Physical Science EOC REVIEW

PRACTICE EOC EXAM 1

A car plays its radio as it drives toward a student at a steady speed. How can the driver of the car best decrease the wavelength of the sounds heard by the student?

- Amake the radio louder
- Bmake the radio less loud
 Cincrease the speed of the car
- Ddecrease the speed of the car
 - ANSWER: C

The Doppler effect is an apparent change in what?

- Aamplitude
- Bintensity
- Cloudness
- Dpitch

• ANSWER: D

Which of the following is the bending of waves around a barrier?

- Adiffraction
- Binterference
- Creflection
- Drefraction
- ANSWER: A

What is the most penetrating kind of electromagnetic radiation?

- Agamma rays
- Bmicrowaves
- Cradio waves
- DX-rays
- ANSWER: A

The electromagnetic waves with the shortest wavelengths are called what?

- Agamma rays
- Bradio waves
- Clight
- DX-rays
- **ANSWER:** A

What occurs when compressions and rarefactions of different waves coincide?

- Aamplitude
- Bconstructive interference
- Cdestructive interference
- Dnoise pollution



Sound energy traveling through air is an example of what?

- Aenergy
- Bmatter
- Cenergy transferred
- Dmatter transferred through matter
- ANSWER: C

Which of the following has the greatest penetrating ability?

- AX-rays
- Bgamma rays
- Cradio waves
- Dultraviolet waves
- ANSWER: B

Electromagnetic energy comes in what form?

- Ainvisible form only
- Bvisible form only
- Cboth invisible and visible forms
- Dhigh frequencies only
- ANSWER: C

When the wavelength of a water wave is increased, what other characteristic of the wave will also increase?

- Aits frequency
- Bits period
- Cits amplitude
- Dits speed

ANSWER: B

A student connects a 10 V battery to a light bulb and a 20 ohm resistor. Which change will make the light bulb the brightest?

- Ause a 20 V battery with a 5 ohm resistor
- Buse a 20 V battery with a 20 ohm resistor
- Cuse a 10 V battery with a 5 ohm resistor
- Duse a 10 V battery with a 30 ohm resistor

• ANSWER: A

A circuit is in series if which of the following is true?

- Adifferent parts are on separate branches
- Belectrons may take different paths
- Celectrons have only one path to follow
- Dmore than one circuit exists



An instrument that measures the potential difference is called what?

- Aan ammeter
- Ba commutator
- Can ohmmeter
- Da voltmeter
- ANSWER: D

Resistance is measured in a unit called what?

- Aampere
- Bcoulomb
- Cohm
- Dvolt

• ANSWER: C

The rate at which an electrical device converts energy from one form to another is called what?

- Aelectrical energy
- Belectrical power
- Celectrical resistance
- Dvoltage regulation

ANSWER: B



A student walks into a bedroom and notices 2 switches. When one switch is flipped, the light in the room turns on. When the other switch is flipped, both the radio and the fan come on. Which statement best describes how the light, radio, and fan are connected? AAll three are connected in series.

- **BAll three are connected in parallel.**
- **C**The radio and the fan are connected in series.
- DThe radio and the fan are connected in parallel.
 - ANSWER: C

The electrons with iron, nickel and cobalt behave in a unique fashion and can create strong magnetic fields. A coil made of copper wire carrying an electric current can also create strong magnetic fields. Based on these facts, what can you conclude is the fundamental cause of all magnetic fields?

Aobjects made of iron, nickel or cobalt

- Bobjects made of copper
- Cobjects made of metal that are shaped like a horseshoe or a coil

Dobjects that contain moving electric charges

ANSWER: D

The magnetic force of a magnet is which of the following?

- Athe same at all parts of the magnet
- Bstrongest at the center
- Cstrongest at the poles
- Dweakest at the poles
- ANSWER: C

All of the following actions would weaken the strength of a permanent magnet *except* which one?

- Aheating it with a bunsen burner
- Bdropping it on the ground
- Chammering it with a hammer
- Dsubmersing it under water



Two socks are removed from a dryer and stick together because of static electricity. Which statement best describes the charge on the socks?

- AThey each have the same electric charge.
- BThey have opposite electric charges.
- CThey are both neutral.
- DThey have both been grounded.

ANSWER: B

When conducting demonstrations with a gold-leaf electroscope, students find that it is very difficult to cause the two leaves of the electroscope to attract and stick together. The best explanation for this is which of the following?

AIt is difficult to add electrons to one leaf and remove electrons from the other leaf.

