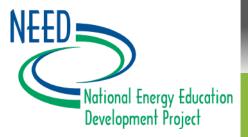


# **Forms of Energy**

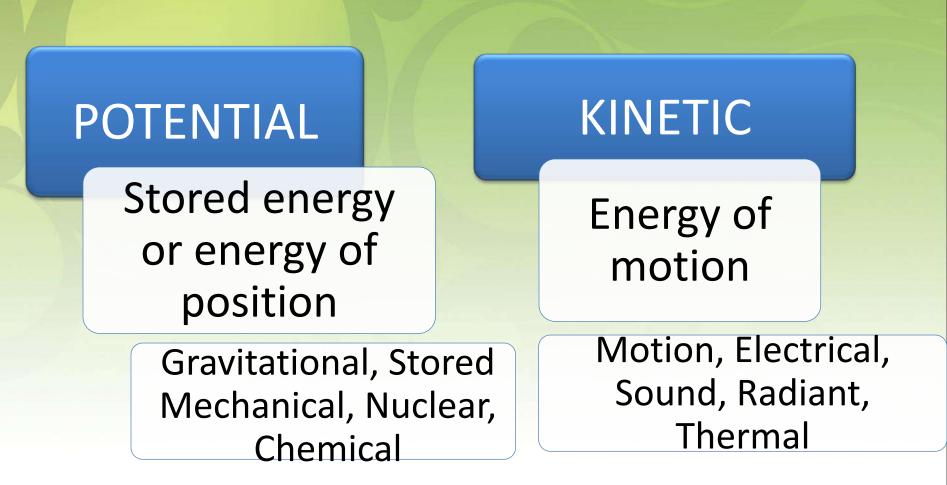


# What is energy?

- Ability to do work or cause change
- Produces Warmth
- Produces Light
- Produces Sound
- Produces Movement
- Produces Growth
- Powers Technology



# **Classes of Energy**



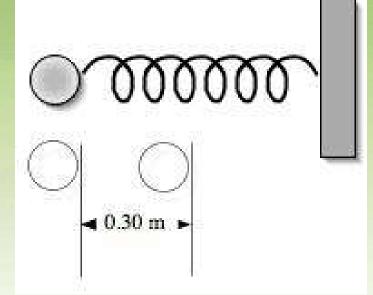
## **Gravitational Energy** –

energy an object or substance has because of its position Anything "up high"



#### **Stored Mechanical**

Energy – stored in an object by the application of force Must push or pull on an object



### **Nuclear Energy** –

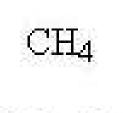
energy stored in the <sup>14</sup><sub>6</sub>Carbon</sup> nucleus of an atom Holds the atom together

> 6 protons 6 electrons 8 neutrons

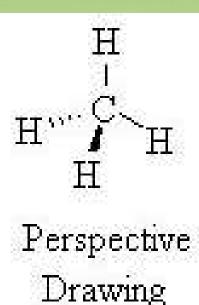
## **Chemical Energy** –

energy stored in the bonds between atoms

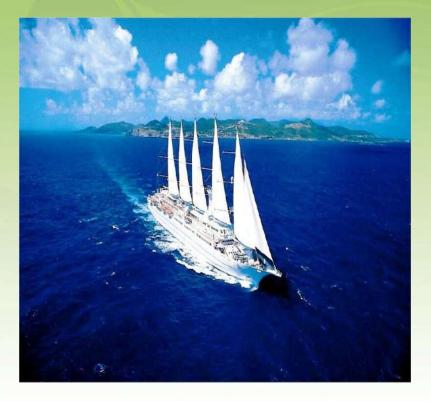
Holds molecules together



Molecular Formula



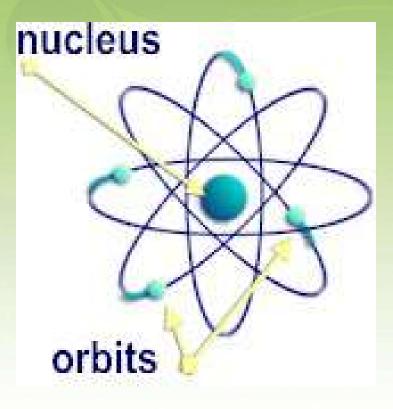
## Mechanical (Motion) Energy – movement of objects or substances from one place to another



