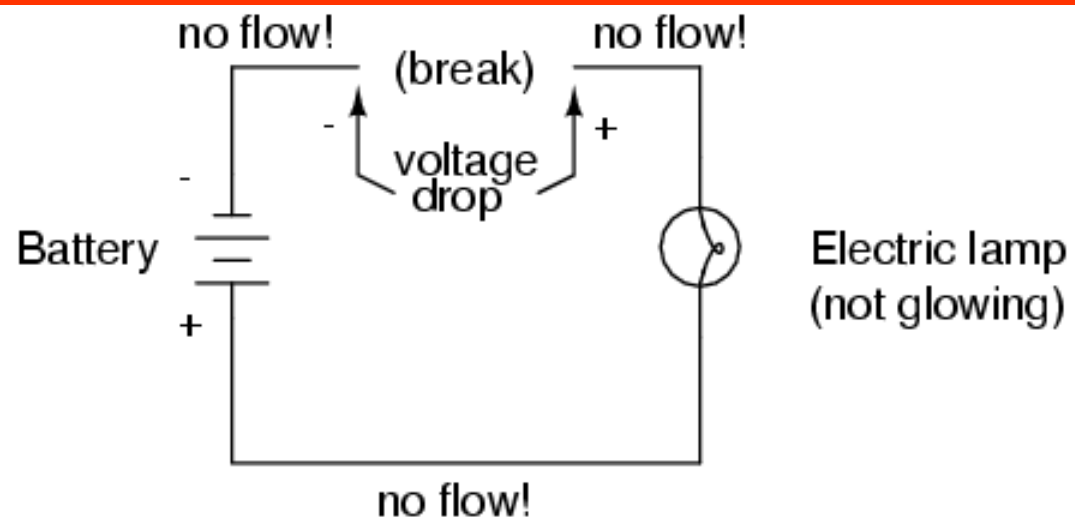
Provides the "push" that makes current move around a circuit.

## Load

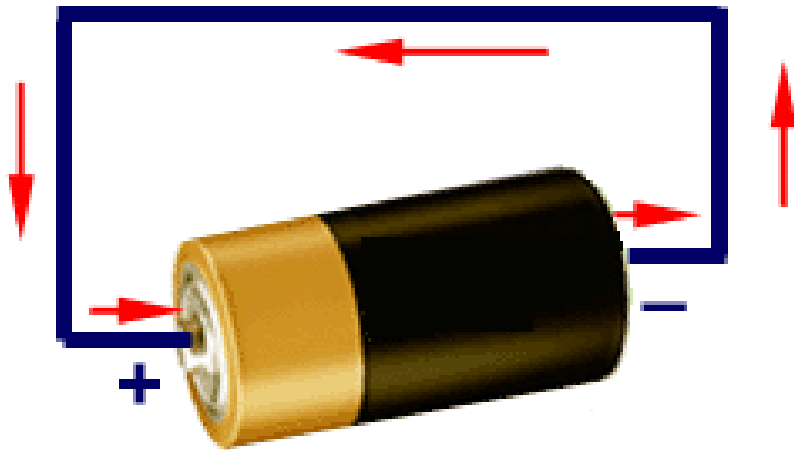
Converts electrical energy to another form (in this case, light and heat).