BIt is difficult to remove electrons from one leaf while leaving the other leaf neutral in charge.

CIt is difficult to add electrons to both leaves.

DIt is difficult to remove electrons from both

leaves.

ANSWER: A

What is static electricity?

- Aelectrons that are moving
- Belectrons that move from one object to another and then stay at rest
- Celectric current
- Dprotons that are moving



Negatively charged particles emitted from a nucleus at a high speed are called what?

- Aalpha particles
- Bbeta particles
- Cgamma rays
- DX rays



The amount of energy it takes to raise the temperature of 1 kg of material 1 K is called what type of energy?

- Aheat energy
- Bmechanical energy
- Cspecific heat
- Dthermal energy



Through which of the following will convection most likely occur?

- Aliquids and gases
- Bsolids and liquids
- Csolids
- Dsolids and gases
 - ANSWER: A

What is matter called that has a definite volume and a definite shape?

- Agas
- Bliquid
- Cplasma
- Dsolid

• ANSWER: D



Which of the following is an example of physical change? Aboiling of water, bursting a balloon, and melting a candle Bburning of gasoline, rotting of an egg, and exploding fireworks Cfreezing of water, evaporation of gasoline, and rusting a nail Dsawing of wood, crushing a can, and • toasting a marshmallow

ANSWER: A



As the temperature of a substance increases which of the following is true?
AThe molecules in the substance are moving slower.
BThe molecules in the substance are at rest.
CThe molecules in the substance are moving faster.

DThe substance contains stored energy.

ANSWER: C

The sun's heat reaches the earth by means of which of the following?

- Aradiation
- Bconvection
- Cconduction
- Dconvection, radiation, and conduction



A car going 15 m/s travels on a level road. The driver uses the brakes to stop the car. Which statement best describes the changes in the kinds of energy the car has as it slows to a stop? AKinetic energy increases, potential energy decreases, no heat energy is given off. **BKinetic energy decreases, potential energy** remains the same, no heat energy is given off. **CKinetic energy decreases, potential energy** decreases, heat energy is given off. DKinetic energy decreases, potential energy C remains the same, heat energy is given off.

• ANSWER: D

Two toy cars are headed toward each other on a frictionless, level track. Car A has a kinetic energy of 30 J. Car B has a kinetic energy of 40 J. The cars collide and 10 J of heat energy is created in the collision. Which of these kinetic energies are most likely for cars A and B after the collision?
ACar A has 40 J of kinetic energy while car B has 30 J of kinetic energy.
BCar A has 25 J of kinetic energy while car B

has 35 J of kinetic energy.

CBoth cars have 35 J of kinetic energy.

DBoth cars have 25 J of kinetic energy

ANSWER: B

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A heat engine is a device used to turn thermal energy into useful work. Which statement best explains why heat engines are never 100% efficient? ASome thermal energy gets turned into mechanical energy and is wasted. BSome thermal energy gets turned into potential energy and is wasted. CSome thermal energy gets turned into kinetic energy and is wasted. DSome thermal energy escapes to the surroundings and is wasted.

ANSWER: D

A hot piece of iron is placed in contact with a cold piece of lead. Which best describes the changes in the amount of thermal energy for each block?

- AThe iron will gain heat energy while the lead will lose heat energy.
- BThe iron will lose heat energy while the lead will gain heat energy.

CBoth metals will lose heat energy.

• DBoth metals will gain heat energy.

ANSWER: B

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A student lifts a book bag from the floor to the seat of a chair. Which method will allow the student to complete this task while expending less power?

- Alifting the book bag more slowly
- Blifting the book bag more quickly
- Clifting a book bag filled with heavier books
- Dlifting the book bag from the floor to a desk in the same amount of time

ANSWER: A

A student lifts a toy car of mass 2 kg to the top of a ramp. When the car rolls to the bottom of the ramp, it gains a speed of 3 m/s. What was the potential energy of the car at the top of the ramp?

- A 3 J
 - B 6 J
 - C 9 J
 - D18 J

• ANSWER: C

The kinetic energy of an object increases as its what increases?

- Agravitational energy
- Bpotential energy
- Cspecific heat
- Dvelocity
- ANSWER: D

The rate at which work is done is called what?

- Aefficiency
- Beffort time
- Cforce
- Dpower
- ANSWER: D

A device that does work with only one movement and changes the size or direction of a force is what?

