

Chapter 14Water Pollution P. 381 AXES Paragraph P. 20 NB

Agenda 12/1

- Welcome back!
- Start CH 14, Warm up, pg. 52
- Pass back: PG & E assignment, Watt Meter labs
 - Need PG & E assignments from....
- Read Opening Story CH 14
- Go over Topic IV Test, Test Corrections due Friday!
- Current Events:
 - Taylor, Jonathan, Brad
- Ecocolumn Lab help

12/1 Water Pollution CH 14

Obj. TSW distinguish the difference between point and nonpoint source pollution. P. 52

NB

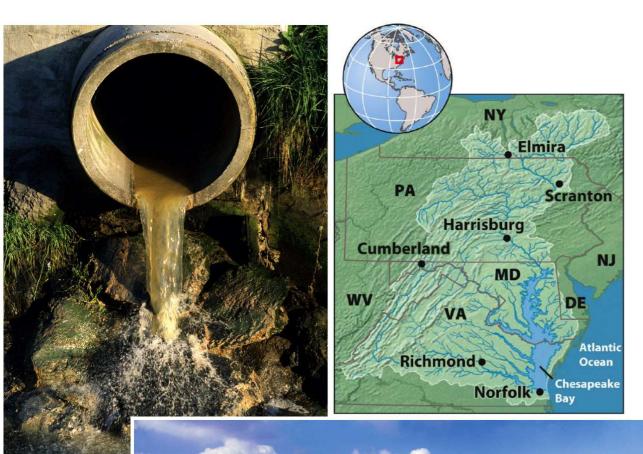


Figure 14.1a
Environmental Science

- I. Compare & ContrastPoint Source & NonPoint Source Pollution.
- What are three reasons scientists are concerned about human waste water?
- 3. Explain BOD.

Water Pollution

Water pollution- the contamination of streams, rivers, lakes, oceans, or groundwater with substances produced through human activities and that negatively affect organisms.





Figure 14.1a

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Which is Point source and Non-point source?



Figure 14.1b

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Water Pollution

- Point sources- distinct locations that pump waste into a waterway.
- Nonpoint sources- diffuse areas such as an entire farming region that pollutes a waterway.
- Important difference because:
 - Distinction can help in controlling pollutant inputs to waterways
 - EX)?

Human Wastewater

Water produced by human activities such as human sewage from toilets and gray water from bathing and washing clothes or dishes.