# **Electrical Energy** –

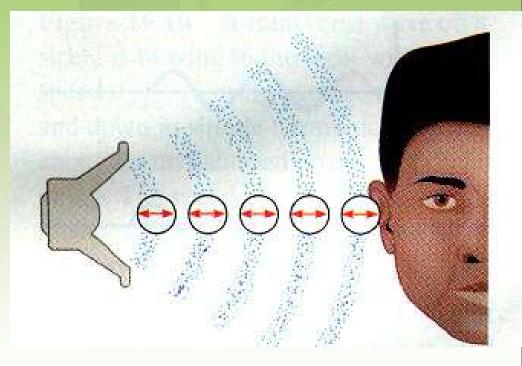
movement of electrons

#### NOT AN ELECTRON PARADE!



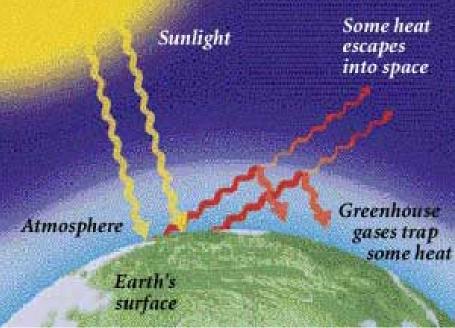
## Sound Energy -

movement of energy through substances in the form of longitudinal/compre ssion waves



### **Radiant Energy** –

electromagnetic energy that travels in transverse waves



## **Thermal (Heat) Energy**

internal energy of a substance due to the vibration of atoms and molecules making up the substance

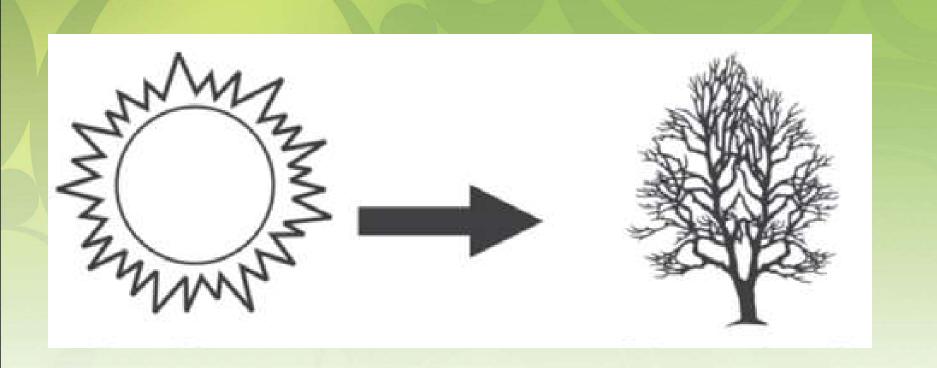


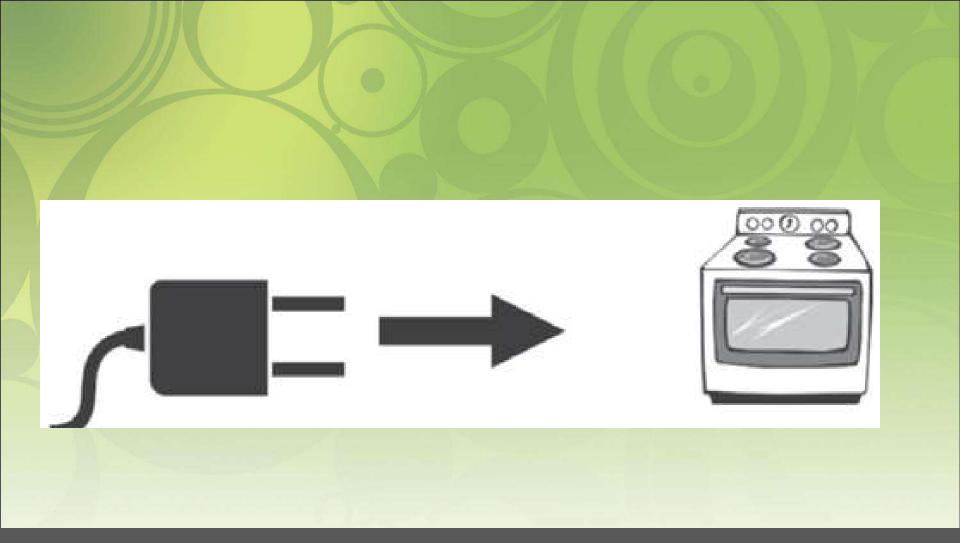
## **Energy Transfers**

1 – Energy can not be created nor destroyed, only changed.

- Law of Conservation of Energy
- First Law of Thermodynamics
- 2 Energy will always transfer from high to low.
- 3 No energy transfer is 100% efficient.









# In Review

Energy does work or causes change.
Two main classes of energy: potential and kinetic. Potential Energy – stored energy or energy of position
Gravitational, Stored
Mechanical, Nuclear, Chemical
Kinetic Energy – energy of motion
Motion/Mechanical, Electrical, Sound, Radiant, Thermal/Heat

•Energy can not be created nor destroyed, only changed.

- •Energy always transfers from high to low.
- •Energy transfers are never 100% efficient.