- Aa compound machine
- Ban effort machine
- Ca screw
- Da simple machine
- ANSWER: D



John goes over the top of a hill at a speed of 4.0 m/s while on his bicycle. Four seconds later, his speed is 24 m/s. What is John's acceleration?

- **A24 m/s**²
 - B20 m/s²
- **С б m/s**²
 - D 5 m/s²

• ANSWER: D



Susan drops a ball, and 2 seconds later the ball has a speed of 20 m/s. What is the acceleration of the ball?

- A40 m/s²
- B20 m/s²
- **C10 m/s**²
- D 5 m/s²

• ANSWER: C



Forces that are equal in size and opposite in direction are what kind of forces?

- Abalanced forces
- Bfrictional forces
- Cgravitational forces
- Dnet forces
 - ANSWER: A

A 3000-N force gives an object an acceleration of 15 m/s^2 . The mass of the object is what?

- A45,000 kg
- B 1,500 kg
- C 200 kg
- D 15 kg

• ANSWER: C

A 300-N force acts on a 25-kg object. The acceleration of the object is what?

- A7,500 m/s²
- B 300 m/s²
- C 25 m/s²
- D 12 m/s²

ANSWER: D



When a force is exerted on a box, an equal and opposite force is exerted by the box. These forces are called what kind of forces?

- Aaction-reaction
 - Bcentripetal
- Cfrictional
 - Dinertial
 - ANSWER: A

When an object moves in a circular path, it accelerates toward the center of the circle as a result of what force?

- Acentripetal force
- Bfrictional force
- Cgravitational force
- Dcentrifugal force
 - **ANSWER:** A



A parked car reaches a velocity of 60 m/sec in 12 seconds. What is the acceleration of the car? A720 m/sec/sec B 0.2 m/sec/sec C 5 m/sec/sec D60 m/sec/sec

ANSWER: C



A mail truck takes 20 seconds to move between mailboxes that are 10 m apart. What is the average speed of the mail truck?

- ■• A5 m/s
 - B2 m/s
- **C0.5 m/s**
 - D0.2 m/s
 - ANSWER: C

You are paddling a boat downstream at a speed of 4 m/sec. The river is flowing at a speed of 10 m/sec. How fast is your boat traveling?

- A 6 m/sec
- B14 m/sec
- C2.5 m/sec
- D 4 m/sec
- **ANSWER:** B



A bug crawls at a speed of 2 m/hr. If the bug has been crawling for 3.5 hours, how far has it traveled?

• A 7 m

- B11 m
- C 0.57 m
- D12 m

• ANSWER: A

Aluminum contains how many valence electrons?

- Athree valence electrons
- Bsix valence electrons
- Ceight valence electrons
- Dno valence electrons
- ANSWER: A

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- Dot diagrams are used to represent which of the following?
- Aatomic numbers
- Batomic mass
- Cisotopes
- Douter electrons
- ANSWER: D



- A certain atom has 26 protons, 26 electrons, and 30 neutrons. Its mass number is determined to be what?
- A26
 B30
 C52
- D56

• ANSWER: D



- Of the following elements which one has the most even distribution of isotopes?
- Afluorine
 - Bchlorine
- Cbromine
 - Diodine

• ANSWER: B

- A particle that moves around the nucleus is called which of the following?
- Aan electron
- Ban ion
- Ca neutron
- Da proton

• ANSWER: A

Atoms of the same element with different numbers of neutrons are called what?

- Aisotopes
 - Bmetals

- Cmetalloids
- **Dtransition elements**
 - ANSWER: A



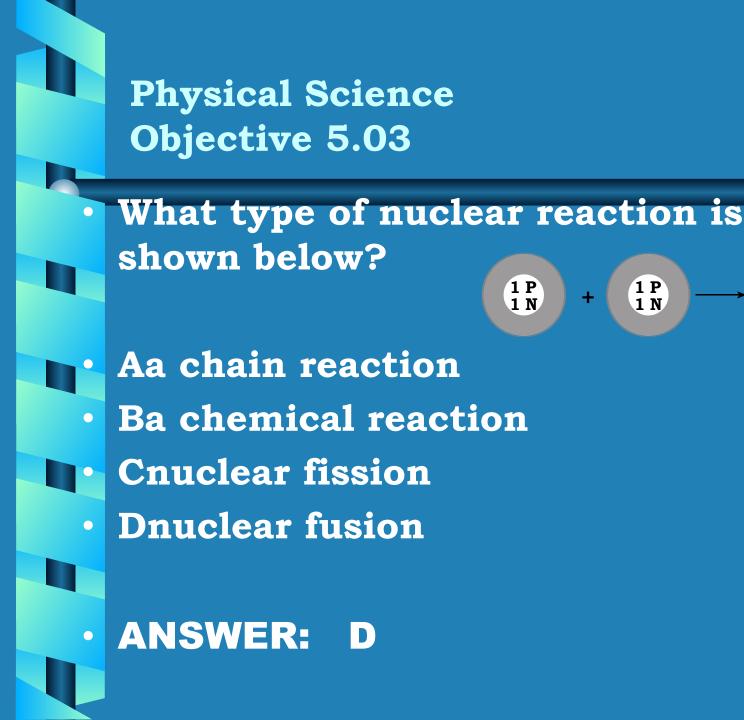
- When two or more substances are combined so each substance maintains its own properties, the result is called what?
- Aa chemical change
 - Ba compound
- Can element
 - Da mixture
 - ANSWER: D

