

Name _____
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Environmental Chemistry Unit Vocabulary – Nuclear Energy

1. Radioactivity: _____

2. Radioactive Nucleus: _____

3. Isotope: _____

4. Atomic Number: _____

5. Mass Number: _____

6. Nuclear Radiation: _____

7. Alpha Particle: _____

8. Beta Particle: _____

9. Gamma Ray: _____

10. Neutron Emission: _____

11. Nuclear Decay: _____

12. Transmutation: _____

13. Half Life: _____

14. Radioactive Dating: _____

15. Nuclear Fission: _____

16. Fission Chain Reaction: _____

17. Critical Mass: _____

18. Nuclear Fusion: _____

19. Background Radiation: _____

20. Radioactive Tracers: _____

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Environmental Chemistry Unit Vocabulary – Alternative Energy

1. Fossil Fuels: _____

2. Renewable Resource: _____

3. Nonrenewable Resource: _____

4. Solar Power: _____

5. Wind Power: _____

6. Hydroelectric Power: _____

7. Geothermal Energy: _____

8. Nuclear Energy: _____

9. Biomass: _____

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Environmental Chemistry Unit Vocabulary – Periodic Table

1. Pure Substance: _____

2. Element: _____

3. Compound: _____

4. Atom: _____

5. Molecule: _____

6. Nucleus: _____

7. Proton: _____

8. Neutron: _____

9. Electron: _____

10. Valence Electrons: _____

11. Periodic: _____

12. Chemical Symbol: _____

13. Isotope: _____

14. Average Atomic Mass: _____

15. Atomic Number: _____

16. Atomic Mass: _____

17. Group (Family): _____

18. Period: _____

19. Bohr Diagram: _____

20. Electron Cloud Model: _____

21. Lewis Dot Diagram: _____

22. Anion: _____

23. Cation: _____

24. Metal: _____

25. Nonmetal: _____

26. Metalloid: _____

27. Alkali Metals: _____

28. Alkaline Earth Metals: _____

29. Transition Metals: _____

30. Halogen: _____

31. Noble Gas: _____

32. Hydrogen: _____

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Environmental Chemistry Unit Vocabulary – Chemical Bonding & Polymers

1. Chemical Reaction: _____

2. Chemical Bond: _____

3. Covalent Bond: _____

4. Ionic Bond: _____

5. Chemical Symbol: _____

6. Chemical Formula: _____

7. Subscript: _____

8. Coefficient: _____

9. Reactants: _____

10. Products: _____

11. Yields: _____

12. Law of Conservation of Mass: _____

13. Synthesis Reaction: _____

14. Decomposition Reaction: _____

15. Single Displacement: _____

16. Double Displacement: _____

17. Combustion: _____

18. Activation Energy: _____

19. Exothermic Reaction: _____

20. Endothermic Reaction: _____

21. Precipitate: _____

22. Catalyst: _____

23. Inhibitor: _____

24. Organic Compound: _____

25. Hydrocarbon: _____

26. Alkane: _____

27. Alkene: _____

28. Alcohol: _____

29. Polymer: _____

30. Natural Polymers: _____

31. Synthetic Polymers: _____

32. Biochemical Compounds: _____

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Environmental Chemistry Unit Vocabulary – Cycles & Pollution

1. Universal Solvent: _____

2. Polar Molecule: _____

3. Like dissolves like: _____

4. Acid: _____

5. Base: _____

6. Neutral: _____

7. pH: _____

8. Neutralization Reaction: _____

9. Evaporation: _____

10. Transpiration: _____

11. Condensation: _____

12. Precipitation: _____

13. Ground Water: _____

14. Surface Water: _____

15. Pollution: _____

16. Potable: _____

17. Non-potable: _____

18. Artificial Eutrophication: _____

19. Biomagnification: _____

20. Troposphere: _____

21. Stratosphere: _____

22. Ozone: _____

23. Photochemical Smog: _____

24. Free Nitrogen: _____

25. Nitrogen Fixation: _____

26. Denitrification: _____

27. Greenhouse Effect: _____

28. Global Warming: _____

29. 3 R's: _____

30. Biodegradable: _____

31. Brownfield Site: _____
