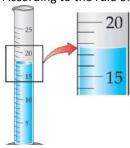
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Select the best answer for each questions

1. Which is the following is true about significant figures?

- a. They help scientists easily handle very large numbers or very small numbers
- Significant figures do not change when converting units b.
- Significant figures indicate the precision of an instrument. c.
- d. B and C
- 2. According to the rule of measured numbers what is the volume of the graduated cylinder?



- 15 ml a.
- b. 17.0 ml
- c. 15.25ml
- 17.5 ml d.
- 3. How many significant figures are in the number 0.9050?
- 3 a.
- 2 b.
- 5 c.
- d.
- How many significant figures are in the number 2201? 4.
- a.
- b. 2
- 1 c.
- d. 4
- 5. What is scientific notation?
- A way that scientists easily handle very large numbers or very small numbers a.
- b. Indicates the precision of an instrument.
- A way express the uncertainty in the measurement c.
- d. A positive and negative number
- 6. Express the following in scientific notation: .0099
- 9.90 x 10⁻⁴ a.
- 9.9 x 10⁻⁴ b.
- 9.9×10^{3} c.
- 9.9×10^{-3} d.
- 7. Express the following in scientific notation: 1800
- 1.8 x 10⁻⁴ a.
- 1.8×10^{3} b.
- 1.8 x 10⁻² c.
- d. 1.8 x 10⁻³

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- 8. Express the following in standard notation: 7.3 x 104
- 703000 a.
- h. 73000
- .00073 c.
- d. 730000
- 9. Express the following in standard notation: 2.6 x 10-4
- 26000 a.
- 26666 b.
- .00026 c.
- d. 0002.6

10. What is the SI Unit for mass?

- Second a.
- Liter h.
- Meter c.
- d. Gram

11. Which below is the correct list of metric prefixes in order of largest to smallest?

- Kilo Mili Centi Deci BasicUnit Hecto Deka Hecto a.
- Kilo Hecto Deka BasicUnit Deci Centi Mili b.
- Mili Centi Deci BasicUnit Deka Hecto Kilo c.
- King Henry Died Unexpectedly Drinking Chocolate Milk d.

12. What is the correct abbreviation for Centimeter

- a.
- cm b.
- cmm c.
- d. Km

13. Convert 7.54 Kg to milligrams.

- a. 7540000 mg
- b. 0.754 mg
- 0.00754 mg c.
- d. 0.00000754 mg

Convert 633 mL to liters. 14.

- 0.00633 L a.
- 0.000633 L b.
- 6.33 L c.
- d. 0.633 L

What is the round metal part of a triple beam balance called, where you place the object to be measured? 15.

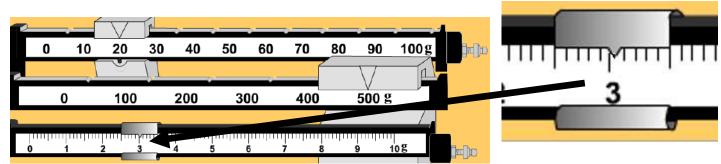
- Thumbscrew a.
- b. Zero Mark
- Plate c.
- d. Pan

16. What is the first thing you do when using a triple beam balance?

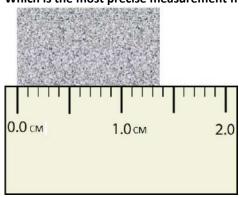
- Start with the largest slider. Move the slider until balance tips, move the slider back to the previous position, move to the next a. slider
- Continue until the final slider until the balance reads zero h.
- Move all three sliders so that they read 'zero'. c.
- Read each of the sliders and add their weights together. d.

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- Which is true about reading a graduated cylinder? 17.
- Read at the bottom of the meniscus a.
- Read at eye level b.
- c. Use clear liquids
- A and B d.
- 18. Which is the best measurement from the triple beam balance?



- 523.05 g a.
- 253.50 g b.
- c. 523.5 g
- 253 g d.
- How many significant figures are there in the best measurement from the triple beam balance above? 19.
- 3 a.
- b. 4
- 5 c.
- d. 6
- 20. Which is the most precise measurement from the meter stick in centimeters?



- .0135 cm a.
- 1.35 cm b.
- 1.350 cm c.
- 1.3 cm d.
- 21. Convert the measurement above into Km
- .000013 a.
- b. .00000135
- .0000135 c.
- d. 130000.

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Represent the measurement above in Scientific notation 22.

- 1.35 x 10⁻⁵ a.
- 1.3×10^{5} b.
- c. 1.35×10^{5}
- 1.3 x 10⁻⁵ d.

Use the two rulers measuring the pencil to answer the questions below



What is the most accurate measurement of the pencil on Ruler B 23.

- a.
- 2.55 cm b.
- 2.5 cm c.
- d. 2 cm

24. Which of the below is/are true about the measurement of the pencil on ruler A?

- a. The measurement of the pencil on ruler A is 2.55 cm
- The measurement of the pencil on ruler A is 000255 km b.
- The Scientific notation of the measurement of the pencil on ruler A is 2.55 X 10⁵ c.
- d. A and C

25. What is the SI Unit for Temperature?

- Fahrenheit (Hahaha!! Ya right, silly Liberia) a.
- Celsius b.
- Kelvin c.

c. d.

d. All of the above

Convert the following

50 °C = _____ °F 26. 122 a. 120 b. 10 c. d. 323 27. 149 °F = _____ °C 300 a. b. 65 100 c. d. 365 28. 273 °C = _____ °K a. b.