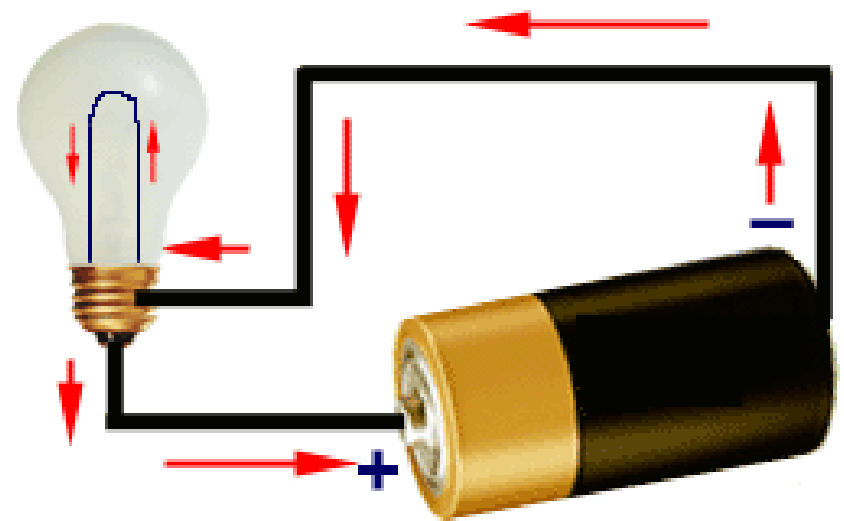
# Open Circuits



# Closed Circuits



**A simple circuit**

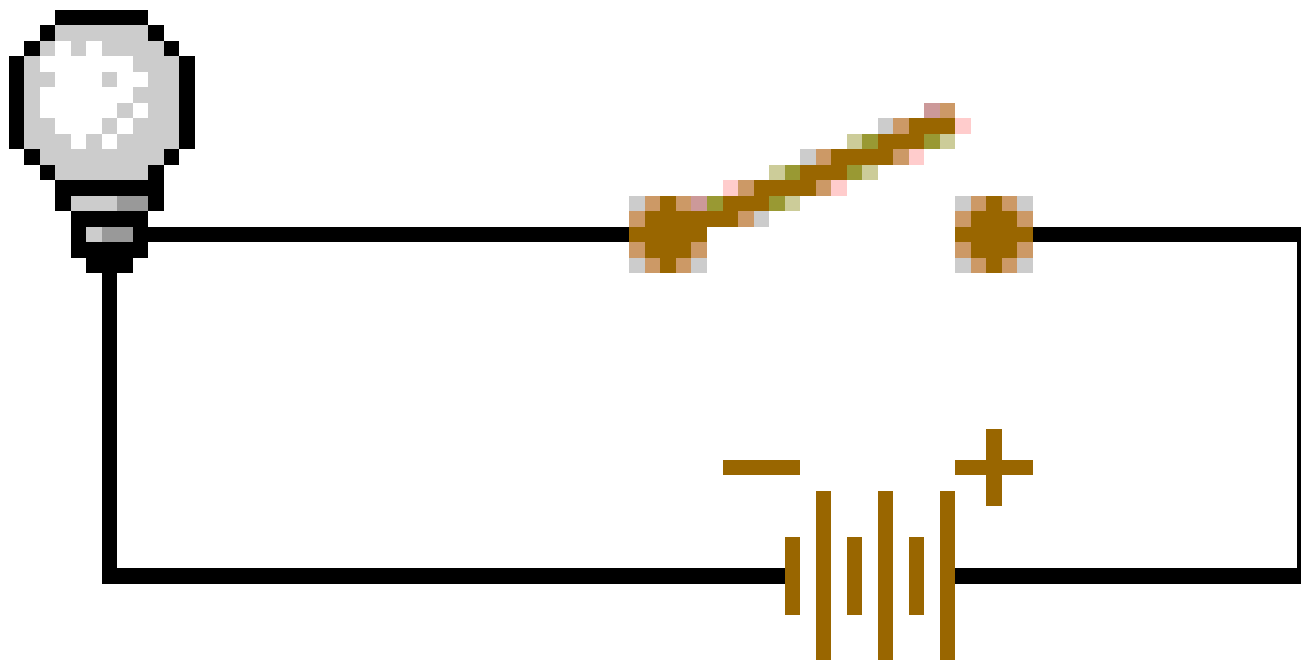


**Simple circuit with light**

# Series Circuit