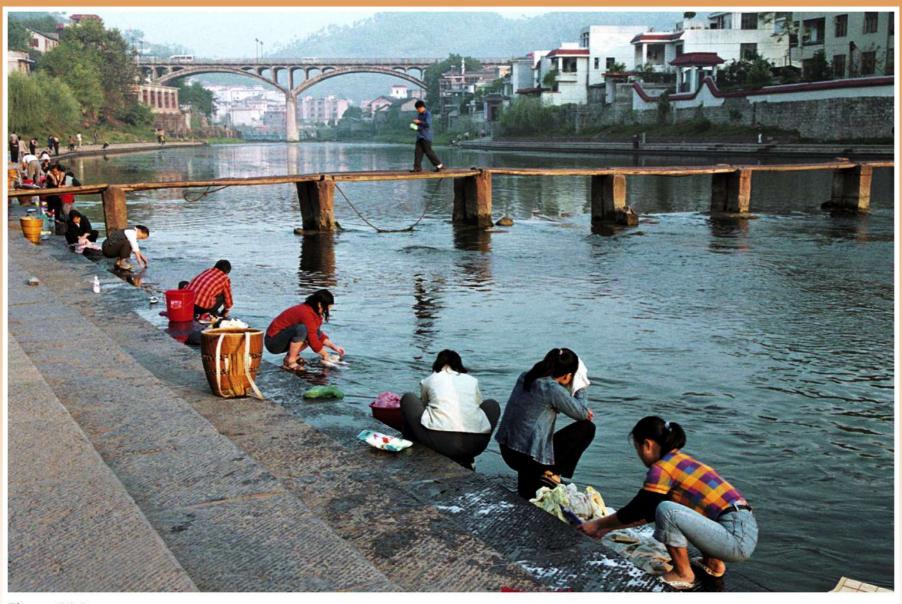


Figure 14.2

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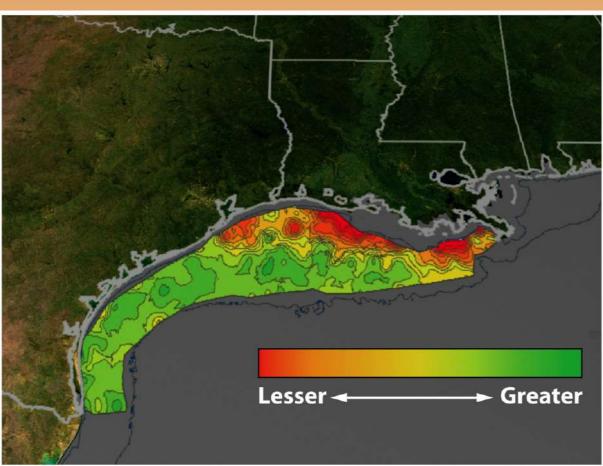
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Three reasons scientists are concerned about human wastewater:

- Oxygen-demanding wastes like bacteria that put a large demand for oxygen in the water.
- Nutrients that are released from wastewater decomposition can make the water more fertile causing eutrophication.
- Wastewater can carry a wide variety of diseasecausing organisms.

Biochemical Oxygen Demand (BOD)

- BOD- the amount of oxygen a quantity of water uses over a period of time at a specific temperature.
- Lower BOD values indicate the water is less polluted and higher BOD values indicate it is more polluted by wastewater.



Oxygen concentrations in Gulf Coast waters



Fish Figure 14.3b Figure 14.3b

Figure 14.3a Environmental Science © 2012 W. H. Freeman and Company

Dead Zone – Raw Sewage

Eutrophication

- Eutrophication is an abundance of fertility to a body of water.
- Eutrophication is caused by an increase in nutrients, such as fertilizers.
- Eutrophication can cause a rapid growth of algae which eventually dies, causing the microbes to increase the BOD.

Agenda 12/2

- Warm up, pg. 54
- Stamp CH 14 MC SG, go over answers, pg. 55
- Pollution Solution Presentations, pg. 55
- Ecocolumn Lab Time
- Video, notes on pg. 59

12/2 Pollution: Waste, Treatment, and Human Health. CH 14 Obj. TSW learn and discuss diseases in our water, how to clean waste water and how inorganic compounds can cause a risk to human health. P. 54 NB

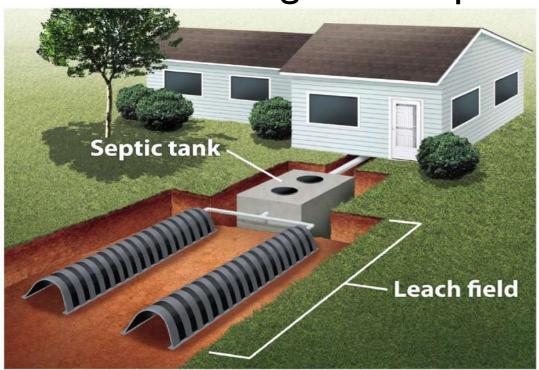


Figure 14.5
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Figure 14.7

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- Identify some disease causing organisms, & explain what an indicator species is.
- Describe and contrast the two most common ways to treat wastewater.
- 3. ID some heavy metals and explain the risk to human health and the environment.

Common Diseases from Human Wastewater

- Cholera
- Typhoid fever
- **■** Stomach flu
- Diarrhea
- Cholera
- Hepatitis

Indicator Species

- Given the risk that so many pathogens pose= Need to be able to test if they are in our drinking water
- Hard to test for all of these pathogens
- Use indicator species
- Organisms that indicate whether or not diseasecausing pathogens are likely to be present
- Best indicator:
 - Fecal Coliform Bacteria

Treatments for Human and Animal Wastewater

■ Septic systems- a large container that receives wastewater from the house.