- Two small nuclei are joined together to form a larger nucleus and giving off energy. What is this process called?
- Agamma decay
- Balpha decay
- Cnuclear fission
- **Dnuclear fusion**

ANSWER: D

- Which of the following radioactive changes result in a decrease in the number of nuclear particles present?
- Anuclear fusion
- Bnuclear fission
- Calpha decay
- Dbeta decay

• ANSWER: A



2 P 2 N

What kind of chemical bond is formed when electrons are transferred from atom to atom?

- Acovalent
- Bhydrate
- Cionic
- Dmagnetic



The oxidation number of an atom is shown with which of the following?

- Anegative number
- Bpositive number
- Csubscript
- Dsuperscript
- **ANSWER: D**

The sum of the oxidation numbers in a neutral compound is which of the following?

- Aa negative number
- Bone
- Ca positive number
- Dzero



The elements that make up a compound and the ratios of the atoms of those elements can be shown in which of the following?

- Achemical formula
- Bchemical symbol
- Csubscript
- Dsuperscript

• ANSWER: A

What is the total number of atoms in the compound Ca(ClO₃)₂?

- A3
- B5
- C6
- D9

• ANSWER: D

A group of atoms that acts together as one charged atom is which of the following?

- Acrystal
- Bmolecule
- Cnegative ion
- Dpolyatomic ion



In a chemical formula, the ratio of atoms in the compound is shown by the numbers called what?

- Apostscripts
- Boxidation numbers
- Csubscripts
- Dsuperscripts
- ANSWER: C

How many hydrogen atoms are present in one molecule of ammonium acetate, NH₄C₂H₃O₂?

- A 3
- **B** 4
- C 7
- D12

• ANSWER: C

What is the name of the compound with the formula NaCl?

- Achlorine sodiate
- Bsodium chlorate
- Csodium chloride
- Dsodium dichloride
- ANSWER: C

What is the ratio of potassium atoms to oxygen atoms in a binary compound made from these two elements?

- A1:1
- B1:2
- C1:3
- D2:1



What is the name of a binary compound made up of lithium and chlorine?

- Achlorine lithiate
- Bchlorine lithium
- Clithium chloride
- Dlithium chlorate
- ANSWER: C

What is the correct name for K₂SO₄?

- Apotassium disulfide
- Bpotassium sulfate
- Cpotassium sulfide
- Dpotassium(II) sulfate



What is the charge of phosphate in K₃PO₄?

- A7–
- **B**3–
- C1+
- D5+



What is the correct formula for magnesium oxide?

- AMgO
- BMgO₂
- **CMg**₂**O**₂
- **DMg**₂**O**

• ANSWER: A

Which of the following is the correct formula for magnesium nitrate?

- AMgNO₃
- **BMg**₂**NO**₃
- CMg(NO₃)₂
- DMg₂ (NO₃)₂
 - ANSWER: C

 $2H_2O \rightarrow 2H_2 + O_2$ is an example of what type of reaction?

- Adouble replacement reaction
- Bsingle replacement reaction
- Csynthesis reaction
- Ddecomposition
- ANSWER: D

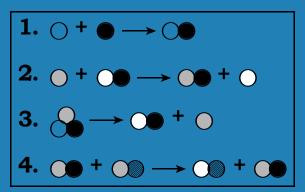
When one element displaces another element in a compound, the reaction is called what?

- Adecomposition
- Bdouble-displacement
- Csingle-displacement
- Dsynthesis



The reaction represented by model 4 in the figure below is called what type of reaction?