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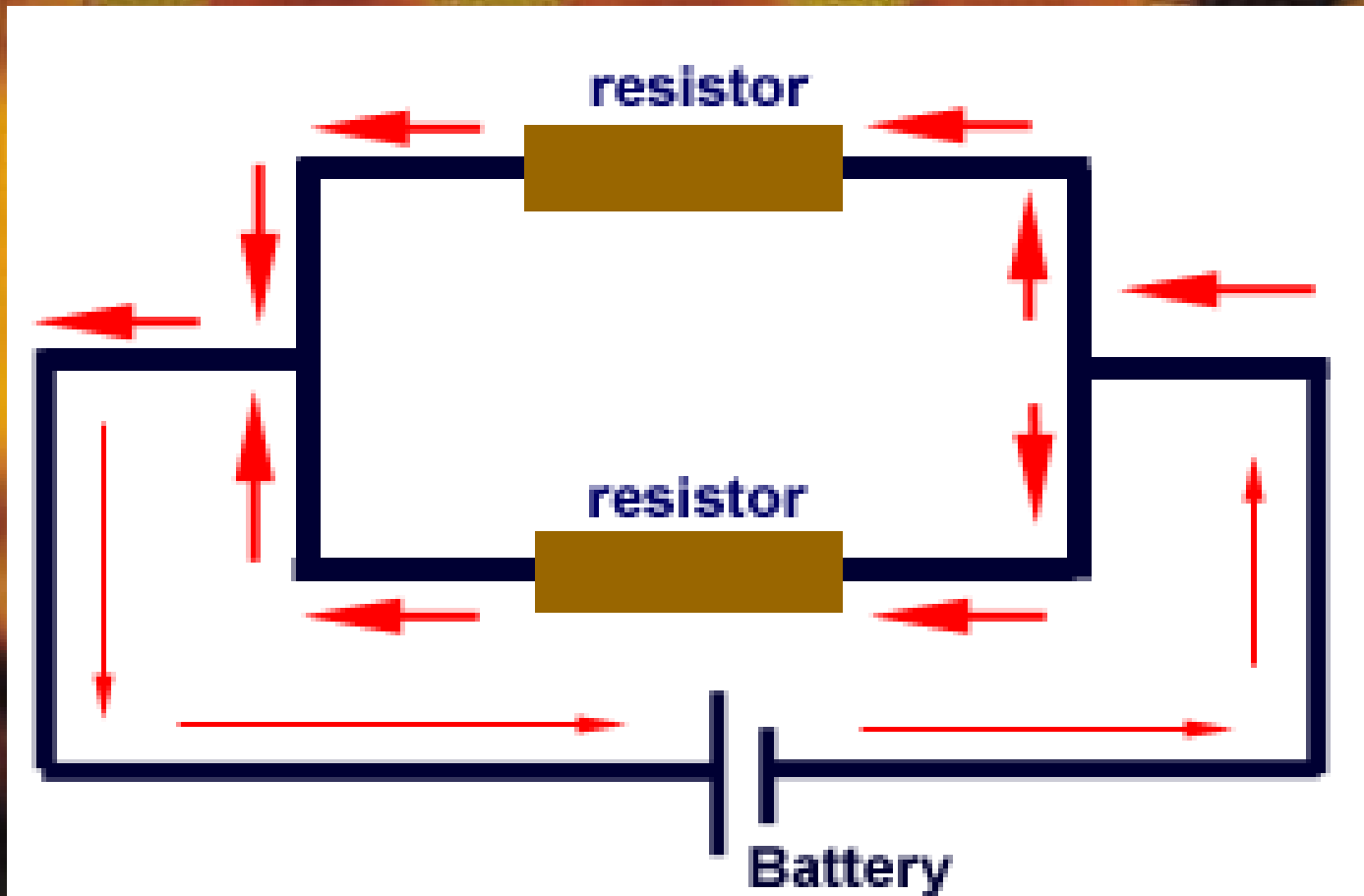
*The current has only one path to follow.*



# Parallel Circuits

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Two or more branches for the current to flow.