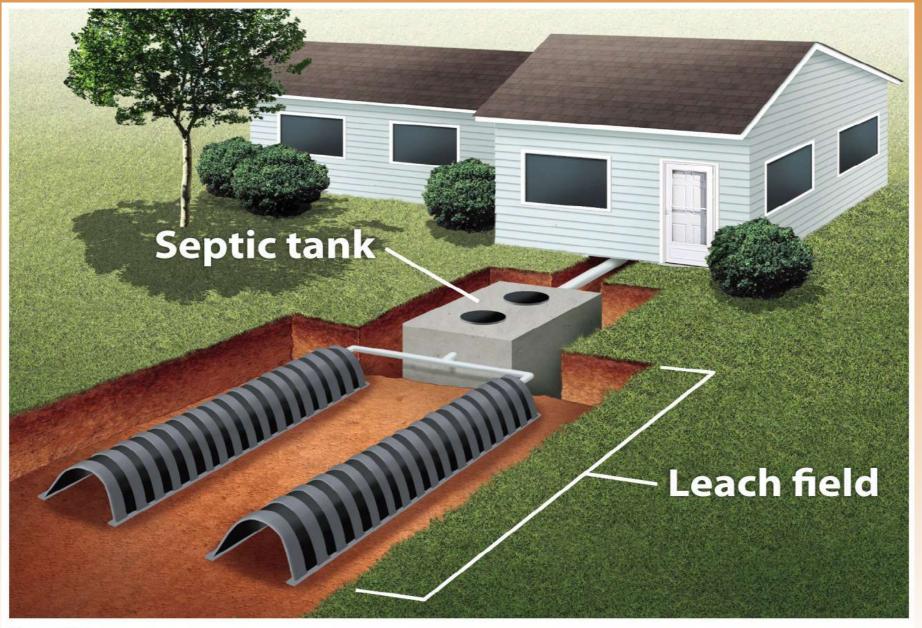


Figure 14.5

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Treatments for Human and Animal Wastewater

Sewage Treatment Plants- centralized plants in areas with large populations that receive wastewater via a network of underground pipes.

A sewage Treatment Plant

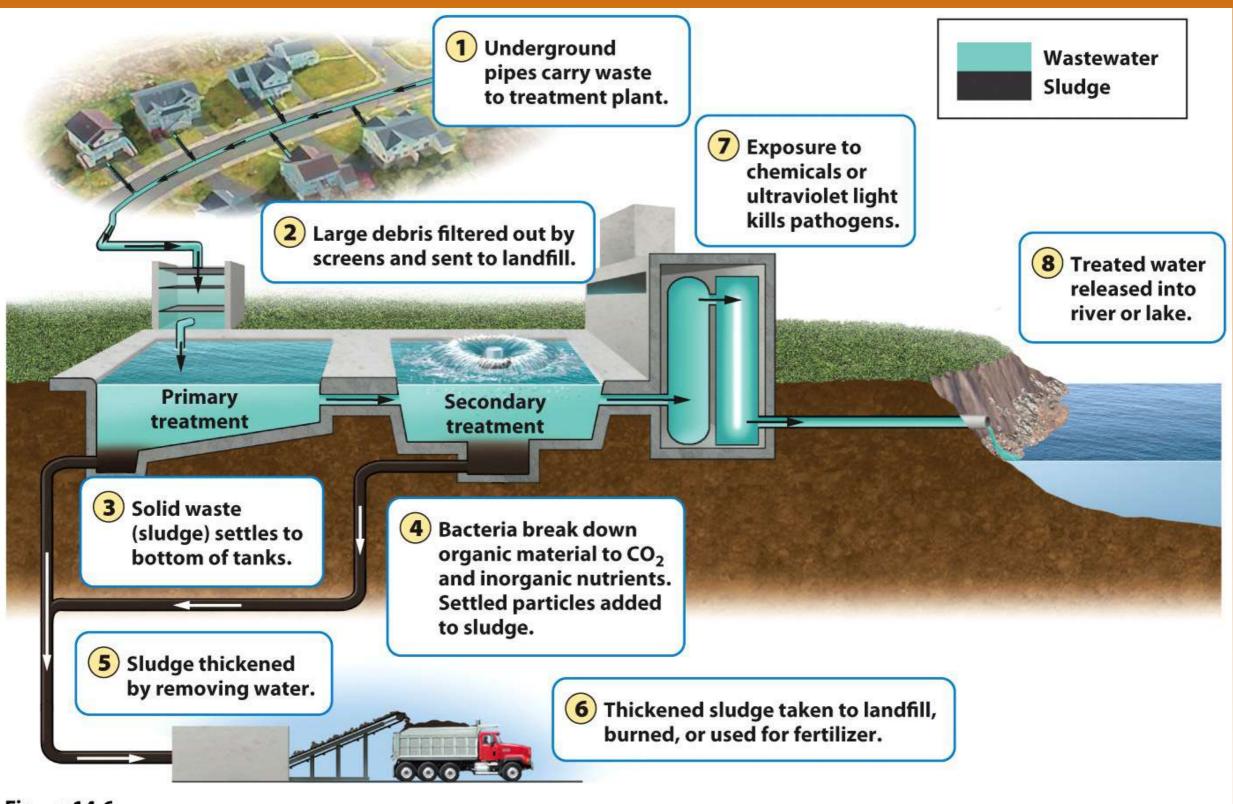


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Treatments for Human and Animal Wastewater

■ Manure lagoons- large, human-made ponds line with rubber to prevent the manure from leaking into the groundwater. After the manure is broken down by bacteria, it is spread onto fields as fertilizers.