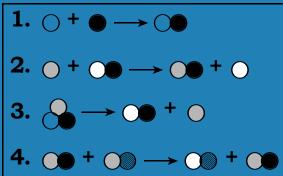
- Asingle displacement
- Bdecomposition
- Cdouble displacement
- Dsynthesis



• ANSWER: C

The reaction represented by model 3 in the figure below is called what type of reaction?

- Asingle displacement
- Bdecomposition
- Cdouble displacement
- Dsynthesis



ANSWER: B

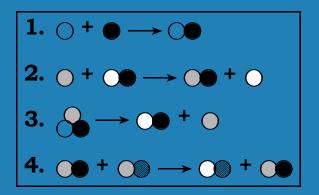
- The reaction represented by model 1 in the figure below is called what type of reaction?
- Asingle displacement
- Bdecomposition
- Cdouble displacement
- Dsynthesis

$$1. \bigcirc + \bullet \longrightarrow \bigcirc \bullet$$
$$2. \bigcirc + \bigcirc \bullet \longrightarrow \bigcirc + \bigcirc$$
$$3. \bigcirc \bullet \longrightarrow \bigcirc + \bigcirc$$
$$4. \bigcirc + \bigcirc \oplus + \bigcirc \bigcirc + \bigcirc \bullet$$

ANSWER: D

The reaction represented by model 2 in the figure below is called what type of reaction?

- Asingle displacement
- Bdecomposition
- Cdouble displacement
- Dsynthesis



• ANSWER: A

What type of reaction is shown in the following chemical equation?

 $NH_3 + HC1$ NH_4C1 —

- Adecomposition
- Bdouble displacement
- Csingle displacement
- Dsynthesis



Which of the following is a balanced chemical equation?

- $AAgNO_3 + NaC1 \rightarrow 4AgC1 + 2NaNO_3$
- $B2AgNO_3 + 2NaC1 \rightarrow 3AgC1 + 2NaNO_3$
- $CAgNO_3 + NaCl \rightarrow AgCl + NaNO_3$
- $DAgNO_3 + 2NaC1 \rightarrow AgC1 + 3NaNO_3$
- ANSWER: C

The breaking down of a compound into simpler substances is called what?

- Adecomposition
- Bdisplacement
- Cproduction
- Dsynthesis
- ANSWER: A

Each substance on the left side of the arrow in a chemical equation is which of the following?

- Acatalyst
- Bcoefficient
- Cproduct
- Dreactant
- ANSWER: D

Each substance to the right of the arrow in a chemical equation is called what?

- Acatalyst
- Binhibitor
- Cprecipitate
- Dproduct
- ANSWER: D

According to the law of conservation of

mass, how does the mass of the products in a chemical reaction compare to the mass of the reactants?

- AThere is no relationship.
 - BThe mass of the products is greater.
- CThe mass of the reactants is greater.
 DThe masses are equal
 - DThe masses are equal.
 - ANSWER: D

What is a solution?

- Atype of element
- Btype of molecule
- Chomogeneous mixture
- Dheterogeneous mixture



Elements with oxidation numbers of -2 belong to what family of elements?

- Anitrogen family
- Boxygen family
- Calkaline earth family
- Dhalogen family
- **ANSWER:** B

Elements in Groups 3 through 12 of the periodic table are called what?

- Ametalloids
- Bmetals
- Cnoble gases
- Dtransition elements
- ANSWER: D

A chemical symbol represents which of the following of an element?

- Aname
- Breaction
- Cgroup
- Dstructure

• ANSWER: A

Horizontal rows of the periodic table are called what?

- Aclusters
- Bfamilies
- Cgroups
- Dperiods
- ANSWER: D

In a chemical equation, the symbol that means "dissolved in water" is which of the following?

- A(aq)
- B(cr)
- C(I)
- D(g)



Numbers that precede symbols and formulas in a chemical equation are called what by chemists?

- Acatalysts
- Bcoefficients
- Csubscripts
- Dsuperscripts
- ANSWER: B

A chemical reaction in which energy is released is called what?

- Aendothermic
- Bexothermic
- Cflammable
- Da formula
- ANSWER: B

A substance that speeds up a chemical reaction without undergoing a permanent change itself is which of the following ?

- Aa catalyst
- Ba coefficient
- Can inhibitor
- Da reactant

ANSWER: A

If thermal energy must be added to a chemical reaction for the reaction to take place, the reaction is called which of the following?