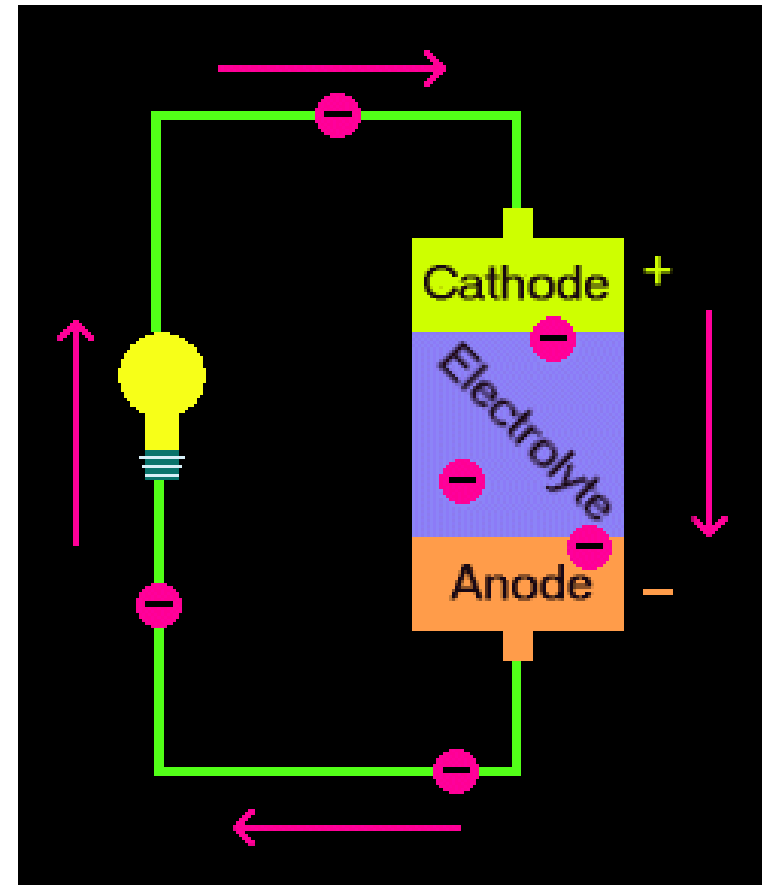




# How do batteries work?

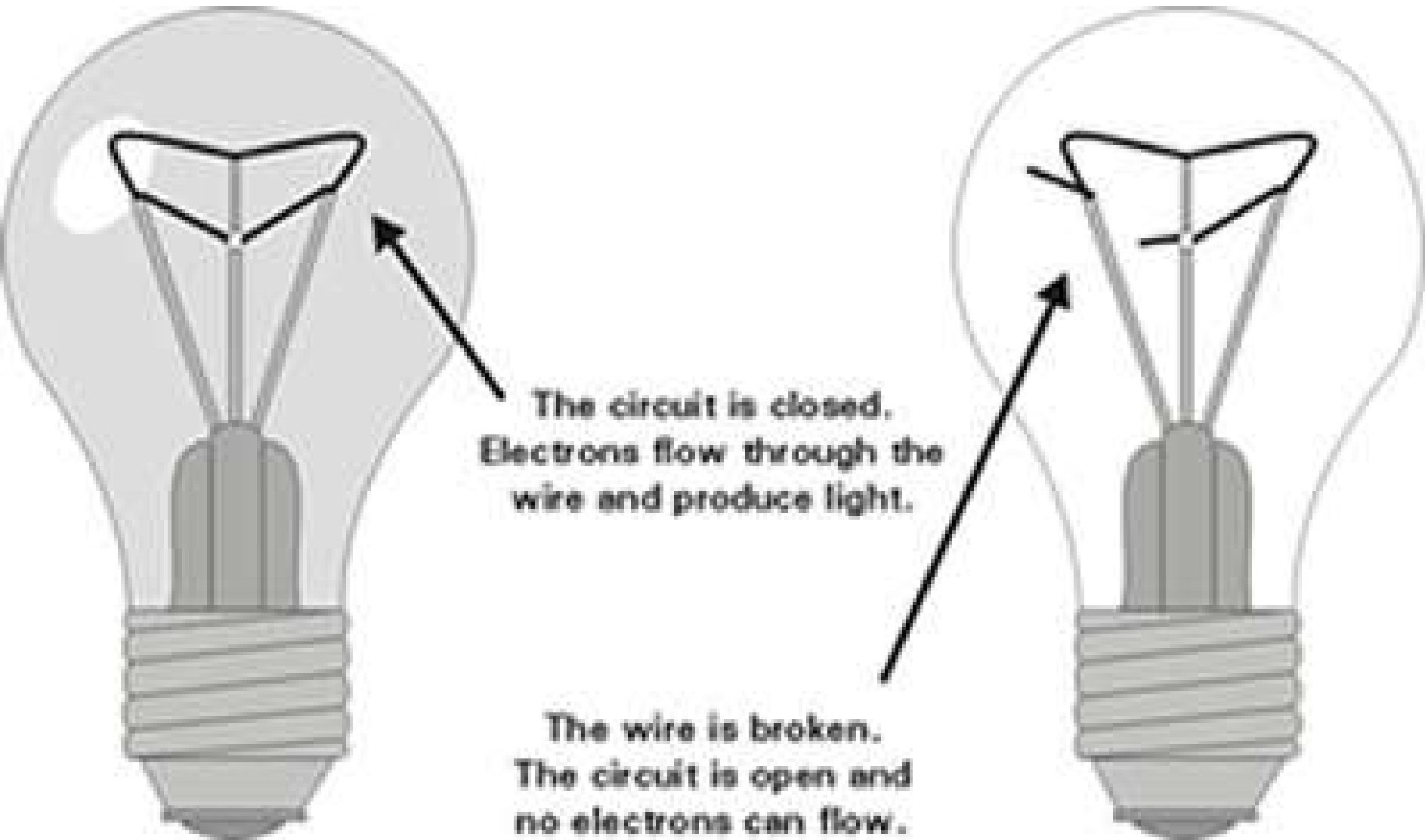
Batteries have three parts, an *anode* (-), a *cathode* (+), and the *electrolyte*. The cathode and anode (the positive and negative sides at either end of a traditional battery) are hooked up to an electrical circuit.

