

Ch. 8: Reading Solubility Charts and Graphs

- How would you classify a solution of 40g of NaCl at 80°C? 1.
- 2. At what temperature would you have a saturated solution with 80g of potassium nitrate (KNO₃)?
- How would you classify a solution of 100g of sodium nitrate (NaNO₃) at 40°C? 3.
- How would you classify a solution of 100g of potassium chromate (K₂Cr₂O₇) at 90°C? 4.
- How would you classify a solution of 100g of potassium nitrate (KNO₃) at 50°C? 5.
- Which compound on this graph shows the *most* change in solubility from 0°C 100°C? 6.
- Which of the compounds shows the *least* change? 7.
- Which of the compounds on this graph is the least soluble at 0°C? 8.
- Which of the compounds is the most soluble at 100°C? 9.

Compound	0° C	20° C	60° C	100° C
Ammonium chloride	29.4	37.2	55.3	77.3
Copper (II) sulfate	23.1	32.0	61.8	114
Lead (II) chloride	0.67	1.0	1.94	3.2
Potassium bromide	53.6	65.3	85.5	104
Potassium chloride	28.0	34.0	45.8	56.3
Sodium acetate	36.2	46.4	139	170.15
Sodium chlorate	79.6	95.9	137	204

Solubility of Compounds in g/100g of Water at various Temperatures

- 1. At 50° C how would you classify a solution of 58.2g of ammonium chloride?
- 2. How much sodium acetate would dissolve in 200g of water at 60° C?
- 3. Which of the compounds listed above is the least soluble in water at 0° C?
- 4. How would you classify a solution of 46.3g of sodium acetate at 20° C?
- 5. Which of the compounds above shows the least change in solubility from 0° C 100°C?
- 6. Which shows the most change?
- 7. Which of the above compounds is the most soluble at 100° C?
- 8. How many grams of potassium bromide will it take to form a saturated solution in 300g of water at 100° C?
- 9. How many grams of sodium chlorate will it take to form a saturated solution in 200g of water at 60° C?
- 10. How much ammonium chloride would dissolve in 300g of water at 0° C?