- Abalanced
- Bendothermic
- Cexothermic
- Dreactant

ANSWER: B

 \bullet

Substances that prevent chemical reactions are called what by scientists?

- Acatalysts
- Binhibitors
- Cproduct
- Dreactants



A solution that contains all the solute it can hold at a given temperature is called what kind of solution?

- Adiluted
- Bsaturated
- Csupersaturated
- Dunsaturated
- ANSWER: B

Which of the following will speed up the dissolving of a solid solute in water?

- ACool the solution.
- BFreeze the solute.
- CGrind up the solvent.
- DStir the solution
 - ANSWER: D

The amount of solute that can be dissolved in a specific amount of solvent at a given temperature is called what?

- Aconcentration
- Bdensity
- Cdilution
- Dsolubility



Adding more solute to a solvent does what to the solvent?

- Adecreases its boiling point
- Bdoes not affect its boiling point
- Cincreases its boiling point
- Dincreases its freezing point
- ANSWER: C

A parked car reaches a velocity of 60 m/sec in 12 seconds. What is the acceleration of the car?

- A720 m/sec/sec
- B0.2 m/sec/sec
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Answer: C



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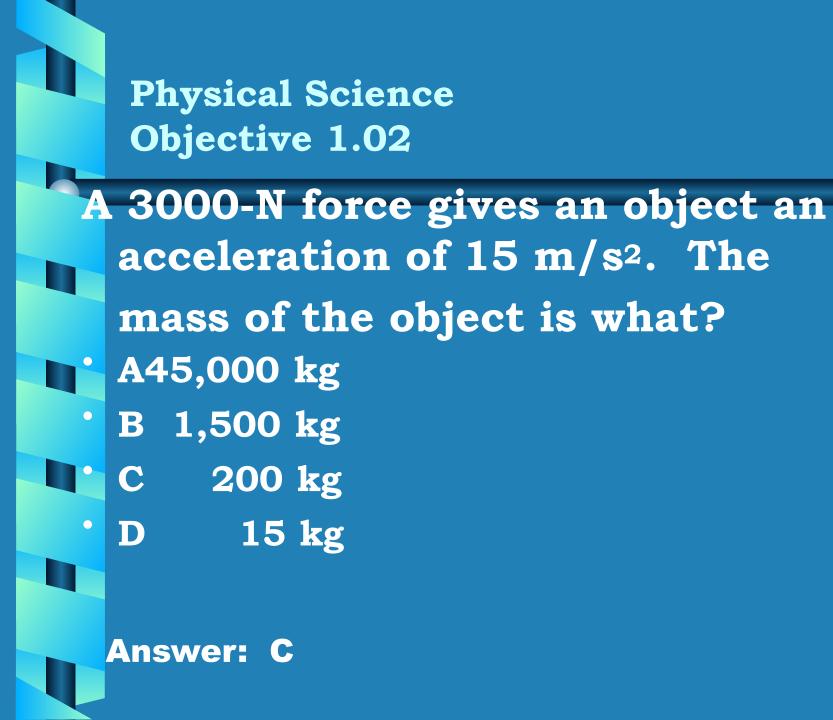
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- $D \quad 12 \text{ m/s}^2$

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To test a hypothesis you
Adraw a conclusion
Bexperiment
Ccollect information
Dcommunicate



Scientists do not
Asearch for evidence
Bhave a good imagination
Cbase conclusions on supernatural powers
Dtest a hypothesis





The number of variables in an experiment should be



The system of measurement commonly by scientist throughout the world is the

- Ametric
 - BEnglish
- Cinternational
 - DAmerican



The base unit of length is

- Ameter
 - Bliter
 - Cgram
 - DCelsius





The base unit of mass is

- Ameter
 - Bliter
 - Cgram
 - DCelsius





The base unit of volume is

- **Ameter**
 - Bliter
 - Cgram
 - DCelsius





The base unit of temperature is

- I Ameter
 - Bliter
- Cgram
- DCelsius





A kilogram equals
A10 grams
B1000 grams
C100 grams
D.1 grams



After a first attempt to solve a problem, a scientist most likely will

- Areport conclusions to other scientists
- Brepeat an experiment

- **Cwrite articles for scientific journals**
- Dproceed to find new and different problems



A factor that changes when introduced into an experiment is called the

- Avariable
- 💽 Bcontrol
- Ccheck factor
- Dweight factor



Factors that stays the same in an experiment are called the

- Avariable
- **Bcontrol**
 - Ccheck factor
- Dweight factor