The chemical reactions in the battery causes a build up of electrons at the anode. This results in an electrical difference between the anode and the cathode. You can think of this difference as an unstable build-up of the electrons. The electrons wants to rearrange themselves to get rid of this difference. But they do this in a certain way. Electrons repel each other and try to go to a place with fewer electrons.

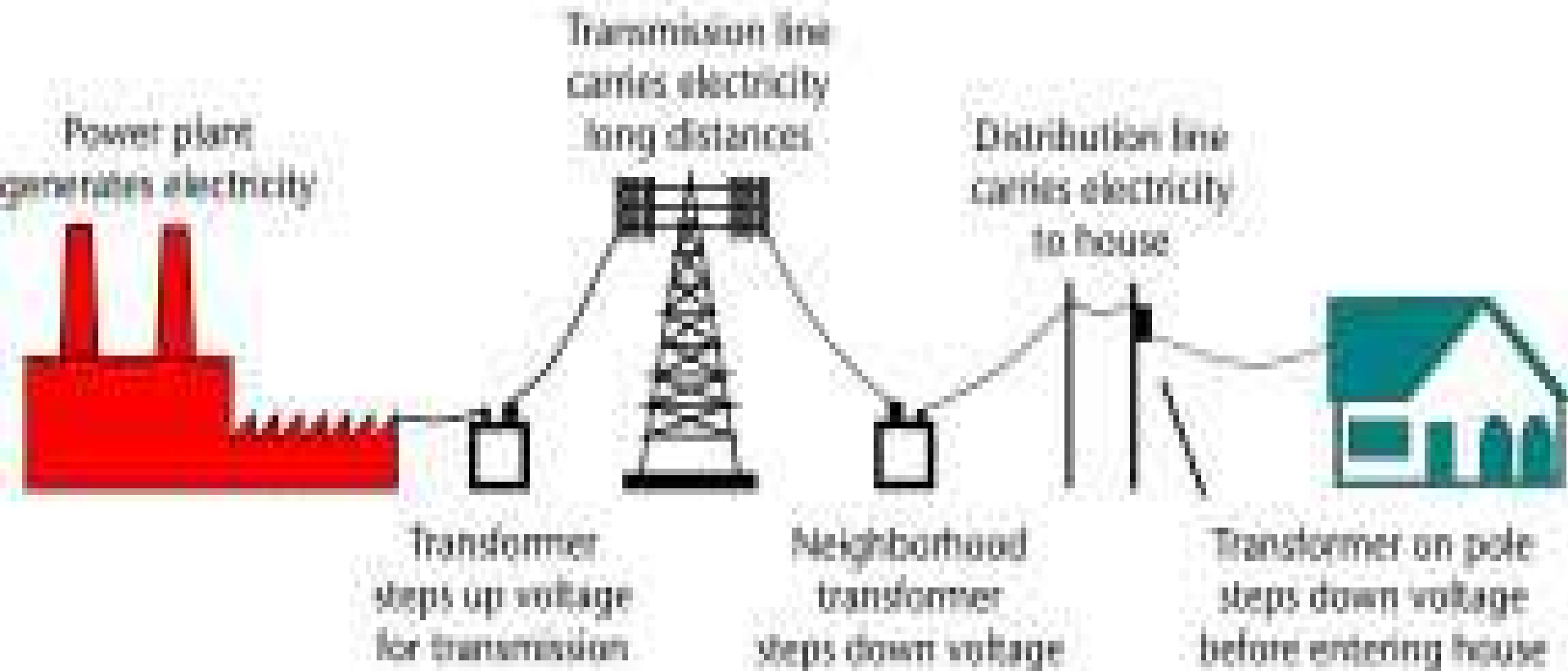


# Review Series and Parallel Circuits clip

# How a Lightbulb Works



# TRANSPORTING ELECTRICITY



$$P = IV$$

# Electric Power

- The rate at which electrical energy is transferred by an electric circuit.
- The SI unit of power is the watt
- Joule's Law  $P = IV$

# What is a kilowatt hour?

- How long you run an appliance.
  - How much energy is used?
- Energy used = Power (kW) x Time (hrs)
- $E = P \times t$
- To find cost:
- Ex: 10¢ per kilowatt hour
  - $E \times \$$

- 105 V are used to power an appliance that needs 15.0 amps. What is the power used?
  - $1575 \text{ W} = 1.575 \text{ kW}$
- How much energy is used when this appliance is used for 30.0 days- 24hrs a day?
  - $1134 \text{ kW-hr}$
- If the power company charges  $8\text{¢/Kw-h}$ , what is the cost of the energy above.
  - $\$90.72$

