Bio-Chemistry Unit Study Guide (Chapter 2, ISN 33-53)

1) Describe the 3 subatomic parts of an atom and location where found. 2) Compare the following pairs of terms: a. Atomic number/ atomic mass b. Covalent bond/ionic bond c. Cohesion/adhesion d. Solvent / solute e. Organic compound/inorganic compound f. Monosaccharide/polysaccharide 3) Why are elements considered neutral? 4) What characteristics are found in carbon that makes it excellent for the building of organic compounds? 5) Name the 4 macromolecules and the monomers that form them. 6) How many amino acids are found in the human body? 7) What is a catalyst? 8) How can pH and temperature affect enzyme function? 9) Why are enzymes referred to as a lock and key model? What represents the lock? Key? 10) In chemical reactions, how are atoms arranged? 11) What are reactants/products in a chemical reaction? How can you identify them in an equation? 12) Describe the polarity of a water molecule. 13) What are some of the biological properties of water?

14) Explain what happens in a Dehydration Synthesis reaction and a Hydrolysis reaction. What is the

15) Review the pH scale and properties of an acid, base, and buffer.

major difference?

16) Why is w	ater neutral in pH despit	e the fact that it produces	hydrogen and hydroxide ions?
17) What ele	ments compose each of	the following types of orga	anic compounds?
a.	Carbohydrates:		
b.	Lipids:		
C.	Proteins:		
d.	Nucleic Acids:		
18) What rol	e do enzymes play inside	e a cell?	
19) What is t	he function of the enzym	ne active site?	
20) Explain w	hy enzymes are specific	for only one type of reacti	on.
21) Why can	proteins be classified as	polymers, but lipids canno	t?
To which gr	oup of organic compo	unds does each of the fo	llowing belong?
A. Carb	ohydrates B. Lipids	s C. Nucleic Acids	D. Proteins
Enzymes	Glucose	Fats and oils	Cholesterol
Sugars	Starch	Hormones	Cellulose
Waxes	Monosaccharide	DNA and RNA	
Polysacchario	des Glyco	ogen	