Biodigesters collect methane produced from the microbial breakdown and convert it to electricity for the electrical grid during peak demand.



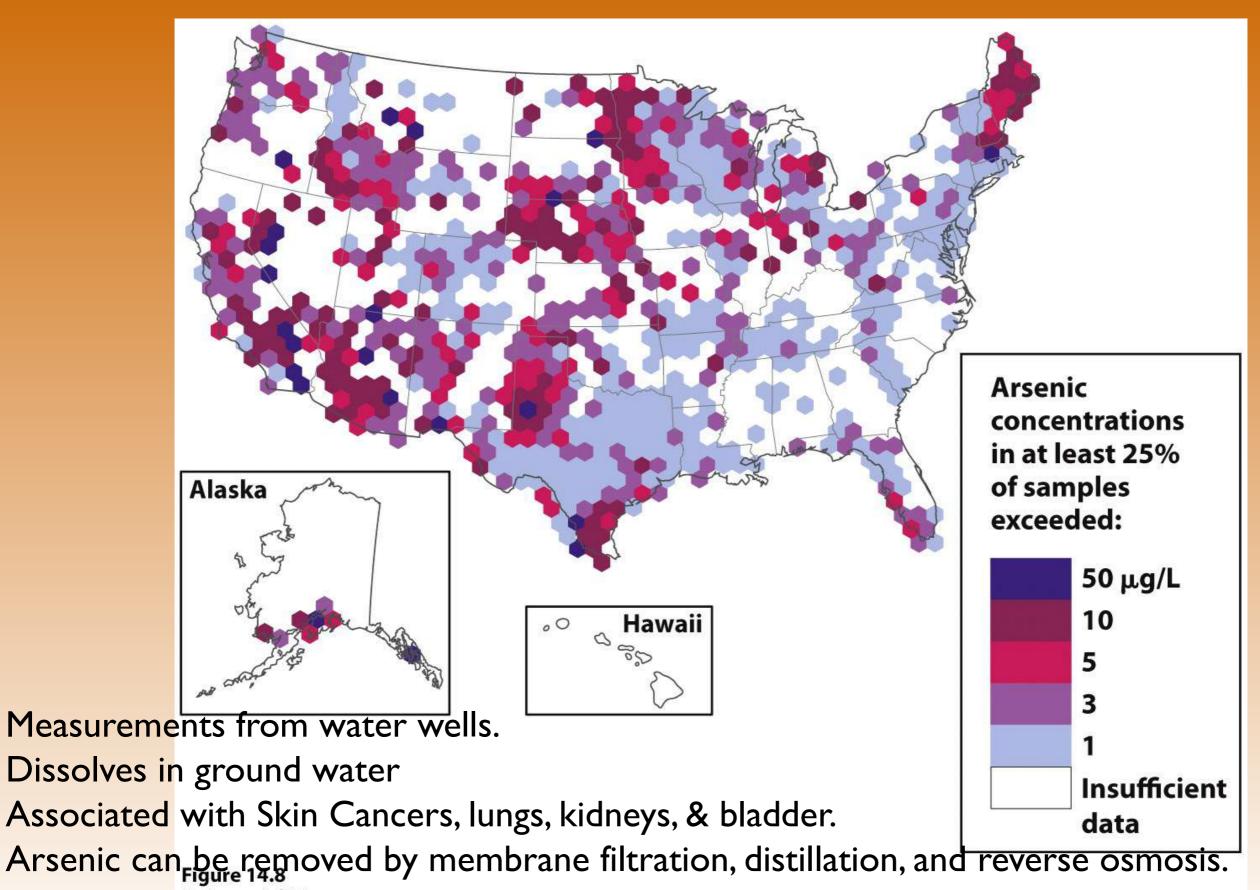
Figure 14.7

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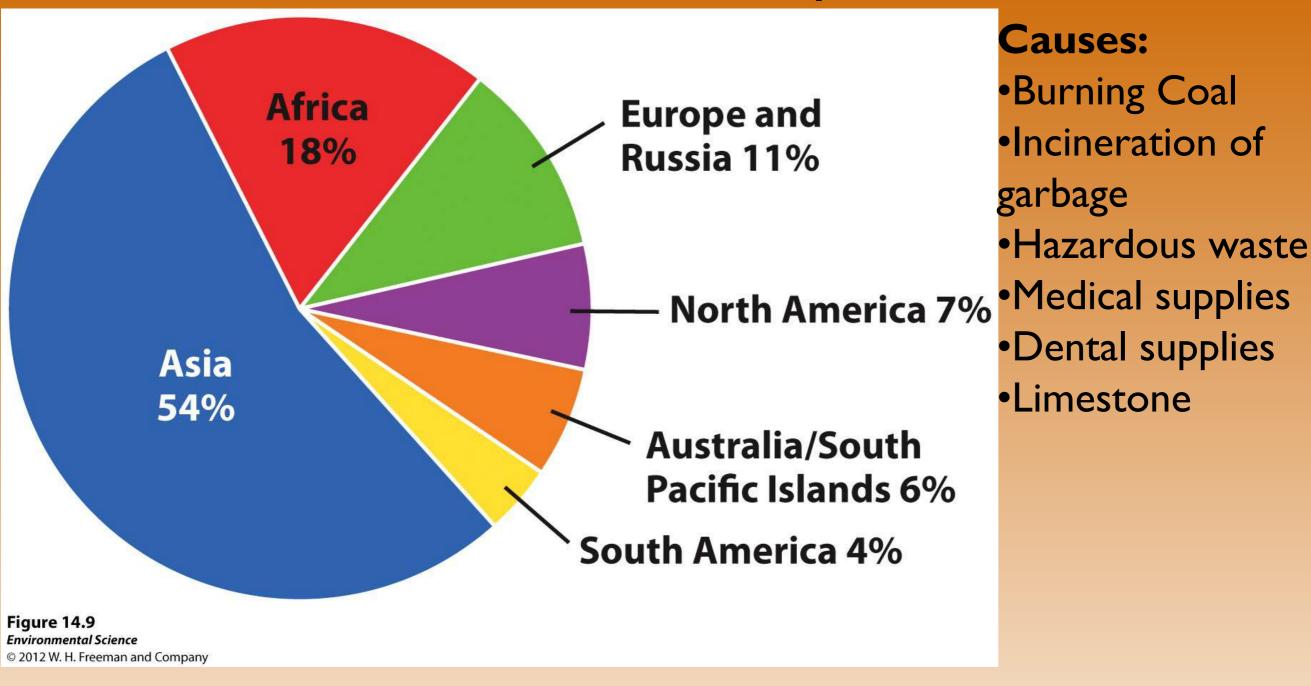
Heavy Metals and Other Substances that can threaten human Health and the Environment

- Lead
- Arsenic
- Mercury
- Acids
- Synthetic compounds (pesticides, pharmaceuticals, and hormones)



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World Mercury Production



Pollution Solution

- Groups of three, Five topics:
 - Oil Pollution: Prishna, Lovpret, Mandeep
 - Solid Waste Pollution: Jeanelle, Matt, Ayat, Jon
 - Sediment Pollution: Monica, Luis, Bryan, Rachana
 - Thermal Pollution: Brad, Magda, Ellie, Taylor
 - Noise Pollution: Daniela Jasmine, Joseph

 Explain what it is and how we can remediate it? (Use book and outside sources)

Do the Math

- In 2010, the South Anna River had a largemouth bass population of 25,000 individuals over a 10-mile stretch. In early 2011, a chemical spill occurred and the population of largemouth bass decreased to only 5000 individuals over the same area. Calculate the **percent change** in largemouth bass in the South Anna River.
- 25,000 5,000 = 20,000
- \bullet 20,000/25,000 = .80
- $.80 \times 100 = 80\%$
- There was an 80% change in the large mouth bass found in the south Anna river after the oil spill.

