Grade 10/11/12

Distance Learning Module 1: Week of: March 30-April 3

Modern Atomic Theory, Atomic Structure Review, Periodic Table

Content Area: Honors Chemistry - Modified from <u>Unit 6 - Atomic Structure</u>, <u>Electron Configuration</u> & Periodic Relationships

Targeted Goals from Stage 1:

Content Knowledge: Atoms are comprised of subatomic particles held together by fundamental forces and their quantity and arrangement determines the atom's properties, identity, and behavior. Each atom has a charged substructure consisting of a nucleus, which is made of protons and neutrons, surrounded by electrons. The periodic table orders elements horizontally by the number of protons in the atom's nucleus and places those with similar chemical properties in columns. The repeating patterns of this table reflect patterns of outer electron states.

Vocabulary: wave, wavelength, frequency, photon, quantum, atomic emission spectrum, electron configuration, principal energy level, sublevel, orbital, periods, groups, valence electrons, ionization energy, atomic radius, and electronegativity

Skills:

Draw a model of a given atom.

Expectation:

Description of Task (s): Monday: Students can set their own pacing, but make sure to meet the weekly expectations shown below: • Watch EdPuzzle Video on History of Atomic Structure & take notes.	Resources and Materials: Unit 6 (2.1-2.4) PowerPoint Slide Show EdPuzzle-History of the Atom Notes Model of Atom Atomic Theory	Daily Checks (Return to Google Classroom or snapshots from a cell phone) Submit ONE or MORE of the following to Google Classroom each day: • pictures of your notes from EdPuzzle Videos or Unit 6 PowerPoint • pictures of completed
 Read through Unit 6 PowerPoint Slide Show & take notes 	Atom_Atomic Theory	Atomic Theory handout
Tuesday: • Watch Edpuzzle Video on Dalton's Atomic Theory & answer embedded multiple choice. • Watch Edpuzzle Video on JJ Thomson & answer embedded multiple choice.	 EdPuzzle- Daltons Atomic Theory EdPuzzle- JJ Thomson Cathode Ray Experiment, and the Plum Pudding Model 	Submit ONE or MORE of the following to Google Classroom: • pictures of your notes from Edpuzzle Videos or Unit 6 Powerpoint • answer embedded multiple choice while watching edpuzzle videos
Watch Edpuzzle Video	EdPuzzle - Rutherford, The	

Description of Task (s):	Resources and Materials:	Daily Checks (Return to Google Classroom or snapshots from a cell phone)
on Ruther & the Gold Foil Experiment & answer embedded multiple choice. • Read through Unit 6 PowerPoint Slide Show & take notes	Gold Foil Experiment, and The Nuclear Atom	
 Wednesday: Watch Edpuzzle Video on Structure of the Atom & answer embedded multiple choice. Watch Edpuzzle Video on Nuclide Symbols Read through Unit 6 PowerPoint Slide Show & take notes 	 EdPuzzle – Structure of the Atom EdPuzzle - Nuclide Symbols: Atomic Number, Mass Number, Ions, and Isotopes A6_Atomic Structure Key_A6_Atomic Structure_Isotopes_Atomic Mass.pdf 	Submit ONE or MORE of the following to Google Classroom: • pictures of your notes from EdPuzzle Videos or Unit 6 PowerPoint • answer embedded multiple choice while watching EdPuzzle videos • practice worksheet on Atomic Structure
Thursday: • Watch Edpuzzle Video on A tour of the Periodic Table • Read through Unit 6 PowerPoint Slide Show & take notes	EdPuzzle – Tour of the Periodic Table	Submit ONE or MORE of the following to Google Classroom: • pictures of your notes from Edpuzzle Video or Unit 6 Powerpoint
Friday: • Complete Distance Learning Practice Test	Distance Learning_Practice Test_Atomic Structure	Submit ONE or MORE of the following to Google Classroom: • Completed Distance Learning Practice Test

Week criteria for success (attach student checklists or rubrics): By the end of this week, students should have:

- watched Edpuzzle videos and responded to embedded video questions where appropriate
- taken notes on EdPuzzle videos or Unit 6 Chapter 2 PowerPoint Slide Show (Atomic Structure)
- completed practice worksheet on Atomic Structure
- completed Distance Learning Practice Test

Supportive resources and tutorials for the week (plans for re-teaching):

- online virtual Q and A help sessions (see Google Classroom for times and invite codes)
- read and re-read the textbook
- practice worksheets and corresponding answer keys in Google Classroom