## Honors Molecular Geometry Balloon Lab (Group 4)

## **Materials:**

11 atom balloons 1 lone pair balloon string scissors permanent marker

## **Procedure:**

- 1. Blow up all your balloons to equal sizes.
- 2. Tie six atom balloons together. Label as AX<sub>6</sub>.
- 3. Tie two atom balloons together. Label as AX<sub>2</sub>.
- 4. Tie three atom balloons and one lone pair balloons together. Label as AX<sub>3</sub>E<sub>1</sub>.
- 5. Make a chart for your data with the following columns: Molecule Name, Molecule Shape, and Drawing of Molecule

## **Questions:**

- 1. Draw the correct 3-D Lewis dot structure for ICl<sub>4</sub>- including the shape.
- 2. Draw the correct 3-D Lewis dot structure for XeF4 including the shape.
- 3. Draw the correct 3-D Lewis dot structure for BH3 including the shape.