12/3 Water & Air Pollution Review CH 14 & 15 Obj. TSW discuss thermal, and our Nations water laws. P. 52 NB



Figure 14.19
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TABLE 14.1

The maximum contaminant levels (MCL) for a variety of contaminants in drinking water as determined by the U.S. Environmental Protection Agency, in parts per billion (ppb)

Contaminant category	Contaminant	Maximum contaminant level (ppb)
Microorganism	Giardia	0
Microorganism	Fecal coliform	0
Inorganic chemical	Arsenic	10
Inorganic chemical	Mercury	2
Organic chemical	Benzene	5
Organic chemical	Atrazine	3

- Source: U.S. Environmental Protection Agency, http://www.epa.gov/safewater/contaminants/index.html.
- **Table 14.1** *Environmental Science* © 2012 W. H. Freeman and Company

- Discuss what thermal pollution is and a solution to the problem.
- 2. Why is Noise considered water pollution?
- 3. Discuss what some of our Nation's water laws are?



Figure 14.10
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Acid Mine Drainage - Italy

Streams with low pH

Tall smokestacks from Industrial plants burning coal were releasing sulfur dioxide & nitrogen dioxide into the air.

Sulfuric acid and nitric acid were formed in the atmosphere and returned hundreds of kilometers away as acid deposition — rain & snow. pH below 5 Installation of Scrubbers to decrease acids.

Contaminants in StreamsWaste Water

- Agriculture
 - Forestry
 - Industry

USGS tested 139 streams across the US they contained antibiotics, reproductive hormones, nonprescription drugs and steroids

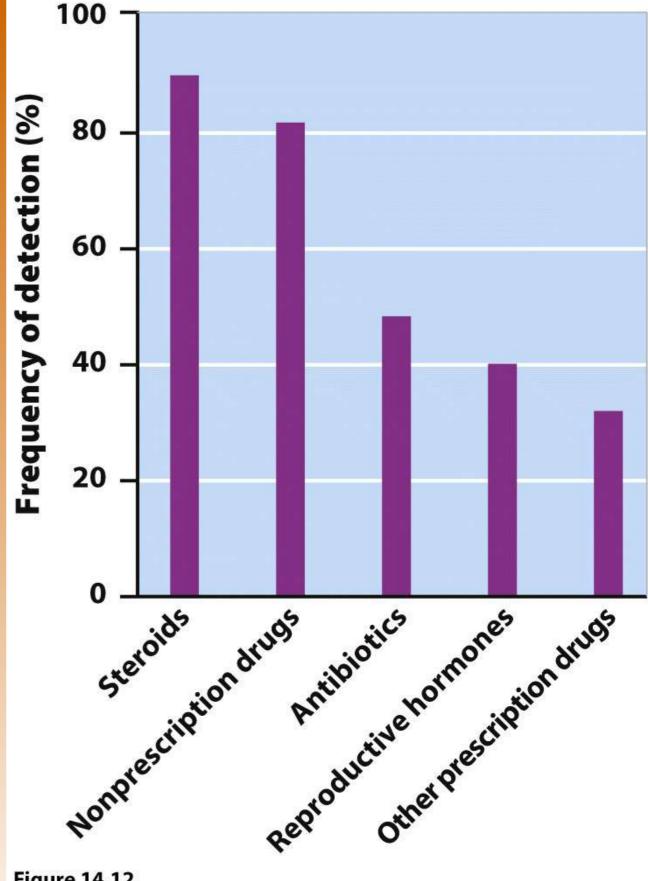


Figure 14.12

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Oil Pollution in the Ocean

Approx. 5,000

offshore oil platforms in North America. Oil Tankers have double hulls to help prevent spills. What surprises you about these graphs?

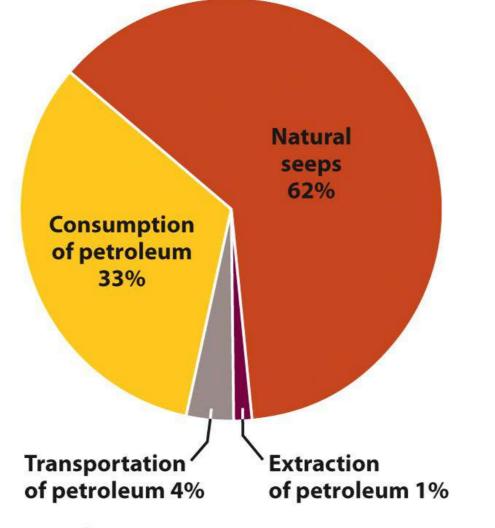
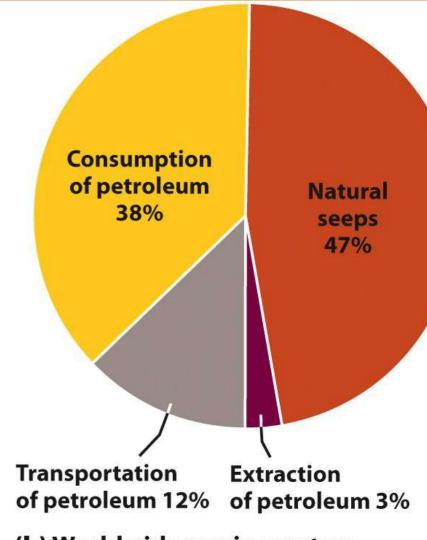




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(b) Worldwide marine waters



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Ways to Remediate Oil Pollution

- Containment using booms to keep the floating oil from spreading.
- Chemicals that help break up the oil, making it disperse before it hits the shoreline.
- Bacteria that are genetically engineered to consume oil

Other Water Pollutants

- Solid waste pollution (garbage)
- Sediment pollution (sand, silt and clay)
- Thermal pollution
- Noise pollution



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Figure 14.17

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Indonesia

Thermal Pollution – Nuclear Reactors

Empty the water into holding ponds before it returns to the river or lake or ocean.

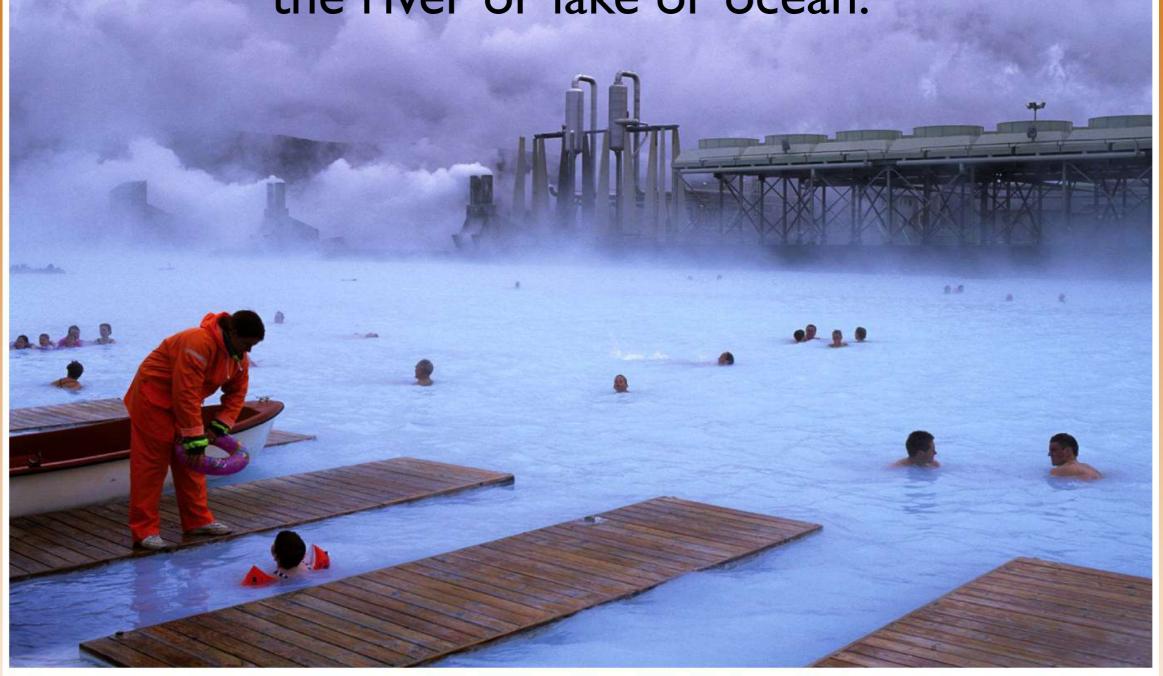
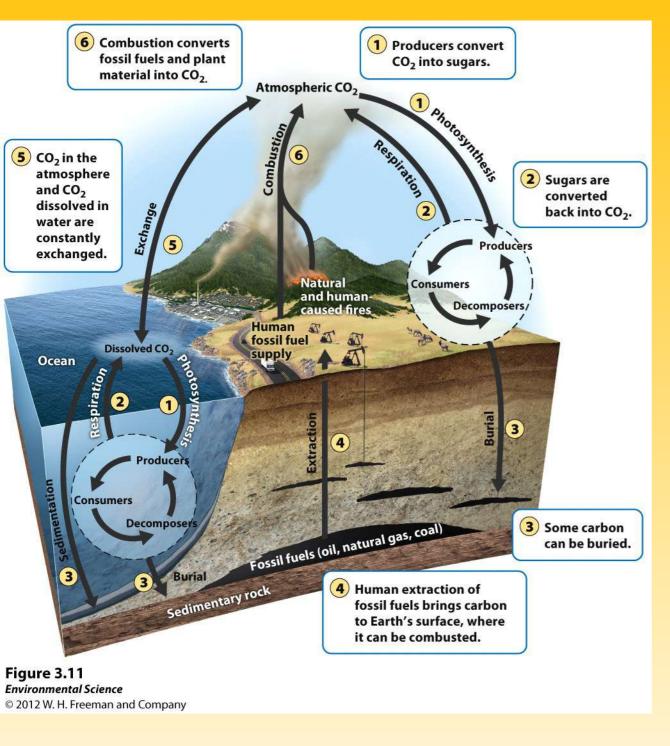


Figure 14.19
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Iceland – Svartsengi Power Plant

Solution – to use a cooling tower to reduce the temperature of water by evaporation.

LAB: Waste and its effect on atmospheric carbon dioxide



- P. 68 Env. Sci book
- Obj. to evaluate the effect of the presence of pollutants such as sewage, agricultural runoff etc. on atmospheric carbon dioxide levels in the environment.
- P. 25 NB

Procedure

- I. In pairs, Get a GLX and a CO2 sensor probe.
- 2. Turn on your GLX, plug in your CO₂ sensor probe. Note the value in the air, write this CO₂ value on your lab.
- 3. Using a graduated cylinder, add 5 ml of tap water to the 250 -ml plastic CO₂ collection bottle, write the CO₂ value on your lab.
- 4. Add 10 ml of milk to the 250- ml gas sampling bottle, write the CO₂ value on your lab. Dry the graduated cylinder.
- 5. Noting the time, add 10 ml of yeast mixture to the 250ml gas sampling bottle and swirl to mix.
- 6. Connect the probe to the bottle and record the data every minute for the next 4 minutes. Control is Time 0.
- 7. Now, repeat the 1-6 steps, but add 30ml of milk.

Data:

CO₂ level in the classroom

Time	0 (Control) minutes	I minute	2 minute	3 minute	4 minutes
I0ml Milk [CO ₂]					
30ml Milk [CO2]					

Qualitative data: Did you feel heat from the 250ml gas sampling bottle? Why?

Lab due:

Water Laws

- Clean Water Act- (1972) supports the "protection and propagation of fish, shellfish, and wildlife and recreation in and on the water".
- Issued water quality standards that defined acceptable limits of various pollutants in U.S. waterways.
- Safe Drinking Water Act- (1974, 1986, 1996) sets the national standards for safe drinking water.
- It is responsible for establishing maximum contaminant levels (MCL) for 77 different elements or substances in both surface water and groundwater.

TABLE 14.1

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Source: U.S. Environmental Protection Agency, http://www.epa.gov/safewater/contaminants/index.html.

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TABLE 14.2

The current leading causes and sources of impaired waterways in the United States

	Causes of impairment	Sources of impairment
Streams and rivers	Bacterial pathogens, habitat alteration, oxygen depletion	Agriculture, water diversions, dam construction
Lakes, ponds, and reservoirs	Mercury, PCBs, nutrients	Atmospheric deposition, agriculture
Bays and estuaries	Bacterial pathogens, oxygen depletion, mercury	Atmospheric deposition, municipal discharges including sewage

Source: Data from U.S. Environmental Protection Agency. 2004. National Water Quality Inventory: Report to Congress.

Table 14.2 *Environmental Science*© 2012 W. H. Freeman and Company

Sustainability: A "Green" wastewater treatment plant in Massachusetts
Greenhouses decompose the organic material and remove excess nitrogen.



Figure 14.21
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Working Toward Sustainability

Building "Green" Solutions to Wastewater
 Treatment

5/28 Water Pollution CH 14 p.74NB

- I. Give an example of Point source pollution.
- 2. Explain Eutrophication.
- 3. Give some examples of Heavy Metals and where they are found.